

PROGRAM REVIEW

# Allan Hancock College

# **CHEMISTRY Program Review**

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# CHEMISTRY PROGRAM REVIEW

## **Status Summary - Plan of Action-Post Validation**

During the academic year, <u>2021</u>, <u>2022</u> completed program review. The self- study and validation teams developed a final plan of action-post validation based on information in the self study and the recommendations of the validation team. For each plan, indicate the action taken, the result of that action, and the current status of the plan, if it is incomplete.

(If any plan was made and action not taken, please state the rationale for not pursuing that particular item.)

PLAN OF ACTION	<u>ACTION TAKEN ,RESULT, AND</u> STATUS
Support the STEM center by informing students through classroom presentations and other communications.	On going
Increase the awareness and participation of students in science related programs through Bull-Dog Bound and Friday Night Science events.	On going
Increase the number of class sections as demand dictates.	On going, as allowed.
Work with neighboring colleges and universities to ensure that classes articulate (C-ID) and topics are aligned.	On going
Increase the general department budget to accommodate the increase in student population.	Funding strategies will be pursued.
Hire additional chemistry faculty to keep up with growing class sections.	On going, as allowed.
Create published course materials for our campus to utilize to help bring down costs to our students.	On going
Work with other disciplines such as Mathematics that align with our PLOs to help make sure students are prepared.	On going

# Chemistry Program Review Self Study

#### I. Program Mission:

The chemistry program is one of many disciplines grouped together in the Life and Physical Science department. Many courses require chemistry as a prerequisite to help prepare them for the rigor of other core courses in Biology, Physics, and especially the medical field. These courses enable students to complete lower division requirements for transfer to institutions of higher learning. For science majors, the UC and CSU systems now requires General and Organic Chemistry before transfer.

#### The chemistry program consists of the following courses:

- CHEM110: Chemistry and Society (distance learning)
- CHEM120: Introduction to Chemistry (C-ID: CHEM101)
- CHEM140: Introduction to Organic and Biological Chemistry (C-ID: CHEM102)
- CHEM150: General Chemistry 1 (C-ID: CHEM110)
- CHEM151: General Chemistry 2 (C-ID: CHEM120S)
- CHEM180: Organic Chemistry 1 (C-ID: CHEM150)
- CHEM181: Organic Chemistry 2 (C-ID: CHEM160S)
  - II. Progress Made Towards Past Program/Department Goals

Headcount for our sections was up to 78% before dropping to 66% due to the COVID-19 pandemic.

Fill rates are greater than 80%, except for CHEM181, which is a continuation course as well as the last course required before transfer.

Full time instruction has increased over part-time instruction (52% vs. 48%) which flipped from last program review.

FTES/FT EF has steadily been increasing from 17.122 to 17.756

Chemistry courses at our satellite campus (LVC) have grown in demand to include the General Chemistry series (CHEM150 and 151). Equipment was purchased for both campuses to ensure equity in instruction and experience in lab.

The Organic Chemistry series (CHEM180 and 181) continues to thrive and fill each Fall semester it is offered. UC and CSU systems recently began requiring these courses for proper transfer.

The chemistry team participates in the Friday Night Science event created by Rob Jorstad. This yearly event continues to help attract attention to our campus and fosters deeper learning in our students before they transfer or graduate.

The chemistry team provides tours of our facilities and works with other disciplines cooperatively when hosting outreach events for elementary and high-school students (Bull-Dog Bound, slime room and other hand-on activities are provided).

The chemistry team provides our time and support to AHC's STEM and MESA programs as well as having a table at BOW-WOW events on both SM and LVC campuses to help promote excitement in the sciences.

#### III. Analysis of Resource Use and Program Implementation

Current technology and fiscal resources are being utilized to ensure Emergency Remote Learning during the COVID-19 pandemic was as successful as possible. The pandemic stopped all hands-on learning in most chemistry courses and ZOOM/CANVAS became the main modes of teaching. Special cameras and televisions were outfitted in the rooms to capture the instructor and material as best as possible.

As the lab sections grow, the M-building is fully utilized as we accommodate all the different chemistry labs. With only three chemistry labs (M-204/M-213/LVC3-102), the need for more lab space will eventually arise. The increase in lab sections along with retirements has left the chemistry department severely lacking in personnel. Chemistry sections have increased from 43 in 2017-2018 to 47 in 2020-2021. Because of limited rooms for lab, this expansion cannot continue without additional facilities that include gas, water, and fume hoods. Potentially, lab facilities could be added on the Lompoc Valley Campus. During the 2022 year, administrations hopes to hire two full-time chemists to help out the current full-time chemists that have been overloaded for many years now.

### IV. Program SLOs/Assessment

Based upon the course statistics and overall data collected from Spring 2016 to Spring 2019:

73% of the students have demonstrated mastery of CHEM PSLO - ... the approach and rationale of the scientific method and ability to apply these principles to solve problems.

80% of the students have demonstrated mastery of CHEM PSLO - ... stoichiometric calculations.

78% of the students have demonstrated mastery of CHEM PSLO - ... laboratory technique.

Our program has been lacking enough full-time faculty to satisfy the growing student population needing access to our courses. While we do have one full-time faculty member consistently teaching one of each of our science major courses (CHEM150, 151, and 180/181), any extra sections of CHEM150, and the majority of CHEM120, is taught by part-time instructors. In addition, we have seen an upward trend in the number of chemistry classes taught per semester (increasing FTES), with more classes being taught by a growing part-time instructor pool. This inconsistency in the instructor pool (sometimes with quick turnover) for some of our most in-demand classes creates data that can be inconsistent. Extra classes not taught by part-time instructors are added onto the load of our understaffed group of full-time faculty members.

Outcomes data for Chemistry from only Spring 2016 to Spring 2019 is available. The previous software for outcomes data was phased out in 2019, so no data was recorded since that time. In addition, data input leading up to the eLumen phase-out lessened, as the department was assured that old data could be entered into the new system once it is up and running. While these are possible reasons for a noticeably lower mastery of PSLO compared to previous years, steps have been discussed and some have already been taken to bolster student success rates in the future. These include updating and/or diversifying instruction methods and materials, and the implementation of outside resources to help improve student success in the classroom and laboratory.

As stated earlier, CHEM120 is primarily taught by part-time instructors, many of which may not teach consecutive semesters for us. As such, SLO data input by part-time instructors is inconsistent. Having a dedicated faculty member teaching CHEM120 consistently may help to achieve a consistent data set for success rate in CHEM120

CHEM150 had been primarily taught by full-time faculty members but has seen an increase in parttime instruction as demand for the class has surged beyond what our full-time faculty can handle. CHEM150 saw the most regular SLO data input out of our courses from Spring 2016 to Fall 2018 (excepting Spring 2017). Most SLOs show steady improvement in standards met after a dip in % from Fall 2016 to Fall 2017. This trend is not entirely clear, though, as there is a lack of SLO data for Spring 2017.

CHEM151 is only taught by full-time faculty. SLO data input was sporadic up until Spring 2018, when a different full-time faculty member became the primary instructor for the course. We feel that data input will continue to be more consistent moving forward. With fewer class offerings than CHEM150, sample sizes will be smaller, though we feel that with regular SLO data input, trends will become more apparent with time.

CHEM180/181 is a relatively new offering. These courses are taught only by full-time faculty. Because these courses are usually required for only a few majors, class sizes tend to be smaller, and the courses are not offered as frequently as our other courses (CHEM120, 150, 151). As such, clear SLO data trends are not available. More regular SLO data input will be practiced moving forward.

It is clear to us that improvements can be made in certain areas of each course to help advance and ensure student success. Even with limited SLO data, the full-time faculty members teaching each course have been working together with clear goals in mind to adapting the courses to better serve the students and encourage success. We feel that the Learning Outcomes Assessment Committee's shift away from SLOs and toward PLOs will help to show how our courses work together to show student success at the program level.

As a program, we will encourage our full-time and part-time instructors to regularly input PLO data into the new SPOL program that has recently launched, so that more telling data can be collected in the future.

#### V. Distance Learning

**CHEMISTRY 110, Catalog Description:** An introduction to the fundamentals of chemistry including the composition of matter, energy, and chemical reactions and their application to everyday living. Applications of chemistry in the areas of medicine, nuclear power, plastics, household products, and society's effect on the environment will be emphasized. Intended for non-science majors. Not open to students who are enrolled in or have completed Chemistry 100, 105, or Chemistry 120.

Many online students might not have the organizational skills and the diligence to succeed in an online course. With low faculty numbers and the pandemic demanding other online resources, finding a consistent instructor has been challenging for this course. As a result, the success rate for this course has fallen from the 77% success rate reported last program review. Currently, we have seen the success rate bounce from 63%, down 45%, and back up to 58%. In Fall 2019, a new instructor started teaching the course. Because the former instructor had retired, the course essentially started from scratch. Within 2 semesters, the at-home lab kit that had been used were discontinued. This required the instructor to write in-house labs for the course. This began improving the course success rate, but during the Spring 2021, Fall 2021, and Spring 2022 semesters, there were three separate instructors who taught the course. This complicated the ability to observe true course success rates with fluctuations in teaching styles, expectations, and learning materials. With more faculty hires, we hope to allow an instructor to focus their attention on this course to help improve student success and retention. Without an instructor present, it is no wonder that students would have a greater difficulty succeeding with lab experiments.

#### VI. Success, Retention, and Equity

We do work very closely with MESA, STEM, Counseling, LAP, and others to ensure student equity and success. Our program is quite highly regarded by our students as shown in the student survey. Our current retention rates in the chemistry program are 85.3 % average over the past six years (2015-2021), with a 71.2% average success rate. The quality of teaching encourages the student diversity and facilitates their success in our programs as shown by the following demographics.

#### **Program Equity: AGE**

	Equity:AGE	Retention %	Success %		
	Chemistry program	86.2 % average (2015-2021)	71.0 % average (2015-2021)		
	Hancock College	89.1 % average (2015-2021)	77.0 % average (2015-2021)		
Table 1.1. Average retention and success percentages for gender compared					

Table 1.1: Average retention and success percentages for gender compared.

The retention and success rate averages over the past six years have been over 70%, with a success percentage of 71.0 compared to the college as a whole with 77.0% as shown in Table 1.1.

Equity:AGE	PPG Retention %	PPG Success %
Chemistry program	-4.1 %	-4.5 %
Hancock College	-1.7 %	-3.9 %

Table 1.2: Impact values for age groups compared.

The success rate of the chemistry program at 65.4 % is below that of the college as a whole as shown in Table 1.2. This factor was due to incoming students <20 years of age having a Percentage Point Gap (PPG) of -4.1% in retention and -4.5% for success. Hancock college as a whole is also struggling in the age category with a PPG of -3.9% for success. From this data, it appears our younger students are coming in less prepared. We need to keep an eye on this trend as more basic skills courses (English and Math) are removed due to recently passed legislature at the state level. Our long term goals will help attempt to address this issue.

## **Program Equity: ETHNICITY**

Γ	Equity:ETHNICITY	Retention %	Success %		
	Chemistry program	78.5 % average (2015-2021)	63.5 % average (2015-2021)		
	Hancock College	87.9 % average (2015-2021)	74.0 % average (2015-2021)		
he 1.2. Average estention and eveness nervoustages for ethnicity compared					

 Table 1.3: Average retention and success percentages for ethnicity compared.

The retention rate averages over the past six years have been over 70%, with a success percentage of 78.5% compared to the college as a whole with 87.9% as shown in Table 1.3. The success percentage of certain ethnicities is at 63.5% and is broken down further below.

Equity:ETHNICITY	PPG Retention %	PPG Success %
Chemistry program	Hispanic: -3.6 %	Hispanic: -6.2%/Nativ Am: -4.8%
Hancock College	Black: -1.1%	Black: -5.8%
_	Hispanic: -1.6%	Hispanic: -4.6%
	Nativ Am: -2.7%	Nativ Am: -5.9%
	Pac Isl: -1.6%	Pac Isl: -3.7%

Table 1.4: Impact values for ethnicity compared.

For the chemistry program, Hispanic and Native Americans were negatively impacted as shown in Table 1.4. Hispanic had a retention PPG of -3.6% with a -6.2% success this past 2020-2021 year. The Native American success rate was -4.8% PPG. The college as a whole also has impacted ethnicities as well. Black, Hispanic, Native Americans, and Pacific Islanders were ranging with a PPG of -3.7 to -5.9% for success rates. Our long term goals will help attempt to address this issue.

### **Program Equity: GENDER**

	Equity:GENDER	Retention %	Success %				
	Chemistry program	82.4 % average (2015-2021)	66.5 % average (2015-2021)				
	Hancock College	88.1 % average (2015-2021)	71.0 % average (2015-2021)				
Tabl	able 1 Ex Average retention and success persontages for gonder compared						

Table 1.5: Average retention and success percentages for gender compared.

The retention rate averages over the past six years have been over 80%, with a success percentage of 82.4% compared to the college as a whole with 88.1% as shown in Table 1.5. The success percentage of the female gender is 64.4% versus 66.8% for males.

# **Program Equity: 1<sup>st</sup> Time Student**

Equity: 1 <sup>st</sup> Time Student	Retention %	Success %
Chemistry program	86.3 % average (2015-2021)	72.4 % average (2015-2021)
Hancock College	91.6 % average (2015-2021)	78.9 % average (2015-2021)

Table 1.6: Average retention and success percentages for 1<sup>st</sup> time student compared.

The retention and success rate averages over the past six years have been over 70%, with a success percentage of 72.4% compared to the college as a whole with 78.9% as shown in Table 1.6.

Equity:1 <sup>st</sup> Time	PPG Retention %	PPG Success %
Chemistry program	0.5 %	-10.0 %
Hancock College	-2.2 %	-13.5 %

 Table 1.7: Impact values for 1<sup>st</sup> time students compared.

For 1<sup>st</sup> time students, the chemistry program had a -10.0% PPG while Hancock college was at -13.5 % PPG as shown by Table 1.7. Our long term goals will help attempt to address this issue.

## VII. Trend Analyses/Outlook

The COVID-19 pandemic forced public places to close and would not allow large gatherings of groups. Classes and labs could no longer be held on campus and everyone was forced to teach remotely. Canvas and ZOOM technology was utilized to bring the information to the students as best as possible. Laboratory tactile and technical skills were abandoned for simple exposure by viewing the lab being performed for them. Lab writing skills and knowledge were continued by trying to complete the experiments as best as possible. This will definitely create a small pocket of students that didn't get the chance to handle glassware or have direct exposure in a laboratory setting. Patience will be a virtue going forward to ensure student safety at the higher levels of chemistry.

VIII. Long-Term Program Goals and Action Plans (Aligned With the College Educational Master Plan)

Bring all chemistry courses up-to-date with the Course Identification Numbering System (C-ID). Currently, CHEM140, our introductory organic chemistry course, is not mapped to the C-ID CHEM102. The biochemistry portion of the lecture and lab needs to be updated as well as the organic labs to ensure the students are getting the best exposure possible before moving on in the medical field.

The chemistry program has been mapped to the Guided Pathways to help students navigate the educational system and obtain the degrees desired: https://www.hancockcollege.edu/pathways/sciences-technologies/chemistry.php

It is of utmost concern and interest to the chemistry team to increase retention and success for all AHC students taking the general chemistry series, organic classes, or simply any non-majors chemistry classes in which they are enrolled. It has been assessed that one of the major hurdles to academic success in

chemistry classes is mathematical competency. Because most chemistry classes require (at the least) a solid understanding of algebra, it is imperative that students enter these classes with algebraic fluency. However, due to recent legislation, Algebra I and II cannot be offered at the community college level. This has caused the chemistry team to consider requiring a supplemental "Math in Chemistry" course to be offered concurrently with the following courses: Chemistry 120, 150, and 151. These courses would allow the hours spent in chemistry lecture and lab to be exclusively focused on chemical theory and application instead of review of mathematical concepts.

A loose breakdown of these supplemental courses would be as follows:

- 1. Online asynchronous instruction
- 2. One hour supplemental instruction/week through video lectures
- 3. One assignment/week focusing on mathematical principles that coincide with concurrent chemistry problems

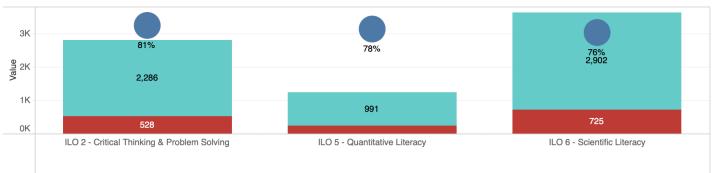
# Chemistry Program Review Assessment Plan

## ILO Data





**ILO Performance Chart: Chemistry-** This is the ILO performance of the program for the past 6 academic years in a table that includes the number of courses that are connected to each ILO.

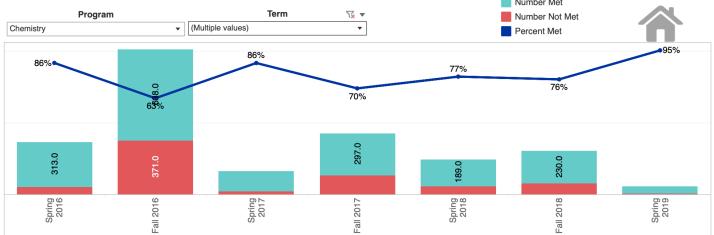


ILO Performance Table: Chemistry- This is the ILO performance of the program for the past 6 academic years.

	# of Connected Courses	Avg. Percent Met	Number Met	Number Not Met
ILO 2 - Critical Thinking & Problem Solving: Explore issues through various information sources; evaluate the credibility and significance of both the information and the source to arrive at a reasoned conclusion.	4	81%	2,286	528
ILO 5 - Quantitative Literacy: Use mathematical concepts and models to analyze and solve real life issues or problems.	4	78%	991	239
ILO 6 - Scientific Literacy: Use scientific knowledge and methodologies to assess potential solutions to real-life challenges.	5	76%	2,902	725

# PLO Data

PLO Performance Chart: Chemistry This is a chart showing the PLO percent and the count of students that met standards by term. Number Met



PLO Performance Table: Chemistry- This is a table showing the overal PLO performance over the last 6 academic years, including percent and numbers of students meeting standards.

			Number Met	Number Not Met	Percent Met
	CHEM1	CHEM PSLO - The student will demonstrate mastery of the approach and rationale of the scientific method and be able to apply these principles to solve problems.	339.0	125.0	73%
Chemistry	CHEM2	CHEM PSLO - The student will demonstrate mastery of stoichiometric calculations.	825.0	203.0	80%
Chemistry	CHEM3	CHEM PSLO - The student will demonstrate mastery of laboratory technique.	157.0	45.0	78%
	CHEM4	CHEM PSLO - Course doesn't map to a degree or certificate.	540.0	330.0	62%

# SLO/CLO Data Spring 2016 – Spring 2019



Historical CLO Performance Table: Chemistry- This is a chart of the table above.

6. Historical Course Performance: Chemistry- This is SLO assessment by course, including percent and number of students that met standards.

CHEM150	266.0		932	2.0	<b>78%</b>
CHEM120	318.	0	507.0	61%	
CHEM151	96.0	331.0			78%
CHEM180	<mark>950</mark> 0				<b>83%</b>
CHEM110	<mark>32</mark> .0				73%
CHEM181	2300				87% 🗖

10010.					
			Number Met	Number Not Met	Percent Met
CHEM110	CHEM110.1	CHEM110 SLO1 - Describe the structure and composition of matter and its relationship to the macroscopic properties of substanc	22.0	8.0	73%
CHE	CHEM110.5	CHEM110 SLO5 - Demonstrate proficiency with basic chemistry apparatus and analyzing data from experiments.	11.0	4.0	73%
	CHEM120.1	CHEM120 SLO1 - Use scientific notation, significant figures, and dimensional analysis.	220.0	82.0	73%
-	CHEM120.2	CHEM120 SLO2 - Write formulas and names of simple compounds and ions.	71.0	59.0	55%
CHEM120	CHEM120.3	CHEM120 SLO3 - Solve problems related to chemical equations and density.	71.0	59.0	55%
	CHEM120.4	CHEM120 SLO4 - Define and give examples of chemical terms.	74.0	59.0	56%
	CHEM120.5	CHEM120 SLO5 - Able to use the appropriate laboratory apparatus to perform accurate & precise measurements.	71.0	59.0	55%
	CHEM150.1	CHEM150 SLO1 - Perform stoichiometric calculations.	213.0	65.0	77%
	CHEM150.2	CHEM150 SLO2 - Balance chemical equations, including oxidation-reduction.	197.0	42.0	82%
CHEM150	CHEM150.3	CHEM150 SLO3 - Solve questions involving gas laws	142.0	31.0	82%
CHE	CHEM150.4	CHEM150 SLO4 - Provide the quantum numbers for any specific electron.	184.0	59.0	76%
	CHEM150.5	CHEM150 SLO5 - Perform calculations involving thermodynamics.	81.0	44.0	65%
	CHEM150.6	CHEM150 SLO6 - Perform laboratory quantitative analysis.	115.0	25.0	82%
	CHEM151.1	CHEM151 SLO1 - Perform kinetic calculations.	95.0	9.0	91%
	CHEM151.2	CHEM151 SLO2 - Perform equilibrium calculations.	108.0	26.0	81%
M151	CHEM151.3	CHEM151 SLO3 - Perform thermodynamic calculations.	70.0	30.0	70%
CHEM151	CHEM151.4	CHEM151 SLO4 - Define and explain concepts of equilibria.	11.0	8.0	58%
	CHEM151.5	CHEM151 SLO5 - Interpret a pH graph from an acid-base titration.	16.0	9.0	64%
	CHEM151.6	CHEM151 SLO6 - Perform qualitative analysis.	31.0	14.0	69%

					15	
	CHEM180.1	CHEM180 SLO1 - Make predictions on physical properties and chemical reactivity based on molecular structure.	15.0	3.0	83%	
CHEM180	CHEM180.2	CHEM180 SLO2 - Define structures of alcohols, alkyl halides, and hydrocarbons and be able to draw the condensed and line-bond formul	15.0	3.0	83%	
0	CHEM180.3	CHEM180 SLO3 - Determine reaction mechanisms and propose synthesis routes for organic reactions to be carried out in the	15.0	3.0	83%	
CHEM1	CHEM181.1	CHEM181 SLO1 - Make predictions on physical properties and chemical reactivity based on molecular structure.	13.0	2.0	87%	
0		structure.				

# SLO/CLO Data Spring 2016



6. Historical Course Performance: Chemistry- This is SLO assessment by course, including percent and number of students that met standards.

CHEM150	30.0		126.0	81%
CHEM120	15.0	100.0		87%
CHEM151	<mark>4.</mark> 0	87.0		96%

Historical CLO Performance Table: Chemistry- This is a chart of the table above.

			Number Met	Number Not Met	Percent Met
CHEM1	CHEM120.1	CHEM120 SLO1 - Use scientific notation, significant figures, and dimensional analysis.	100.00	15.00	87%
	CHEM150.1	CHEM150 SLO1 - Perform stoichiometric calculations.	21.00	5.00	81%
	CHEM150.2	CHEM150 SLO2 - Balance chemical equations, including oxidation-reduction.	21.00	5.00	81%
1150	CHEM150.3	CHEM150 SLO3 - Solve questions involving gas laws	21.00	5.00	81%
CHEM150	CHEM150.4	CHEM150 SLO4 - Provide the quantum numbers for any specific electron.	21.00	5.00	81%
	CHEM150.5	CHEM150 SLO5 - Perform calculations involving thermodynamics.	21.00	5.00	81%
	CHEM150.6	CHEM150 SLO6 - Perform laboratory quantitative analysis.	21.00	5.00	81%
CHEM1.	CHEM151.2	CHEM151 SLO2 - Perform equilibrium calculations.	87.00	4.00	96%

# SLO/CLO Data

Fall 2016



6. Historical Course Performance: Chemistry- This is SLO assessment by course, including percent and number of students that met above. standards.

CHEM120		295.0	55	<b>F5</b> .0
CHEM150	49.0	185.0		79%
CHEM180	<b>946</b> .0			83% 🗖
CHEM110	<mark>83.0</mark>			73%
CHEM151	8000			77%

Historical CLO Performance Table: Chemistry- This is a chart of the table

			Number Met	Number Not Met	Percent Met
CHEM110	CHEM110.1	CHEM110 SLO1 - Describe the structure and composition of matter and its relationship to the macroscopic properties of substanc	22.00	8.00	73%
CHEI	CHEM110.5	CHEM110 SLO5 - Demonstrate proficiency with basic chemistry apparatus and analyzing data from experiments.	11.00	4.00	73%



					10
	CHEM120.1	CHEM120 SLO1 - Use scientific notation, significant figures, and dimensional analysis.	71.00	59.00	55%
	CHEM120.2	CHEM120 SLO2 - Write formulas and names of simple compounds and ions.	71.00	59.00	55%
CHEM120	CHEM120.3	CHEM120 SLO3 - Solve problems related to chemical equations and density.	71.00	59.00	55%
-	CHEM120.4	CHEM120 SLO4 - Define and give examples of chemical terms.	71.00	59.00	55%
	CHEM120.5	CHEM120 SLO5 - Able to use the appropriate laboratory apparatus to perform accurate & precise measurements.	71.00	59.00	55%
	CHEM150.1	CHEM150 SLO1 - Perform stoichiometric calculations.	98.00	28.00	78%
	CHEM150.2	CHEM150 SLO2 - Balance chemical equations, including oxidation-reduction.	16.00	1.00	94%
CHEM150	CHEM150.3	CHEM150 SLO3 - Solve questions involving gas laws	33.00	7.00	83%
CHEI	CHEM150.4	CHEM150 SLO4 - Provide the quantum numbers for any specific electron.	12.00	5.00	71%
	CHEM150.5	CHEM150 SLO5 - Perform calculations involving thermodynamics.	9.00	8.00	53%
	CHEM150.6	CHEM150 SLO6 - Perform laboratory quantitative analysis.	17.00	0.00	100%
CHEM1	CHEM151.6	CHEM151 SLO6 - Perform qualitative analysis.	20.00	6.00	77%
-	CHEM180.1	CHEM180 SLO1 - Make predictions on physical properties and chemical reactivity based on molecular structure.	15.00	3.00	83%
CHEM180	CHEM180.2	CHEM180 SLO2 - Define structures of alcohols, alkyl halides, and hydrocarbons and be able to draw the condensed and line-bond formul	15.00	3.00	83%
	CHEM180.3	CHEM180 SLO3 - Determine reaction mechanisms and propose synthesis routes for organic reactions to be carried out in the	15.00	3.00	83%

# SLO/CLO Data Spring 2017

Program Term \<mark>x</mark> ▼ Number Met Chemistry • Spring 2017 • Number Not Met Percent Met

6. Historical Course Performance: Chemistry- This is SLO assessment by course, including percent and number of students that met standards.

CHEM151	2.00	89.00	98%
CHEM120	8.00 28.00		<b>78%</b>
CHEM181	2.003.00		87%
CHEM150	10.00 10.00	50%	

Historical	CLO	Performance	Table:	Chemist	ry-	This is a	chart of	f the
table above.								
					er.	er	let	t

			Numbe Met	Numbe Not Me	Percer Met
CHEM1 CHEM1	CHEM120.1	CHEM120 SLO1 - Use scientific notation, significant figures, and dimensional analysis.	28.00	8.00	78%
CHEM1	CHEM150.1	CHEM150 SLO1 - Perform stoichiometric calculations.	10.00	10.00	50%
M151	CHEM151.1	CHEM151 SLO1 - Perform kinetic calculations.	84.00	1.00	99%
CHEM151	CHEM151.5	CHEM151 SLO5 - Interpret a pH graph from an acid-base titration.	5.00	1.00	83%
CHEM1	CHEM181.1	CHEM181 SLO1 - Make predictions on physical properties and chemical reactivity based on molecular structure.	13.00	2.00	87%

# SLO/CLO Data Fall 2017



6. Historical Course Performance: Chemistry- This is SLO assessment by course, including percent and number of students that met tandards.

CHEM150	115.0	287.0	71% 📃
CHEM151	<b>11100</b> 0	42%	

Historical CLO Performance Table: Chemistry- This is a chart of t	he
table above.	

			Number Met	Number Not Met	Percent Met
	CHEM150.1	CHEM150 SLO1 - Perform stoichiometric calculations.	45.0	14.0	76%
	CHEM150.2	CHEM150 SLO2 - Balance chemical equations, including oxidation-reduction.	120.0	29.0	81%
CHEM150	CHEM150.3	CHEM150 SLO3 - Solve questions involving gas laws	50.0	9.0	85%
CHEI	CHEM150.4	CHEM150 SLO4 - Provide the quantum numbers for any specific electron.	34.0	25.0	58%
	CHEM150.5	CHEM150 SLO5 - Perform calculations involving thermodynamics.	33.0	26.0	56%
	CHEM150.6	CHEM150 SLO6 - Perform laboratory quantitative analysis.	5.0	12.0	29%
CHEM1	CHEM151.2	CHEM151 SLO2 - Perform equilibrium calculations.	10.0	14.0	42%

# SLO/CLO Data

CHEM150 21.00

CHEM120 0.021.00

34.00

CHEM151

# Spring 2018

Program		Term	¥ 🔻	Number Met
Chemistry	•	Spring 2018	•	Number Not Met
				Percent Met

6. Historical Course Performance: Chemistry- This is SLO assessment by course, including percent and number of students that met standards.

85.00

80%

100% 🗖

83.00 71%

net	above.	

Historical CLO Performance Table: Chemistry- This is a chart of the table

			Number Met	Number Not Met	Percent Met
CHEM1	CHEM120.1	CHEM120 SLO1 - Use scientific notation, significant figures, and dimensional analysis.	21.00	0.00	100%
	CHEM150.1	CHEM150 SLO1 - Perform stoichiometric calculations.	7.00	2.00	78%
	CHEM150.2	CHEM150 SLO2 - Balance chemical equations, including oxidation-reduction.	7.00	2.00	78%
CHEM150	CHEM150.3	CHEM150 SLO3 - Solve questions involving gas laws	27.00	7.00	79%
CHEN	CHEM150.4	CHEM150 SLO4 - Provide the quantum numbers for any specific electron.	30.00	6.00	83%
	CHEM150.5	CHEM150 SLO5 - Perform calculations involving thermodynamics.	7.00	2.00	78%
	CHEM150.6	CHEM150 SLO6 - Perform laboratory quantitative analysis.	7.00	2.00	78%





					18
	CHEM151.1	CHEM151 SLO1 - Perform kinetic calculations.	4.00	2.00	67%
	CHEM151.2	CHEM151 SLO2 - Perform equilibrium calculations.	4.00	2.00	67%
CHEM151	CHEM151.3	CHEM151 SLO3 - Perform thermodynamic calculations.	63.00	24.00	72%
CHEI	CHEM151.4	CHEM151 SLO4 - Define and explain concepts of equilibria.	4.00	2.00	67%
	CHEM151.5	CHEM151 SLO5 - Interpret a pH graph from an acid-base titration.	4.00	2.00	67%
	CHEM151.6	CHEM151 SLO6 - Perform qualitative analysis.	4.00	2.00	67%

# SLO/CLO Data Fall 2018



**6. Historical Course Performance: Chemistry-** This is SLO assessment by course, including percent and number of students that met standards.

Historical CLO Performance Table: Chemistry-	This is a chart of the table
above.	

CHEM150	38.0		185.0	83%
CHEM151	36.0	42.0	54%	
CHEM120	9.0			100% 🗖

			Number Met	Number Not Met	Percent Met
CHEM1	CHEM120.4	CHEM120 SLO4 - Define and give examples of chemical terms.	3.00	0.00	100%
	CHEM150.1	CHEM150 SLO1 - Perform stoichiometric calculations.	32.00	6.00	84%
	CHEM150.2	CHEM150 SLO2 - Balance chemical equations, including oxidation-reduction.	33.00	5.00	87%
CHEM150	CHEM150.3	CHEM150 SLO3 - Solve questions involving gas laws	11.00	3.00	79%
CHEI	CHEM150.4	CHEM150 SLO4 - Provide the quantum numbers for any specific electron.	87.00	18.00	83%
	CHEM150.5	CHEM150 SLO5 - Perform calculations involving thermodynamics.	11.00	3.00	79%
	CHEM150.6	CHEM150 SLO6 - Perform laboratory quantitative analysis.	11.00	3.00	79%
	CHEM151.1	CHEM151 SLO1 - Perform kinetic calculations.	7.00	6.00	54%
	CHEM151.2	CHEM151 SLO2 - Perform equilibrium calculations.	7.00	6.00	54%
M151	CHEM151.3	CHEM151 SLO3 - Perform thermodynamic calculations.	7.00	6.00	54%
CHEM15	CHEM151.4	CHEM151 SLO4 - Define and explain concepts of equilibria.	7.00	6.00	54%
	CHEM151.5	CHEM151 SLO5 - Interpret a pH graph from an acid-base titration.	7.00	6.00	54%
	CHEM151.6	CHEM151 SLO6 - Perform qualitative analysis.	7.00	6.00	54%

# SLO/CLO Data

# Spring 2019



**6. Historical Course Performance: Chemistry-** This is SLO assessment by course, including percent and number of students that met standards. Historical

CHEM150 3.00	54.00	95%			
			CHEM1	CHEM150.6	CHEM150 SLO6 - Perform laboratory quantitative analysis

Historical CLO Performance Table: Chemistry- This is a chart of the table above.

19	

Percent Met

95%

Number Not Met

3.00

Number Met

54.00



# Annual Update Student Learning Outcomes Packet

# III. Quality and Innovation in the Program and Curriculum Review

Please refer to the current SLO data set for your program found at: <u>http://research.hancockcollege.</u> <u>edu/student\_learning\_outcomes/matrix.html#Top</u>

a. Are you on track in your assessment plan for course and program SLOs? If not, please explain why.

b. Have you shared your assessments or improvement plans with your department, program or advisory committee? If so, what actions resulted? If not, how do you plan to do so in the future?

c. Did any of section, course or program improvement plans indicate that your program would benefit from specific resources in order to support student learning and/or faculty development? If so, please explain.

d. In reviewing your outcomes and assessments have you identified any and all that indicate a modification should be made to the course outline, the student learning outcomes or the program outcomes? Please state what modifications you will be making.

e. Have all course outlines been reviewed within the last 5 years? If not, please explain the plan to bring course outlines up to date and include time-lines for the review and submission to AP&P.

	ASSESS SCHEI		'rogram:			page of
Use one row for SLC 1.1 Using the rocke mission parameters determine the mass exit velocity of an id	D t equation and s, students will s fraction and	To be assessed in semester: Fall 2011, Fall 2012, Fall 2013	Assessment method (s) Student answers on final exam scores will be, graded anonymously by two RP instructors, on a scale of 1 to 5, using a predetermined rubric.	Resources needed to conduct assessment	Individual responsible for Improvement Plan Joe Yi	Date to complete review February 15, following the final exam semester.
1.2 Given mission ro students will lead a session to sketch or features	brainstorm	Spring 2012, Spring 2013, Spring 2014	precetermined rubinc. The outside experts will use a scoring rubric to judge the team leader's performance. They will also provide a qualitative strengths and weaknesses analysis of the final design, which will be a team grade.	May need room and meals for judge training.	Vesuvious McNuttal	May 15 of the semester wher the judging takes place.

\*This document is not in this packet, but can be found, if completed, on the Program Matrix (linked to the left). If not complete, a blank form can be found on the Student Learning Outcomes myHancock portal.

Expected Action	Action Type	Respondent	Action Taken	Date	Resourc
Allan Hancock College >> Math		MATH123 - Fall 2	017		Reques
	No action	Anonymous	The assessment data showed that (of the teacher who	2018-	
	type	, anonymous	entered data), 81.47% of their students met or exceeded		
strengths of your course?	960		the standard. Since our goal is 70%, we were well above	02.00	
			the goal. Almost every teacher who responded, stated		
			that students demonstrated an understanding of		
			hypothesis testing. Several teachers commented that		
			the standards were efficiently taught, that students		
			understood key points, and that only minor mistakes		
What did the assessment	No action	Anonymous	The comments on the weaknesses of the course were	2018-	
data indicate about the	type		varied according to the responding teachers. Some of	02-05	
weaknesses of your course?			the teachers mentioned that their students mixed up the		
			order of operations, mixed up their inequality symbols		
			when comparing the standardized test statistic to the		
			critical value, mixed up the way to verify normality on a 2		
			sample proportion hypothesis test with a 2 sample		
			proportion confidence interval. One teacher mentioned		
			that 1/3 of their class was not able to fully complete the		
			hypothesis test question. Other teachers mentioned that		
			students were unable to retain knowledge of hypothesis tests for the final exam and did not verify their apswers		

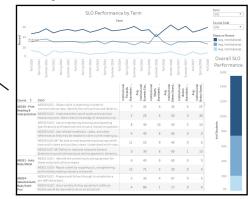
\*This section from the this document is under the "Action Plan". This sub-section contains any "Course Improvement Plan" that has been written for this course. If there is no text, then there was no Improvement plan submitted on eLumen. It will also have any resource requests that were entered into eLumen.

#### Assessments Summer 2017

Grammar and Vocabulary

SLO	Scored	Institutional Exceeds Standards	Institutional Meets Standards	Institutional Below Standards	N/A
SPAN101 SLO1 - Use grammar and vocabulary at the appropriate level.	21 of 87	10	2	6	3
Grammar and Vocabulary Te	est				
SLO	Scored	Institutional Exceeds Standards	Institutional Meets Standards	Institutional Below Standards	N/A
SPAN101 SLO1 - Use grammar and vocabulary at the appropriate level.	19 of 87	3	8	5	3

\*This section from the this document is under the heading "Assessments contains all of the outcomes that were measured and indicate performance. Below, you can find the dashboard with SLO performance by outcomes. You can filter based on outcome, discipline, and term. You can use the "Snipping Tool" to add any visual charts to your update. Also, you can use the data to make conclusions about assessment practices.



# **Context Statistics And Evidence**

Chemistry Date: 02/28/2019 Terms Spring 2018, Fall 2017, Summer 2017

# Summary

Statistic	Number of Courses	Courses
Courses in the Department	7	CHEM110, CHEM120, CHEM140, CHEM150, CHEM151, CHEM180, CHEM181
Courses with CSLOs	7	CHEM110, CHEM120, CHEM140, CHEM150, CHEM151, CHEM180, CHEM181
Courses without CSLOs	0	
Courses with CSLOs mapped to PSLOs	7	CHEM110, CHEM120, CHEM140, CHEM150, CHEM151, CHEM180, CHEM181
Courses without CSLOs mapped to PSLOs	0	
Courses with direct assessment of PSLOs	0	
Courses with CSLOs mapped to ILOs	7	CHEM110, CHEM120, CHEM140, CHEM150, CHEM151, CHEM180, CHEM181
Courses without CSLOs mapped to ILOs	0	
Courses with direct assessment of ILOs	0	
Courses with at least one planned Assessment	3	CHEM120, CHEM150, CHEM151
Courses with planned Assessments scored	2	CHEM150, CHEM151
Courses with some Assessments scored	1	CHEM120
Courses without any Assessment scored	0	
Courses with no planned Assessments	4	CHEM110, CHEM140, CHEM180, CHEM181
Courses with at least one planned Action Plan	7	CHEM110, CHEM120, CHEM140, CHEM150, CHEM151, CHEM180, CHEM181
Courses with Action Plan Responses	0	
Courses with some Action Plan Responses	0	
Courses without Action Plan Responses	7	CHEM120, CHEM110, CHEM140, CHEM150, CHEM151, CHEM180, CHEM181
Courses with no planned Action Plans	0	
CHEM110 - Chemistry ar	nd Society	1
SLOs	the macrosco » CHEM110 S	SLO1 - Describe the structure and composition of matter and its relationship pic properties of substances. SLO2 - Describe the nature and characteristics of chemical reactions. SLO3 - Apply their knowledge of chemistry to analyze current science and

» ILO 5 - Quantitative Literacy: Use mathematical concepts and models to analyze and solve

 » CHEM110 SLO5 - Demonstrate proficiency with basic chemistry apparatus and analyzing data from experiments.

 Mapped PSLOs
 Chemistry Program Outcomes

 % CHEM PSLO - Course doesn't map to a degree or certificate.

 ILO

ILO 5 - Quantitative Literacy

real life issues or problems.

CSLOs

			6 - Scientifi		,	ntific knowledge	and mathadal	logioo t	a access notest	
					iteracy: Use scie challenges.	entific knowledge	e and methodol	logies to	o assess potent	
Action Plans										
Fall 2017										
2017 Course Improvement Pla	n Action					···· <b>-</b> · · · ·		Data	Resource	
Expected Action	Туре		pondent		Ac	tion Taken		Date	Request	
Allan Hancock College >> Che	emistry >> CF	IEM110	) - Fall 2017							
Spring 2018 2017 Context Improvement Pla	an									
Expected Action	Action	Res	pondent		Ac	tion Taken		Date	Resource	
Allan Hancock College >> Che	Type		-	18					Request	
2017 Course Improvement Pla	-		- Opinig 20	10						
Expected Action	Action Type	Res	pondent		Ac	tion Taken		Date	Resource Request	
Allan Hancock College >> Che		IEM110	) - Spring 20	18						
CHEM120 - Introd	luctorv	Che	mistrv							
SLOs										
		» C⊦	IEM120 SL	.01 -	Use scientific no	otation, significa	nt figures, and	dimens	ional analysis.	
CSLOs					Write formulas a Solve problems					
00203					Define and give		-		iony.	
					Able to use the			us to pe	erform accurate	
		preci	se measur	emer	nts.					
			•	-	Outcomes					
Mapped PSLOs			Chemistry Program Outcomes							
			IEM PSLO	- Co	urse doesn't ma	o to a degree or	certificate.			
		ILO	5 Quantita	ativo	litoracy					
			ILO 5 - Quantitative Literacy » ILO 5 - Quantitative Literacy: Use mathematical concepts and models to analyze and sol							
			ife issues of						u	
Managath		ILO (	6 - Scientifi	c Lite	eracy					
Mapped ILOs			» ILO 6 - Scientific Literacy: Use scientific knowledge and methodologies to assess potenti solutions to real-life challenges.							
			ILO 2 - Critical Thinking & Problem Solving							
		» ILO 2 - Critical Thinking & Problem Solving: Explore issues through various information								
					e credibility and	significance of b	oth the information	ation an	d the source to	
Assessments		arriv	e at a reaso	oned	conclusion.					
Summer 2017										
SLO#1										
	Score	d	Institution Exceed		Institutional	Institutional	N/A			
SLO	SCOLE	a	Standar		Meets Standards	Below Standards	IN/A			
CHEM120 SLO1 - Use scientific notation, significant										
figures, and dimensional	18 of 1	51	16		1	1	0			
analysis. SLO5 use la apparatus acc	curatel:v.an	d prore	risoly							
			Institutio	nal	Institutional	Institutional		٦		
SLO	Score	ed	Exceed			Below Standards	N/A			
CHEM120 SLO5 - Able to use			Standar	uə						
the appropriate laboratory	28 of 1	51	22		4	1	1			
apparatus to perform accurate & precise measurements.										
SLO5 use lab apparatus a	ccuratelv &	precis	ely							
			Institutio		Institutional	Institutional				
SLO	Score	ed	Exceed Standar			Below Standards	N/A			
CHEM120 SLO5 - Able to use			orandar							
the appropriate laboratory	28 of 1	51	19		6	0	3			
apparatus to perform accurate & precise measurements.										
Spring 2018			1			1	1			
CHEM120										
	Score	ed	Institution Exceed		Institutional	Institutional	N/A			
SLO	00010		Standar		Meets Standards	Below Standards				

scientific notation, significant figures, and dimensional analysis.	0 of 25	254 0 0 0 0							
CHEM120 SLO2 - Write formulas and names of simple compounds and ions.	0 of 25	54	0		0	0	0		
CHEM120 SLO3 - Solve problems related to chemical equations and density.	0 of 25	54	0		0	0	0		
CHEM120 SLO4 - Define and give examples of chemical terms.	0 of 25	0 of 254			0	0	0		
CHEM120 SLO5 - Able to use the appropriate laboratory apparatus to perform accurate & precise measurements.	0 of 25	54	0		0	0	0		
SLO#1								_	
SLO	Score	d	Institutic Excee Standa	ds	Institutional Meets Standards	Institutional Below Standards	N/A		
CHEM120 SLO1 - Use scientific notation, significant figures, and dimensional analysis.	23 of 2	54	14		7	0	2		
Action Plans Fall 2017									
2017 Course Improvement Plan	Action	-			•	····		Data	Resource
Expected Action	Туре	-	pondent		Ac	tion Taken		Date	Request
Allan Hancock College >> Chen	nistry >> CH	EM120	) - Fall 2017	7					
Spring 2018 2017 Context Improvement Plar	<b>1</b>								
Expected Action	Action Type	Res	pondent		Ac	tion Taken	Date	Resource Request	
Allan Hancock College >> Chen	,	EM120	) - Spring 20	018					
2017 Course Improvement Plan	Action								Resource
Expected Action	Туре	Res	pondent		Ac	tion Taken		Date	Request
Allan Hancock College >> Chen	nistry >> CH	EM120	) - Spring 20	018					
CHEM140 - Intro C	)rganic	Ch	emistr	v					
81.00				3					
SLOs				<u> </u>					
		» CH orgai	EM140 SI EM140 SI nic compo	LO1 - LO2 - ounds.	Identify the cher		and hazards	of the co	ommon classes c
CSLOs		» CH orgar » CH and h » CH	EM140 SI EM140 SI nic compo EM140 SI nave a wo EM140 SI	LO1 - LO2 - unds. LO3 - rking	Identify the cher Demonstrate the knowledge of the Demonstrate the	mical properties	and hazards on sof identifyin the table of identifyin the table of table	of the co g chem cal techi	iical compounds niques.
		» CH organ » CH and h » CH subs	EM140 SI EM140 SI nic compo EM140 SI nave a wo EM140 SI titution rea	LO1 - LO2 - Junds. LO3 - rking LO4 - actions	Identify the cher Demonstrate the knowledge of the Demonstrate the s take place.	mical properties e common mear e various instrun	and hazards on sof identifyin the table of identifyin the table of table	of the co g chem cal techi	iical compounds niques.
		» CH organ » CH and h » CH subsi Cher	EM140 SI EM140 SI nic compo EM140 SI nave a wo EM140 SI titution rea <b>nistry Pro</b> nistry Pro	LO1 - LO2 - LO3 - rking LO3 - rking LO4 - actions	Identify the cher Demonstrate the knowledge of the Demonstrate the s take place. Dutcomes Outcomes	mical properties e common mear e various instrun	and hazards on the second s	of the co g chem cal techi	iical compounds niques.
CSLOs		» CH organ » CH and h » CH subsi Cher	EM140 SI EM140 SI nic compo EM140 SI nave a wo EM140 SI titution rea <b>nistry Pro</b> nistry Pro	LO1 - LO2 - LO3 - rking LO3 - rking LO4 - actions	Identify the cher Demonstrate the knowledge of the Demonstrate the s take place. Dutcomes Outcomes	mical properties e common mear e various instrun e mechanisms b	and hazards on the second s	of the co g chem cal techi	iical compounds niques.
CSLOs Mapped PSLOs		» CH orgar » CH and f » CH subst Cher Cher » CH ILO 6 » ILO	EM140 SI EM140 SI nic compo EM140 SI nave a wo EM140 SI titution rea <b>nistry Pro</b> EM PSLC S - Scientif O 6 - Scient	LO1 - LO2 - Junds. LO3 - rking I LO4 - actions <b>ogram</b> gram ( ) - Cou fic Lite	Identify the cher Demonstrate the knowledge of the Demonstrate the s take place. <b>n Outcomes</b> Outcomes urse doesn't map eracy iteracy: Use scie	mical properties e common mear e various instrun e mechanisms b	and hazards on the solution of identifyin the the carbon of the solution of th	of the co g chem cal techn common	iical compounds niques.
CSLOs		» CH organ » CH subst Cher Cher » CH ILO ( » ILO ( soluti ILO 2 » ILC 2	EM140 SI EM140 SI nic compo EM140 SI nave a wo EM140 SI titution rea nistry Prog EM PSLC S - Scientif O 6 - Scient ions to rea 2 - Critical O 2 - Critical	LO1 LO2 - LO2 - rking   LO4 - LO4 - LO4 - Co Sgram ( Co Sgram ( Co Sgram)	Identify the cher Demonstrate the knowledge of the Demonstrate the s take place. Dutcomes Dutcomes urse doesn't map eracy iteracy: Use scie challenges. ing & Problem S nking & Problem	mical properties e common mean e various instrum e mechanisms b o to a degree or entific knowledge Golving Solving: Explor	and hazards of identifyin nental analytic y which the co certificate.	of the co g chem cal techn common	iical compounds niques. addition and to assess potenti ous information
CSLOs Mapped PSLOs		» CH organ » CH subs Cher Cher » CH ILO ( » CH ILO ( » ILC soluti ILO 2 » ILC source	EM140 SI EM140 SI nic compo EM140 SI inave a wo EM140 SI titution rea nistry Prog EM PSLC Constro rea 2 - Scientif 0 6 - Scientif 2 - Critical 0 2 - Critical 0 2 - Critical 0 2 - Critical	LO1 - LO2 - LO2 - LO3 - rking   LO4 - cactions ogram () - Coo fic Lite dal-life ( Think al Thin ate th	Identify the cher Demonstrate the knowledge of the Demonstrate the s take place. Dutcomes Dutcomes urse doesn't map eracy iteracy: Use scie challenges. ing & Problem S nking & Problem	mical properties e common mean e various instrum e mechanisms b o to a degree or entific knowledge Golving Solving: Explor	and hazards of identifyin nental analytic y which the co certificate.	of the co g chem cal techn common	ical compounds niques. addition and to assess potenti
CSLOs Mapped PSLOs		» CH organ » CH subs Cher Cher » CH ILO ( » CH ILO ( » ILC soluti ILO 2 » ILC source	EM140 SI EM140 SI nic compo EM140 SI inave a wo EM140 SI titution rea nistry Prog EM PSLC Constro rea 2 - Scientif 0 6 - Scientif 2 - Critical 0 2 - Critical 0 2 - Critical 0 2 - Critical	LO1 - LO2 - LO2 - LO3 - rking   LO4 - cactions ogram () - Coo fic Lite dal-life ( Think al Thin ate th	Identify the cher Demonstrate the knowledge of the Demonstrate the s take place. Dutcomes Unco	mical properties e common mean e various instrum e mechanisms b o to a degree or entific knowledge Golving Solving: Explor	and hazards of identifyin nental analytic y which the co certificate.	of the co g chem cal techn common	iical compounds niques. addition and to assess potenti ous information
CSLOs Mapped PSLOs Mapped ILOs Action Plans		» CH organ » CH subs Cher Cher » CH ILO ( » CH ILO ( » ILC soluti ILO 2 » ILC source	EM140 SI EM140 SI nic compo EM140 SI inave a wo EM140 SI titution rea nistry Prog EM PSLC Constro rea 2 - Scientif 0 6 - Scientif 2 - Critical 0 2 - Critical 0 2 - Critical 0 2 - Critical	LO1 - LO2 - LO2 - LO3 - rking   LO4 - cactions ogram () - Coo fic Lite dal-life ( Think al Thin ate th	Identify the cher Demonstrate the knowledge of the Demonstrate the s take place. Dutcomes Unco	mical properties e common mean e various instrum e mechanisms b o to a degree or entific knowledge Golving Solving: Explor	and hazards of identifyin nental analytic y which the co certificate.	of the co g chem cal techn common	ical compounds niques. addition and to assess potention ous information nd the source to
CSLOs Mapped PSLOs Mapped ILOs Action Plans Fall 2017	Action	» CH organ » CH and I » CH Cher Cher » CH ILO 6 » ILC soluti ILO 2 » ILC sourc arrive	EM140 SI EM140 SI nic compo EM140 SI inave a wo EM140 SI titution rea nistry Prog EM PSLC Constro rea 2 - Scientif 0 6 - Scientif 2 - Critical 0 2 - Critical 0 2 - Critical 0 2 - Critical	LO1 - LO2 - LO2 - LO3 - rking   LO4 - cactions ogram () - Coo fic Lite dal-life ( Think al Thin ate th	Identify the cher Demonstrate the knowledge of the Demonstrate the stake place. <b>Doutcomes</b> Outcomes urse doesn't map eracy iteracy: Use scie challenges. ing & Problem S hking & Problem S hking & Problem S	mical properties e common mean e various instrum e mechanisms b o to a degree or entific knowledge Golving Solving: Explor	and hazards of identifyin nental analytic y which the co certificate.	of the co g chem cal techn common	ical compounds niques. addition and to assess potention ous information nd the source to
CSLOs Mapped PSLOs Mapped ILOs Action Plans Fall 2017 2017 Course Improvement Plan Expected Action Allan Hancock College >> Chen	Action Type	» CH organ » CH and f » CH Subs: Cher » CH ILO 6 » ILC soluti ILO 2 » ILC soluti Res	EM140 SI EM140 SI nic compo EM140 SI inave a wo EM140 SI inave a wo EM140 SI initry Prog EM PSLC 6 - Scientif 0 6 - Scientif 0 6 - Scientif 2 - Critical 0 2 - Critical	LO1 - LO2 - LO2 - rking l LO3 - rking l LO4 - actions ogram () - Col tic Lite tal-life ( Think al Thin ate th soned	Identify the cher Demonstrate the knowledge of the Demonstrate the stake place. <b>Doutcomes</b> Outcomes urse doesn't map eracy iteracy: Use scie challenges. ing & Problem S hking & Problem S hking & Problem S	mical properties e common mean e various instrum e mechanisms b o to a degree or entific knowledge Solving Solving: Explor significance of b	and hazards of identifyin nental analytic y which the co certificate.	of the co g chem cal techn ommon	ical compounds niques. addition and to assess potention ous information nd the source to
CSLOs Mapped PSLOs Mapped ILOs Action Plans Fall 2017 2017 Course Improvement Plan Expected Action Allan Hancock College >> Chen Spring 2018	Action Type nistry >> CH	» CH organ » CH and f » CH Subs: Cher » CH ILO 6 » ILC soluti ILO 2 » ILC soluti Res	EM140 SI EM140 SI nic compo EM140 SI inave a wo EM140 SI inave a wo EM140 SI initry Prog EM PSLC 6 - Scientif 0 6 - Scientif 0 6 - Scientif 2 - Critical 0 2 - Critical	LO1 - LO2 - LO2 - rking l LO3 - rking l LO4 - actions ogram () - Col tic Lite tal-life ( Think al Thin ate th soned	Identify the cher Demonstrate the knowledge of the Demonstrate the s take place. I Outcomes U	mical properties e common mean e various instrum e mechanisms b o to a degree or entific knowledge Solving Solving: Explor significance of b	and hazards of identifyin nental analytic y which the co certificate.	of the co g chem cal techn ommon	ical compounds niques. addition and to assess potenti ous information nd the source to Resource
CSLOs Mapped PSLOs Mapped ILOs Action Plans Fall 2017 2017 Course Improvement Plan Expected Action Allan Hancock College >> Chen	Action Type nistry >> CH Action	» CH organ » CH subs: Cher Cher » CH ILO 6 » ILC soluti ILO 2 » ILC source arrive	EM140 SI EM140 SI nic compo EM140 SI inave a wo EM140 SI inave a wo EM140 SI initry Prog EM PSLC 6 - Scientif 0 6 - Scientif 0 6 - Scientif 2 - Critical 0 2 - Critical	LO1 - LO2 - LO2 - rking l LO3 - rking l LO4 - actions ogram () - Col tic Lite tal-life ( Think al Thin ate th soned	Identify the cher Demonstrate the knowledge of the Demonstrate the s take place. <b>Dutcomes</b> Outcomes urse doesn't map eracy iteracy: Use scie challenges. ing & Problem S nking & Problem S nking & Problem S nking & Problem S nking & Conclusion.	mical properties e common mean e various instrum e mechanisms b o to a degree or entific knowledge Solving Solving: Explor significance of b	and hazards of identifyin nental analytic y which the co certificate.	of the co g chem cal techn ommon	ical compounds niques. addition and to assess potenti ous information nd the source to Resource Request
CSLOs Mapped PSLOs Mapped ILOs Action Plans Fall 2017 2017 Course Improvement Plan Expected Action Allan Hancock College >> Chen Spring 2018 2017 Context Improvement Plan	Action Type nistry >> CH Action Type	» CH organ » CH and f » CH Soluts ILO for soluts ILO 2 » ILC soluts ILO 2 soluts ILO 3 Soluts ILO 3 SOLUTA SOLUT	EM140 SI EM140 SI nic compo EM140 SI nave a wo EM140 SI titution rea <b>nistry Pro</b> EM PSLC 6 - Scientif 0 6 - Scientif 0 6 - Scientif 2 - Critical 0 2 - Critical 2 - Critical 0 2 - Critical 0 2 - Critical 0 2 - Critical 0 2 - Critical 0 - Fall 2017	LO1 - LO2 - LO2 - rking J LO4 - actions ogram ( D - Cool fic Lite tal-life ( Think al-life ( Think al Thin ate th soned	Identify the cher Demonstrate the knowledge of the Demonstrate the s take place. <b>Dutcomes</b> Outcomes urse doesn't map eracy iteracy: Use scie challenges. ing & Problem S nking & Problem S nking & Problem S nking & Problem S nking & Conclusion.	mical properties e common mean e various instrum e mechanisms b o to a degree or entific knowledge Solving Solving: Explor significance of b	and hazards of identifyin nental analytic y which the co certificate.	of the co g chem cal tech ommon blogies ugh varia nation a <b>Date</b>	ical compounds niques. addition and to assess potenti ous information nd the source to Resource Request
CSLOs Mapped PSLOs Mapped ILOs Action Plans Fall 2017 2017 Course Improvement Plan Expected Action Allan Hancock College >> Chen Spring 2018 2017 Context Improvement Plan	Action Type nistry >> CH Action Type nistry >> CH	» CH organ » CH and f » CH Soluts ILO for soluts ILO 2 » ILC soluts ILO 2 soluts ILO 3 Soluts ILO 3 SOLUTA SOLUT	EM140 SI EM140 SI nic compo EM140 SI nave a wo EM140 SI titution rea <b>nistry Pro</b> EM PSLC 6 - Scientif 0 6 - Scientif 0 6 - Scientif 2 - Critical 0 2 - Critical 2 - Critical 0 2 - Critical 0 2 - Critical 0 2 - Critical 0 2 - Critical 0 - Fall 2017	LO1 - LO2 - LO2 - rking J LO4 - actions ogram ( D - Cool fic Lite tal-life ( Think al-life ( Think al Thin ate th soned	Identify the cher Demonstrate the knowledge of the Demonstrate the s take place. <b>Dutcomes</b> Outcomes urse doesn't map eracy iteracy: Use scie challenges. ing & Problem S nking & Problem S nking & Problem S nking & Problem S nking & Conclusion.	mical properties e common mean e various instrum e mechanisms b o to a degree or entific knowledge Solving Solving: Explor significance of b	and hazards of identifyin nental analytic y which the co certificate.	of the co g chem cal tech ommon blogies ugh varia nation a <b>Date</b>	ical compounds niques. addition and to assess potenti ous information nd the source to Resource Request

Allan Hancock College >> Cher	nistry >> CHEM140 - Spring 2018
CHEM150 - Gener	al Chemistry 1
SLOs	
	» CHEM150 SLO1 - Perform stoichiometric calculations.
	» CHEM150 SLO2 - Balance chemical equations, including oxidation-reduction.
CSLOs	» CHEM150 SLO3 - Solve questions involving gas laws
CSLUS	» CHEM150 SLO4 - Provide the quantum numbers for any specific electron.
	» CHEM150 SLO5 - Perform calculations involving thermodynamics.
	» CHEM150 SLO6 - Perform laboratory quantitative analysis.
	Chemistry Program Outcomes
	Chemistry Program Outcomes
Mapped PSLOs	» CHEM PSLO - The student will demonstrate mastery of the approach and rationale of th
Mapped PSLOS	scientific method and be able to apply these principles to solve problems.
	» CHEM PSLO - The student will demonstrate mastery of stoichiometric calculations.
	» CHEM PSLO - The student will demonstrate mastery of laboratory technique.
	ILO
	ILO 5 - Quantitative Literacy
	» ILO 5 - Quantitative Literacy: Use mathematical concepts and models to analyze and so
	real life issues or problems.
	ILO 6 - Scientific Literacy
Mapped ILOs	» ILO 6 - Scientific Literacy: Use scientific knowledge and methodologies to assess poten
	solutions to real-life challenges.
	ILO 2 - Critical Thinking & Problem Solving
	» ILO 2 - Critical Thinking & Problem Solving: Explore issues through various information
	sources; evaluate the credibility and significance of both the information and the source to
	arrive at a reasoned conclusion.

# Assessments

Fall 2017

CHEM150\_ALLSLOS\_SMG\_F2017

SLO	Scored	Institutional Exceeds Standards	Institutional Meets Standards	Institutional Below Standards	N/A
CHEM150 SLO1 - Perform stoichiometric calculations.	42 of 151	3	28	11	0
CHEM150 SLO2 - Balance chemical equations, including oxidation-reduction.	42 of 151	9	22	11	0
CHEM150 SLO3 - Solve questions involving gas laws	42 of 151	5	33	4	0
CHEM150 SLO4 - Provide the quantum numbers for any specific electron.	42 of 151	15	12	15	0
CHEM150 SLO5 - Perform calculations involving thermodynamics.	42 of 151	1	25	16	0
All SLOs					

SLO	Scored	Institutional Exceeds Standards	Institutional Meets Standards	Institutional Below Standards	N/A
CHEM150 SLO1 - Perform stoichiometric calculations.	17 of 151	2	12	3	0
CHEM150 SLO2 - Balance chemical equations, including oxidation-reduction.	17 of 151	2	11	4	0
CHEM150 SLO3 - Solve questions involving gas laws	17 of 151	2	10	5	0
CHEM150 SLO4 - Provide the quantum numbers for any specific electron.	17 of 151	2	5	10	0
CHEM150 SLO5 - Perform calculations involving thermodynamics.	17 of 151	2	5	10	0
CHEM150 SLO6 - Perform laboratory quantitative analysis.	17 of 151	2	3	12	0
SLO2 Balance Redox					

SLO	Scored	Institutional Exceeds Standards	Institutional Meets Standards	Institutional Below Standards	N/A
CHEM150 SLO2 - Balance chemical equations, including oxidation-reduction.	30 of 151	18	6	4	2

SLO2 Balance Redox

Intermediations, including       31 of 151       18       10       3       0         SLO2 Balance Rodox       Scoodd       Institutional Action Statements, Balandaria, Balandari, Balandaria, Balandaria, Balandaria, Balanda	SLO	Score	ed	Institutio Excee Standa	ds	Institutional Meets Standards	Institutional Below Standards	N/A		
Stored         Board Bundards Bundards         Institutional Weeks Standards Book Study - Bundards         N/A           CHEM 50 SL02 - Balance chemical equators, including gas law         31 of 151         13         11         7         0           Spring 2018 gas law         Scored         Exceeded Standards         Institutional Meets Standards         N/A           SL0         Scored         Institutional Exceeded Action         N/A         2           SL0         Scored         To         16         4         2           SU7 Course Improvement Plan         Exceeded Action         Action Taken         Date         Resource Request           SU7 Course Improvement Plan         Expected Action         Action Taken         Date         Resource Request           SU1C Course Improvement Plan         Expected Action         Action Taken         Date         Re	CHEM150 SLO2 - Balance chemical equations, including oxidation-reduction.	31 of 1	51			10	3	0		
Sto         Stored         Exceeds Bandroff         Weeks Standards Poly         N/A           DetEM 159 SLO2 - Balance deviced aquators, including 31 of 151         13         11         7         0           Spring 2018 gas law         Stored         Stored         Institutional Exceeded Standards         Institutional Institutional Store         Institutional Institutional Store         Institutional Institutional Store         N/A           Stored         Stored         Stored Standards         Institutional Institutional Store         Institutional Institutional Store         N/A           Stored         Stored         Stored Standards         Institutional Institutional Institutional Store         N/A           Stored         Stored         Stored Standards         Institutional Institentition Institutional Institentinstitutional Institution Instit	SLO2 Balance Redox	1								
Action equations, including       31 of 151       13       11       7       0         Spring 2018       gas low       Scored       Final status       NA         SLO       Scored       Scored       Scored       Resource Request         Action Plans       Scored       Scored       Scored       Resource Request         Action Improvement Plan       Scored       Action Taken       Date       Resource Request         VIT Course Improvement Plan       Scored       Action Taken       Date       Request         VIT Course Improvement Plan       Scored       Action Taken       Date       Request         VIT Course Impro	SLO	Score	ed	Excee	ds			N/A		
Spring 2018 gas law         L         Image and the set of	chemical equations, including	31 of 1	51	13		11	7	0		
Description         SLO       Scored       FieldWindow       Institutional       Institutional       NA         SLO       Scored       Standards       Meets Standards       NA         SLO       Scored       Standards       Meets Standards       NA         SLO       Scored       Standards       NA       A       2         SLO       Provide the parameters in any para										
Stored         Scored         Exceeds         Meet Model         Institutional Meet Standards         N/A           DHEM 50 LO3 - Solve justions involving as laws         27 of 94         9         11         5         2           SLO4 - quantum numbers										
upeation invoking gas law       27 of 94       9       11       5       2         SLO4 - quantum numbers       institutional Medis Standards       NiA         SLO       Scored       Scored       Institutional Medis Standards       NiA         SLO       Provide the yearting       20 of 94       7       16       4       2         SLO       Provide the yearting       20 of 94       7       16       4       2         SLO       Provide the yearting       Resource Request       Request       Request       Request         Stored       Action Plans       Action Taken       Date       Resource Request       Request         Stored       Action Taken       Date       Resource Request       Request       Not Resource Request         Wain Hancock College >> Chemistry >> CHEM150 - Spring 2018       Expected Action       Date       Resource Request         Wain Hancock College >> Chemistry >> CHEM151 SLO1 - Perform kinetic calculations.       > CHEM151 SLO2 - Perform equilibrium calculations.       > CHEM151 SLO4 - Define and explain concepts of equilibria.       > CHEM151 SLO4 - Define and explain concepts of equilibria.       > CHEM151 SLO4 - Define and explain concepts of equilibria.       > CHEM151 SLO4 - Define and explain concepts of equilibria.       > CHEM151 SLO4 - Define and explain concepts of equilibria.       > CHEM151 SLO4 - Def	SLO	Score	ed	Excee	eds			N/A		
Scored         Institutional Standards         Institutional Meets Standards         Institutional Bolow Standards         N/A           PRM H50 SLO4 - Provide the yaantun numbers for any peorite electron.         29 of 94         7         16         4         2           Action Plans Fail 2017         To         4         2         2         6         4         2           O'7 Course improvement Plan         Chion Plans Fail 2017         Action Taken         Date         Resource Request           Wann Hancock College >> Chemistry >> CHEM150 - Fail 2017         Spring 2018         Resource         Resource           Stor Course Improvement Plan         Action Taken         Date         Resource Request         Resource           Stor Course Improvement Plan         Action Taken         Date         Resource Request         Request           VIG Course Improvement Plan         Action Taken         Date         Resource Request         Request           VIG Course Improvement Plan         Action Taken         Date         Resource Request         Request           VIG Course Improvement Plan         Action Taken         Date         Resource Request         Resource Request           VIG Course Improvement Plan         Action Taken         Date         Resource Request           SLO S	questions involving gas laws	27 of 9	94	9		11	5	2		
DHEMISO SLO4 - Provide the guardination of the stress of any guardine quedres for any guardine quedres of any guardine quedres of the stress of th	·	Score	ed	Excee	eds			N/A		
Action Plans Fall 2017       Action Type       Respondent       Action Taken       Date       Resource Request         Spring 2018       Ot7 Contral Improvement Plan       Action Taken       Date       Resource Request         V17 Contrast Improvement Plan       Action Taken       Date       Resource Request         V12 Contrast Improvement Plan       Action Taken       Date <td>CHEM150 SLO4 - Provide the quantum numbers for any</td> <td></td> <td>94</td> <td></td> <td>iius</td> <td>16</td> <td>4</td> <td>2</td> <td></td> <td></td>	CHEM150 SLO4 - Provide the quantum numbers for any		94		iius	16	4	2		
Expected Action         Action Type         Respondent (Provided Action Type)         Respondent (Provided Action Type)         Resource (Request)           Spring 2018         2017 Context Improvement Plan         Action Taken         Date         Resource (Request)           Allan Hancock College >> Chemistry >> CHEM150 - Spring 2018         Action Taken         Date         Resource (Request)           Allan Hancock College >> Chemistry >> CHEM150 - Spring 2018         Action Taken         Date         Resource (Request)           Allan Hancock College >> Chemistry >> CHEM150 - Spring 2018         Action Taken         Date         Resource (Request)           Allan Hancock College >> Chemistry >> CHEM150 - Spring 2018         Action Taken         Date         Resource (Request)           Allan Hancock College >> Chemistry >> CHEM151 SLO1 - Perform kinetic calculations.	Action Plans					1				
Alian Hancock College >> Chemistry >> CHEM150 - Fall 2017           Spring 2018           2017 Course timprovement Plan           Expected Action         Action Type         Respondent Respondent Type         Action Taken         Date         Resource Request           2017 Course Improvement Plan         Action Taken         Date         Resource Request           Allan Hancock College >> Chemistry >> CHEM150 - Spring 2018         Action Taken         Date         Resource Request           Allan Hancock College >> Chemistry >> CHEM150 - Spring 2018         Action Taken         Date         Resource Request           Allan Hancock College >> Chemistry >> CHEM151 SLO1 - Perform kinetic calculations. <ul> <li>&gt; CHEM151 SLO2 - Perform equilibrium calculations. <li>&gt; CHEM151 SLO3 - Perform thermodynamic calculations. <li>&gt; CHEM151 SLO5 - Interpret a pH graph from an acid-base titration. <ul> <li>&gt; CHEM151 SLO5 - Interpret a pH graph from an acid-base titration. <ul> <li>&gt; CHEM151 SLO5 - Interpret a pH graph from an acid-base titration.</li> <li>&gt; CHEM151 SLO5 - The student will demonstrate mastery of the approach and rationale of the scintlife method and be able to apply these principles to solve problems.</li> <li>&gt; CHEM PSLO - The student will demonstrate mastery of laboratory technique.</li> </ul> </li> <li>ILO 4. Coantitative Literacy: Use scientific knowledge and methodologies to assess potenti solutions to real-life challenges. ILO 6. Scientific Literacy <ul>                 ILO 2. Critical Thinking 8 Probl</ul></li></ul></li></li></li></ul>	•	Action	Res	pondent		Ac	tion Taken		Date	
Spring 2018         Action Type         Respondent Request         Action Taken         Date         Resource Request           Name Hancock College >> Chemistry >> CHEM150 - Spring 2018         0017 Course Improvement Plan         Date         Resource Request           017 Course Improvement Plan         Action Taken         Date         Resource Request           017 Course Improvement Plan         Action Taken         Date         Resource Request           Value Hancock College >> Chemistry >> CHEM150 - Spring 2018         Expected Action         Action Taken         Date         Resource Request           Allan Hancock College >> Chemistry >> CHEM150 - Spring 2018         EXPEcted Action         Action Taken         Date         Resource           HEM151 - General Chemistry 2         SLOS <ul> <li>CHEM151 SLO2 - Perform equilibrium calculations.</li> <li>CHEM151 SLO5 - Interpret a pH graph from an acid-base titration.</li> <li>CHEM151 SLO5 - Interpret a pH graph from an acid-base titration.</li> <li>CHEM151 SLO5 - The student will demonstrate mastery of the approach and rationale of the scientific method and be able to apply these principles to solve problems.</li> <li>CHEM PSLO - The student will demonstrate mastery of sloichiometric calculations.</li> <li>CHEM PSLO - The student will demonstrate mastery of laboratory technique.</li> </ul> <li>Mapped ILOs</li> <li>Mapped ILOs</li> <li>Mapped ILOs</li> <li>C Q - Ortical Thinking &amp; Problem Solving</li> <li>ILO 6 - Scientific Literacy: Use s</li>	•			-	7					Request
Expected Action         Action Type         Respondent (Request)         Action Taken         Date         Resource Request           Wian Hancock College >> Chemistry >> CHEM150 - Spring 2018	Ŭ									
Expected Action         Type         Respondent         Action Taken         Date         Request           Wilan Hancock College >> Chemistry >> CHEM150 - Spring 2018	2017 Context Improvement Pla	an								
Expected Action         Action Type Type         Respondent Respondent         Action Taken         Date         Resource Request           Wlan Hancock College >> Chemistry >> CHEM150 - Spring 2018         Image: College >> Chemistry >> CHEM151 SLO1 - Perform kinetic calculations.         Respondent	-	Туре		-		Ac	tion Taken		Date	
Expected Action Type         Action Type         Respondent Request         Action Taken         Date         Resource Request           Mian Hancock College >> Chemistry >> CHEM150 - Spring 2018         Image: Chemistry >> CHEM151 SLO1 - Perform kinetic calculations.         Image: Chemistry >> CHEM151 SLO2 - Perform equilibrium calculations.         Image: Chemistry >> CHEM151 SLO2 - Perform mendphanic calculations.           SLOS         Image: Chemistry Program Outcomes         Image: Chemistry Program Outcomes         Image: Chemistry Program Outcomes           CSLOS         Chemistry Program Outcomes         Chemistry Program Outcomes         Image: Chemistry Program Outcomes           Vapped PSLOS         Image: Chemistry PSLO - The student will demonstrate mastery of the approach and rationale of the scientific method and be able to apply these principles to solve problems.         Image: Chemistry Program Outcomes           Vapped PSLOS         Image: Chemistry Program Outcomes         Image: Chemistry Program Outcomes           Vapped ILOS         Image: Chemistry Program Outcomes         Image: Chemistry Program Outcomes           Vapped ILOS         Image: Chemistry Program Outcomes         Image: Chemistry Program Outcomes           VID 1: LO         Chemistry Program Outcomes         Image: Chemistry Program Outcomes           VID 1: Chemistry Program Outcomes         Image: Chemistry Program Outcomes         Image: Chemistry Program Outcomes               VID 2: Chemistry Propremistry Prog			IEM150	- Spring 2	018					
HEM151 - General Chemistry 2         SLOs         SLOs         CSLOS         CSLOS         CSLOS         CSLOS         CSLOS         CSLOS         CSLOS         CCSLOS         CHEM151 SLO2 - Perform equilibrium calculations.         > CHEM151 SLO3 - Perform thermodynamic calculations.         > CHEM151 SLO4 - Define and explain concepts of equilibria.         > CHEM151 SLO5 - Interpret a pH graph from an acid-base titration.         > CHEM151 SLO6 - Perform qualitative analysis.         Chemistry Program Outcomes         Chemistry Program Outcomes         > CHEM PSLO - The student will demonstrate mastery of the approach and rationale of the scientific method and be able to apply these principles to solve problems.         > CHEM PSLO - The student will demonstrate mastery of laboratory technique.         ILO         ILO S         Wapped ILOs         Wapped ILOs         VIAPPE - Critical Thinking & Problem Solving         > ILO 6 - Scientific Literacy: Use scientific knowledge and methodologies to assess potenti solutions to real-life challenges.         ILO 2 - Critical Thinking & Problem Solving         > ILO 2 - Critical Thinking & Problem Solving         > ILO 2 - Critical Thinking & Problem Solving: Explore issues through various information sources; evaluate th	•	Action	Res	pondent		Ac	tion Taken		Date	
SLOs       » CHEM151 SLO1 - Perform kinetic calculations.         » CHEM151 SLO2 - Perform equilibrium calculations.       » CHEM151 SLO3 - Perform thermodynamic calculations.         » CHEM151 SLO4 - Define and explain concepts of equilibria.       » CHEM151 SLO5 - Interpret a pH graph from an acid-base titration.         » CHEM151 SLO6 - Perform qualitative analysis.       Chemistry Program Outcomes         Mapped PSLOs       » CHEM PSLO - The student will demonstrate mastery of the approach and rationale of the scientific method and be able to apply these principles to solve problems.         » CHEM PSLO - The student will demonstrate mastery of stoichiometric calculations.       » CHEM PSLO - The student will demonstrate mastery of laboratory technique.         ILO       ILO 5 - Quantitative Literacy       w ILO 5 - Quantitative Literacy:         will LO 6 - Scientific Literacy       w ILO 6 - Scientific Literacy:         will LO 6 - Scientific Literacy:       w ILO 6 - Scientific Literacy:         will 0 6 - Scientific Literacy:       w ILO 2 - Critical Thinking & Problem Solving         will 0 2 - Critical Thinking & Problem Solving       w ILO 2 - Critical Thinking & Problem Solving:         will 0 2 - Critical Thinking & Problem Solving:       x ILO 2 - Critical Thinking & Problem Solving:         will 0 2 - Critical Thinking & Problem Solving:       x ILO 2 - Critical Thinking & Problem Solving:         x ILO 2 - Critical Thinking & Problem Solving:       x ILO 2 - Critical Thinking & Problem Solving:	Allan Hancock College >> Che	-			018					
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CSLOS <sup>*</sup> CHEM151 SLO2 - Perform equilibrium calculations.          CSLOS <sup>*</sup> CHEM151 SLO3 - Perform thermodynamic calculations.          Website <sup>*</sup> CHEM151 SLO3 - Define and explain concepts of equilibria.          Website <sup>*</sup> CHEM151 SLO3 - Define and explain concepts of equilibria.          Website <sup>*</sup> CHEM151 SLO3 - Interpret a pH graph from an acid-base titration.          Website <sup>*</sup> CHEM151 SLO3 - Perform qualitative analysis.          Chemistry Program Outcomes          Chemistry Program Outcomes          CHEM PSLO - The student will demonstrate mastery of the approach and rationale of the scientific method and be able to apply these principles to solve problems.          Website <sup>*</sup> CHEM PSLO - The student will demonstrate mastery of laboratory technique.          ILO          ILO - The student will demonstrate mastery of laboratory technique.          ILO 5 - Quantitative Literacy:          N ILO 5 - Quantitative Literacy: Use mathematical concepts and models to analyze and sol real life issues or problems.          Wapped ILOS          N ILO 6 - Scientific Literacy: Use scientific knowledge and methodologies to assess potenti solutions to real-life challenges.          ILO 2 - Critical Thinking & Problem Solving          N ILO 2 - Critical Thinking & Problem Solving: Explore issues through various information anources; evaluate the credibility and signif	HEM151 - Gener	al Cher	mist	ry ∠						
CSLOS       > CHEM151 SLO3 - Perform thermodynamic calculations.         > CHEM151 SLO4 - Define and explain concepts of equilibria.         > CHEM151 SLO5 - Interpret a pH graph from an acid-base titration.         > CHEM151 SLO6 - Perform qualitative analysis.         Mapped PSLOs         Mapped PSLOs         Chemistry Program Outcomes         > CHEM PSLO - The student will demonstrate mastery of the approach and rationale of the scientific method and be able to apply these principles to solve problems.         > CHEM PSLO - The student will demonstrate mastery of stoichiometric calculations.         > CHEM PSLO - The student will demonstrate mastery of laboratory technique.         ILO         ILO 5 - Quantitative Literacy         > ILO 6 - Scientific Literacy         > ILO 2 - Critical Thinking & Problem Solving         > ILO 2 - Critical Thinking & Problem Solving         NLO 2 - Critical Thinking & Problem Solving         = ILO 2 - Critical Thinking & Problem Solving:         = Scientific Literacy:         > ILO 6 - Scientific Literacy:         > ILO 2 - Critical Thinking & Problem Solving:		al Chei	mist	ry ∠						
Schements          » CHEM151 SLO4 - Define and explain concepts of equilibria.         » CHEM151 SLO5 - Interpret a pH graph from an acid-base titration.         » CHEM151 SLO6 - Perform qualitative analysis.          Mapped PSLOs          Chemistry Program Outcomes         Chemistry Program Outcomes         » CHEM PSLO - The student will demonstrate mastery of the approach and rationale of the scientific method and be able to apply these principles to solve problems.         » CHEM PSLO - The student will demonstrate mastery of stoichiometric calculations.         » CHEM PSLO - The student will demonstrate mastery of stoichiometric calculations.         » CHEM PSLO - The student will demonstrate mastery of laboratory technique.         ILO         ILO 5 - Quantitative Literacy:		al Chei			LO1 -	Perform kinetic	calculations.			
* CHEM151 SLO4 - Define and explain concepts of equilibria.         * CHEM151 SLO5 - Interpret a pH graph from an acid-base titration.         * CHEM151 SLO6 - Perform qualitative analysis.         Chemistry Program Outcomes         * CHEM PSLO5         Chemistry Program Outcomes         * CHEM PSLO - The student will demonstrate mastery of the approach and rationale of the scientific method and be able to apply these principles to solve problems.         * CHEM PSLO - The student will demonstrate mastery of stoichiometric calculations.         * CHEM PSLO - The student will demonstrate mastery of laboratory technique.         ILO         ILO5 - Quantitative Literacy         * ILO 5 - Quantitative Literacy:         * ILO 6 - Scientific Literacy:         * ILO 2 - Critical Thinking & Problem Solving         * ILO 2 - Critical Thinking & Problem Solving         * ILO 2 - Critical Thinking & Problem Solving:         * ILO 2 - Critical Thinking & Problem Solving:         * ILO 2 - Critical Thinking & Problem Solving:         * ILO 2 - Critical Thinking & Problem Solving:         * ILO 2 - Critical Thinking & Problem S		ral Chei	» CH	IEM151 S						
* CHEM151 SLO6 - Perform qualitative analysis.         Mapped PSLOs       Chemistry Program Outcomes Chemistry Program Outcomes         * CHEM PSLO - The student will demonstrate mastery of the approach and rationale of the scientific method and be able to apply these principles to solve problems.         * CHEM PSLO - The student will demonstrate mastery of stoichiometric calculations.         * CHEM PSLO - The student will demonstrate mastery of laboratory technique.         ILO         ILO 5 - Quantitative Literacy         * ILO 5 - Quantitative Literacy:         * ILO 6 - Scientific Literacy         * ILO 2 - Critical Thinking & Problem Solving         * ILO 2 - Critical Thinking & Problem Solving         * ILO 2 - Critical Thinking & Problem Solving         * ILO 2 - Critical Thinking & Problem Solving         * ILO 2 - Critical Thinking & Problem Solving         * ILO 2 - Critical Thinking & Problem Solving         * ILO 2 - Critical Thinking & Problem Solving         * ILO 2 - Critical Thinking & Problem Solving         * ILO 2 - Critical Thinking & Problem Solving: Explore issues through various information sources; evaluate the credibility and significance of both the information and the source to arrive at a reasoned conclusion.	SLOs	ral Chei	» CH » CH	IEM151 S IEM151 S	LO2 -	Perform equilibr	ium calculations			
Mapped PSLOs       Chemistry Program Outcomes         Mapped PSLOs       CHEM PSLO - The student will demonstrate mastery of the approach and rationale of the scientific method and be able to apply these principles to solve problems.         » CHEM PSLO - The student will demonstrate mastery of stoichiometric calculations.         » CHEM PSLO - The student will demonstrate mastery of laboratory technique.         ILO         ILO 5 - Quantitative Literacy         » ILO 5 - Quantitative Literacy: Use mathematical concepts and models to analyze and sol real life issues or problems.         ILO 6 - Scientific Literacy:         » ILO 6 - Scientific Literacy:         » ILO 6 - Scientific Literacy:         » ILO 2 - Critical Thinking & Problem Solving         » ILO 2 - Critical Thinking & Problem Solving:         » ILO 2 - Critical Thinking & Problem Solving:         » ILO 2 - Critical Thinking & Problem Solving:         » ILO 2 - Critical Thinking & Problem Solving:         » ILO 2 - Critical Thinking & Problem Solving:         » ILO 2 - Critical Thinking & Problem Solving:         » ILO 2 - Critical Thinking & Problem Solving:         » ILO 2 - Critical Thinking & Problem Solving:         » ILO 2 - Critical Thinking & Problem Solving:         » ILO 2 - Critical Thinking & Problem Solving:         » ILO 2 - Critical Thinking & Problem Solving:         » ILO 2 - Critical Thinking & Problem Solving:     <	SLOs	ral Chei	» CH » CH » CH	IEM151 S IEM151 S IEM151 S	LO2 - LO3 -	Perform equilibr Perform thermo	ium calculations dynamic calcula	tions.		
Mapped PSLOs       Chemistry Program Outcomes         Wapped PSLOs       > CHEM PSLO - The student will demonstrate mastery of the approach and rationale of the scientific method and be able to apply these principles to solve problems.         >> CHEM PSLO - The student will demonstrate mastery of stoichiometric calculations.         >> CHEM PSLO - The student will demonstrate mastery of laboratory technique.         ILO         ILO 5 - Quantitative Literacy         >> ILO 5 - Quantitative Literacy: Use mathematical concepts and models to analyze and sol real life issues or problems.         ILO 6 - Scientific Literacy         >> ILO 6 - Scientific Literacy:         >> ILO 6 - Scientific Literacy:         >> ILO 2 - Critical Thinking & Problem Solving         >>> ILO 2 - Critical Thinking & Problem Solving         >>>>>>>>>>>>>>>>>>>>>>>>>>>>>>	SLOs	al Chei	» CH » CH » CH » CH » CH	IEM151 S IEM151 S IEM151 S IEM151 S	LO2 - LO3 - LO4 -	Perform equilibr Perform thermo Define and expl	ium calculations dynamic calcula ain concepts of e	tions. equilibria.	ın.	
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ILO 2 - Critical Thinking & Problem Solving         » ILO 2 - Critical Thinking & Problem Solving: Explore issues through various information sources; evaluate the credibility and significance of both the information and the source to arrive at a reasoned conclusion.         Assessments         Fall 2017	SLOs CSLOs	ral Chei	» CH » CH » CH » CH » CH Cher » CH scier » CH scier » CH scier » CH scier » CH	EM151 S EM151 S EM151 S EM151 S EM151 S EM151 S IEM151 S IEM151 S IEM PSLC EM PSLC EM PSLC EM PSLC	LO2 - LO3 - LO3 - LO3 - LO5 - LO6 - Dogram (D) - The od and (D) - The tative hittative or product	Perform equilibr Perform thermo Define and expl Interpret a pH g Perform qualitat <b>n Outcomes</b> e student will der d be able to appl e student will der e student will der Literacy e Literacy: Use r oblems.	ium calculations dynamic calculat ain concepts of e raph from an aci ive analysis. nonstrate maste y these principle nonstrate maste nonstrate maste	tions. equilibria. d-base titratio ry of the appr s to solve pro ry of stoichior ry of laborato	oach an blems. netric ca ry techni	lculations. ique.
<ul> <li>» ILO 2 - Critical Thinking &amp; Problem Solving: Explore issues through various information sources; evaluate the credibility and significance of both the information and the source to arrive at a reasoned conclusion.</li> <li>Assessments         Fall 2017     </li> </ul>	SLOs CSLOs Mapped PSLOs	ral Chei	» CH » CH » CH » CH » CH Cher » CH scier » CH scier » CH ILO { v ILO real I ILO { v ILO v ILO v ILO v ILO	EM151 S EM151 S EM151 S EM151 S EM151 S EM151 S IEM151 S DISTY Pro- DISTY Pro	LO2 - LO3 - LO4 - LO5 - LO5 - LO6 - Cogram () - The od and O - The tative () - - - - - - - - - - - - - - - - - - -	Perform equilibr Perform thermo Define and expl Interpret a pH g Perform qualitat <b>n Outcomes</b> e student will der d be able to appl e student will der e student will der Literacy e Literacy: Use r oblems. eracy iteracy: Use scie	ium calculations dynamic calcula ain concepts of e raph from an aci ive analysis. nonstrate maste y these principle nonstrate maste nonstrate maste	tions. equilibria. d-base titratio ry of the appr s to solve pro ry of stoichior ry of laborator	oach an blems. netric ca ry techni odels to	alculations. ique. analyze and sol
Assessments Fall 2017	SLOs CSLOs Mapped PSLOs	ral Chei	» CH » CH » CH » CH » CH Cher » CH scier » CH scier » CH ILO { v ILC real I ILO { v ILC solut	EM151 S EM151 S EM151 S EM151 S EM151 S EM151 S IEM151 S DISTY Pro- DISTY Pro	LO2 - LO3 - LO3 - LO4 - LO5 - LO5 - LO6 - Cogram () - The od and () - The od and () - The tative htitative or pro- fic Lite thif c Lite	Perform equilibr Perform thermo Define and expl Interpret a pH g Perform qualitat <b>n Outcomes</b> e student will der d be able to appl e student will der e student will der Literacy e Literacy: Use r oblems. eracy iteracy: Use scie challenges.	ium calculations dynamic calcula ain concepts of e raph from an aci ive analysis. nonstrate maste y these principle nonstrate maste nonstrate maste	tions. equilibria. d-base titratio ry of the appr s to solve pro ry of stoichior ry of laborator	oach an blems. netric ca ry techni odels to	alculations. ique. analyze and sol
Fall 2017	CSLOS CSLOS Mapped PSLOS	ral Chei	» CH » CH » CH » CH » CH Cher » CH Scier » CH ILO { » ILC real I ILO { » ILC solut ILO 2 » ILC source	IEM151 S IEM151 S IEM151 S IEM151 S IEM151 S IEM151 S IEM151 S IEM151 S IEM751 S IEM PSLC IEM PSLC IEM PSLC IEM PSLC 5 - Quantif IEM PSLC 5 - Quantif 5 - Quantif 5 - Quantif 5 - Quantif 0 5 - Quantif 0 5 - Quantif 0 6 - Scienti 0 7 - Critical 0 2 - Critical	LO2 - LO3 - LO4 - LO5 - LO6 - <b>ogram</b> () - The od and () - The tative   tative   tat	Perform equilibr Perform thermo Define and expl Interpret a pH g Perform qualitat <b>n Outcomes</b> e student will der d be able to appl e student will der e student will der titeracy e Literacy: Use r bblems. eracy iteracy: Use scie challenges. sing & Problem S nking & Problem e credibility and	ium calculations dynamic calcular ain concepts of e raph from an aci ive analysis. nonstrate maste y these principle nonstrate maste nonstrate maste nonstrate maste nonstrate maste nonstrate constrate maste nonstrate maste	tions. equilibria. d-base titratio ry of the appr s to solve pro ry of stoichior ry of laborator ncepts and me and methodo	oach an blems. netric ca ry techni odels to ologies to plogies to	analyze and sol o assess potent us information
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	SLOS CSLOS Mapped PSLOs Mapped ILOs		» CH » CH » CH » CH » CH Cher » CH Scier » CH ILO { » ILC real I ILO { » ILC solut ILO 2 » ILC source	IEM151 S IEM151 S IEM151 S IEM151 S IEM151 S IEM151 S IEM151 S IEM151 S IEM751 S IEM PSLC IEM PSLC IEM PSLC IEM PSLC 5 - Quantif IEM PSLC 5 - Quantif 5 - Quantif 5 - Quantif 5 - Quantif 0 5 - Quantif 0 5 - Quantif 0 6 - Scienti 0 7 - Critical 0 2 - Critical	LO2 - LO3 - LO4 - LO5 - LO6 - <b>ogram</b> () - The od and () - The tative   tative   tat	Perform equilibr Perform thermo Define and expl Interpret a pH g Perform qualitat <b>n Outcomes</b> e student will der d be able to appl e student will der e student will der titeracy e Literacy: Use r bblems. eracy iteracy: Use scie challenges. sing & Problem S nking & Problem e credibility and	ium calculations dynamic calcular ain concepts of e raph from an aci ive analysis. nonstrate maste y these principle nonstrate maste nonstrate maste nonstrate maste nonstrate maste nonstrate constrate maste nonstrate maste	tions. equilibria. d-base titratio ry of the appr s to solve pro ry of stoichior ry of laborator ncepts and me and methodo	oach an blems. netric ca ry techni odels to ologies to plogies to	analyze and so o assess potent us information
	SLOS CSLOS Mapped PSLOs Mapped ILOs	ral Cher	» CH » CH » CH » CH P CH Cher Cher » CH scier » CH ILO { v ILC real I ILO { v ILC solut ILO 2 v ILC source	IEM151 S IEM151 S IEM151 S IEM151 S IEM151 S IEM151 S IEM151 S IEM151 S IEM751 S IEM PSLC IEM PSLC IEM PSLC IEM PSLC 5 - Quantif IEM PSLC 5 - Quantif 5 - Quantif 5 - Quantif 5 - Quantif 0 5 - Quantif 0 5 - Quantif 0 6 - Scienti 0 7 - Critical 0 2 - Critical	LO2 - LO3 - LO4 - LO5 - LO6 - <b>ogram</b> () - The od and () - The tative   tative   tat	Perform equilibr Perform thermo Define and expl Interpret a pH g Perform qualitat <b>n Outcomes</b> e student will der d be able to appl e student will der e student will der titeracy e Literacy: Use r bblems. eracy iteracy: Use scie challenges. sing & Problem S nking & Problem e credibility and	ium calculations dynamic calcular ain concepts of e raph from an aci ive analysis. nonstrate maste y these principle nonstrate maste nonstrate maste nonstrate maste nonstrate maste nonstrate constrate maste nonstrate maste	tions. equilibria. d-base titratio ry of the appr s to solve pro ry of stoichior ry of laborator ncepts and me and methodo	oach an blems. netric ca ry techni odels to ologies to plogies to	analyze and so analyze and so o assess potent us information

	Score	Institutio	Institutional	Institutional Below Standards	N/A		
SLO CHEM151 SLO2 - Perform	24 of 2	Standa 24 6	ards 4	14	0	-	
equilibrium calculations. Spring 2018							
SLO3 - Perform thermodyn	amic calcul	ations					
SLO	Score	Institutio	eds Institutional Meets Standards	Institutional Below Standards	N/A		
CHEM151 SLO3 - Perform thermodynamic calculations.	28 of 9			6	2		
SLO3 - thermodynamic cal	culations						
SLO	Score	d Institution d Excee Standa	eds Meets Standards	Institutional Below Standards	N/A		
CHEM151 SLO3 - Perform thermodynamic calculations.	28 of 9	92 12	8	7	1		
SLO3 - thermodynamic cal	culations	-					
SLO	Score	d Excee Standa	eds Meets Standards	Institutional Below Standards	N/A		
CHEM151 SLO3 - Perform thermodynamic calculations.	29 of 9	92 8	11	9	1		
Action Plans Fall 2017 2017 Course Improvement Pla	n Action			1			Resource
Expected Action	Туре	Respondent		tion Taken		Date	Request
Allan Hancock College >> Che	mistry >> CH	EM151 - Fall 2017	7				
Spring 2018							
2017 Context Improvement Pla	Action	Deenendent		tion Tokon		Dete	Resource
Expected Action	Туре	Respondent		tion Taken		Date	Request
Allan Hancock College >> Che 2017 Course Improvement Pla		EM151 - Spring 2	2018				
Expected Action	Action	Respondent	A.	tion Taken		Date	Resource
-	Туре	-				Date	Request
Allan Hancock College >> Che			.010				
CHEM180 - Organ	lic Chei	mistry i					
SLOs CSLOs		on molecular s » CHEM180 S able to draw th » CHEM180 S organic reactio » CHEM180 S polarimetry, ar » CHEM180 S and distillation	LO2 - Define structure the condensed and line LO3 - Determine reac ons to be carried out in LO4 - Identify compound IR, MS, and NMR s LO5 - Synthesize and through macro- and n	es of alcohols, al -bond formulas ( tion mechanisms the laboratory. unds through the pectroscopy. purify compoun	kyl halides, ar Kekulé structi s and propose use of chrom ds utilizing cry	nd hydro ures). synthe atograp	ocarbons and be sis routes for hy, refractometry,
Mapped PSLOs		Chemistry Pro » CHEM PSLC scientific methe » CHEM PSLC	ogram Outcomes ogram Outcomes O - The student will dea od and be able to app O - The student will dea	ly these principle	es to solve pro	blems.	
Mapped ILOs	<ul> <li>» CHEM PSLO - The student will demonstrate mastery of laboratory technique.</li> <li>ILO</li> <li>ILO 6 - Scientific Literacy</li> <li>» ILO 6 - Scientific Literacy: Use scientific knowledge and methodologies to assess potent solutions to real-life challenges.</li> <li>ILO 2 - Critical Thinking &amp; Problem Solving</li> <li>» ILO 2 - Critical Thinking &amp; Problem Solving: Explore issues through various information sources; evaluate the credibility and significance of both the information and the source to arrive at a reasoned conclusion.</li> </ul>						ous information
Action Plans							
Fall 2017							
2017 Course Improvement Pla	n Action						Resource
Expected Action	Action Type	Respondent	Ac	tion Taken		Date	Request
Allan Hancock College >> Che							
	mistry >> CH	EM180 - Fall 2017	7				
Spring 2018	mistry >> CH	EM180 - Fall 2017	7				
Spring 2018 2017 Context Improvement Pla		EM180 - Fall 2017	7				Resource

26

Allan Hancock College >> Che	emistry >> CH	IEM180 - Spring 2	018						
2017 Course Improvement Pla	-								
Expected Action	Action Type	Respondent	Action Taken	Date	Resource Request				
Allan Hancock College >> Che	emistry >> CH	EM180 - Spring 2	018						
CHEM181 - Orgai	nic Che	mistry II							
SLOs		-							
			LO1 - Make predictions on physical properties and	chemica	I reactivity base				
			on molecular structure.						
			LO2 - Define structures of aldehydes, amides, amin		•				
			tones and be able to draw the condensed and line-b	ona tori	mulas (Kekule				
		structures).	LO3 - Determine reaction mechanisms and propose	synthe	sis routes for				
CSLOs			ons to be carried out in the laboratory.	2 Syntho	313 100103 101				
00203		U	LO4 - Identify compounds through the use of chrom	atograr	hv IR NMR a				
		UV-spectrosco		5	··· <b>,</b> , · ···, · ····, ···				
			LO5 - Synthesize and purify compounds utilizing cr	ystalliza	tion, sublimatior				
			through macro- and micro-scale procedures.						
		» CHEM181 S	» CHEM181 SLO6 - Relate functional groups to carbohydrate, lipid, nucleic acid, and prote						
		classification a							
		Chemistry Program Outcomes							
		Chemistry Program Outcomes							
Mapped PSLOs		» CHEM PSLO - The student will demonstrate mastery of the approach and rationale of the							
			od and be able to apply these principles to solve pro						
			) - The student will demonstrate mastery of laborate	ory techr	nique.				
		ILO							
		ILO 6 - Scientific Literacy							
		» ILO 6 - Scientific Literacy: Use scientific knowledge and methodologies to assess potent							
Mapped ILOs		solutions to real-life challenges.							
		ILO 2 - Critical Thinking & Problem Solving							
		» ILO 2 - Critical Thinking & Problem Solving: Explore issues through various information sources; evaluate the credibility and significance of both the information and the source to							
		arrive at a reasoned conclusion.							
Action Plans									
Fall 2017									
2017 Course Improvement Pla									
Expected Action	Action Type	Respondent	Action Taken	Date	Resource Request				
Allan Hancock College >> Che	emistry >> CH	EM181 - Fall 201	7						
Spring 2018									
2017 Context Improvement PI					-				
Expected Action	Action Type	Respondent	Action Taken	Date	Resource Request				
Allan Hancock College >> Che	emistry >> CH	IEM181 - Spring 2	018						
2017 Course Improvement Pla									
Expected Action	Action Type	Respondent	Action Taken	Date	Resource Request				
Allan Hancock College >> Che	emistry >> CH	IEM181 - Spring 2	018						

Chemistry Program Review Review of Prerequisites, Corequisites, and Advisories – Summary

#### UC/CSU COMPARISON SHEET

Course Prefix and N	Jumber <u>CHEM 120 (Introductory Chemistry)</u>	
Department <u>Life ar</u>	nd Phyiscal Sciences Responsible Instructor <u>Dustin Nouri</u>	
Prerequisite being reviewed: <u>MATH309</u> or <u>MATH311 (Algebra I)</u> Use one form for each prerequisite/corequisite/advisory if the course has more than one		
The following UC or CSU campus offers the same course and is identified as:		
Institution	Name of Parallel Course Prerequisite of Parallel Course	
CSU Sacramento	CHEM 6A Introduction to Gen. Chemistry one year of high school algebra	
CSU Fullerton	CHEM 100 Survey of Chemistry one year of high school algebra	
CSU LA	CHEM 151 Fundamentals of Chemistry one year of high school algebra	

The prerequisites at the above institutions are the same courses or the same experience (if a sequence is the stated prerequisite) as the Allan Hancock College prerequisite, and it is the recommendation of the faculty that the stated prerequisite be maintained.

Dustin Nour Date Initiator

Ashley Wise 12021 Date **Department Chair** 

Approved: Academic Dear

Completed forms and all backup documentation should be maintained at the department. Transfer conclusion information to the Program Evaluation PCA Summary Report.

~

#### UC/CSU COMPARISON SHEET

Course Prefix and Nur	nber <u>CHEM 140 (Introductory Organic and</u>	<u>Biological Chemistry)</u>	
Department Life and	Phyiscal Sciences Responsible Instruc	tor <u>Dustin Nouri</u>	
Prerequisite being revi	ewed: <u>CHEM 120 (Introductory Chemistr</u> Use one form for each prerequisite/corequisite/advisor		
The following UC or C	SU campus offers the same course and is identifie	d as:	
Institution	Name of Parallel Course	Prerequisite of Parallel Course	
CSU San Diego	CHEM 130 Elementary Organic Chemistry	Chemistry 100 or 200	
CSU Fresno	CHEM 3B Introductory Organic and Biochemistry CHEM 3A		
CSU San Bernadino	CHEM 206 Fundamentals of Chemistry II	CHEM 205 or CHEM 215	

The prerequisites at the above institutions are the same courses or the same experience (if a sequence is the stated prerequisite) as the Allan Hancock College prerequisite, and it is the recommendation of the faculty that the stated prerequisite be maintained.

Dustin Nouri 202Date Initiator

21/2022 202Ĭ Ashley Wise Date **Department Chair** 

Approved: Academic Dean

#### UC/CSU COMPARISON SHEET

Course Prefix and Number <u>C</u>	CHEM 150 (General Chemistry 1)	
Department_Life and Phyisc	al Sciences Responsible Ins	structor <u>Dustin Nouri</u>
Prerequisite being reviewed: <u>CHEM 120 (Introductory Chemistry</u> Use one form for each prerequisite/corequisite/advisory if the course has more than one		
The following UC or CSU cam	pus offers the same course and is ide	ntified as:
Institution	Name of Parallel Course	Prerequisite of Parallel Course
CSU Domingues Hills	CHE 110 General Chemistry I	CHE 108 Introduction to College Chemistry
CSU Bakersfield CHEM	211 Principles of General Chemistry	CHEM 101 Preparation for College Chemistry
CSU Channel Islands	CHEM 121 General Chemistry I	CHEM 105 Introduction to Chemistry

The prerequisites at the above institutions are the same courses or the same experience (if a sequence is the stated prerequisite) as the Allan Hancock College prerequisite, and it is the recommendation of the faculty that the stated prerequisite be maintained.

9/19/2022 Dustin Nouri Initiator

21/2022 2021 Ashley Wise Department Chair Date

Approved Academic Dean Date

#### UC/CSU COMPARISON SHEET

Course Prefix and Nur	mber <u>CHEM 150 (General Chemistry 1)</u>	
Department <u>Life and</u>	Phyiscal Sciences Responsible Ins	tructor_Dustin Nouri
Prerequisite being rev	iewed: <u>AND MATH331 (Algebra 2)</u> Use one form for each prerequisite/corequisite/ad	visory if the course has more than one
The following UC or C	SU campus offers the same course and is ide	ntified as:
Institution	Name of Parallel Course	Prerequisite of Parallel Course
CSU Sacramento	CHEM 1A General Chemistry	High school algebra (two years)
CSU Bakersfield	CHEM 211 Principles of General Chemistry	Math 85 Developmental Mathematics II
CSU LA	CHEM 101 General Chemistry I	two years of high school algebra

The prerequisites at the above institutions are the same courses or the same experience (if a sequence is the stated prerequisite) as the Allan Hancock College prerequisite, and it is the recommendation of the faculty that the stated prerequisite be maintained.

2022 Dustin Nouri 2021 Date Initiator

. 101 2021 9/21/2022 Ashley Wise Date Department Chair

aho, Approved 202 Academic Dean Date

#### UC/CSU COMPARISON SHEET

Course Prefix and Number CHEM 151 General Chemistry 2)			
Department_Life and Phyiscal Sciences Responsible Instructor_Dustin Nouri			
Prerequisite being reviewed: <u>CHEM 150 (General Chemistry 1)</u> Use one form for each prerequisite/corequisite/advisory if the course has more than one			
The following UC or CSU campus offers the same course and is identified as:			
Institution	Name of Parallel Course	Prerequisite of Parallel Course	
UC Berkley	Chemistry 1B General Chemistry	Chemistry 1A/1AL General Chemistry	
CSU Channel Islands	Chem 122 General Chemistry	Chem 121 General Chemistry	
CSU Sacramento	Chem 1B General Chemistry II	Chem 1A General Chemistry 1	

The prerequisites at the above institutions are the same courses or the same experience (if a sequence is the stated prerequisite) as the Allan Hancock College prerequisite, and it is the recommendation of the faculty that the stated prerequisite be maintained.

9/19/2022 Dustin Nouri <del>-2021</del> Initiator

. <u>22021</u> 9/21/2022 Ashley Wise Department Chair Date

9/20/202 Approved; Date Academic Dean

#### UC/CSU COMPARISON SHEET

Course Prefix and Number CHEM 180 (Organic Chemistry 1)			
Department_Life and Phyiscal Sciences Responsible Instructor_Dustin Nouri			
Prerequisite being reviewed: <u>CHEM 151 (General Chemistry 2)</u> Use one form for each prerequisite/corequisite/advisory if the course has more than one			
The following UC or CSU campus offers the same course and is identified as:			
Institution	Name of Parallel Cour	se Prerequ	isite of Parallel Course
CSU Fresno	Chemistry 128A Organ	ic Chemistry	Chemistry 1B
CSU Channel Islands	Chem 311/312 Organic C	Chemistry I&IL	Chem 122 General Chemistry
CSU Sacramento	Chem 24/25 Organic C	hemistry I & I L	Chem 1B

The prerequisites at the above institutions are the same courses or the same experience (if a sequence is the stated prerequisite) as the Allan Hancock College prerequisite, and it is the recommendation of the faculty that the stated prerequisite be maintained.

Dustin Nouri Date Initiator

2022 2 2021 9 Ashley Wise Department Chair Date

Approved: Academic Dea

#### UC/CSU COMPARISON SHEET

Course Prefix and Number _	CHEM 181 (Organic Chemistry 2)	
Department Life and Phyis	scal Sciences Responsible Ins	tructor <u>Dustin Nouri</u>
	CHEM 180 (Organic Chemistry 1) Use one form for each prerequisite/corequisite/ad	
The following UC or CSU car	mpus offers the same course and is ide	ntified as:
Institution	Name of Parallel Course	Prerequisite of Parallel Course
CSU Fresno	Chemistry 128B Organic Chemistry	Chemistry 128A
CSU Channel Islands	Chem 314/315 Organic Chemistry II &	II L Chem 311/312 Organic Chemistry I & I L
CSU Sacramento	Chem 124/125 Organic Chemistry II	& II L Chem 24/25 Organic Chemistry I & I L

The prerequisites at the above institutions are the same courses or the same experience (if a sequence is the stated prerequisite) as the Allan Hancock College prerequisite, and it is the recommendation of the faculty that the stated prerequisite be maintained.

19/2022 Dustin Nouri Initiator Date

. 2021 9/21/2022 Ashley Wise Department Chair Date

Approved Academic Dean Date

# Chemistry Program Review Plan of Action – Pre-Validation

#### **PLAN OF ACTION - PRE-VALIDATION** Six Year

#### DEPARTMENT: Life and Physical Sciences PROGRAM: CHEMISTRY

List below as specifically as possible the actions which the department plans to take as a result of this program review. Be sure to address any problem areas which you have discovered in your analysis of the program. Number each element of your plans separately and for each, please include a target date. Additionally, indicate by the number each institutional goal and objective which is addressed by each action plan. (See Institutional Goals and Objectives)

#### RECOMMENDATIONS TO IMPROVE STUDENT LEARNING OUTCOMES AND ACHIEVMENT

1)	The chemistry faculty will continue to support the MESA and STEM programs as well as the Learning Resource Center at the Santa Maria and Lompoc Valley Center campuses to help inform students of their support services. The faculty will continue to promote high academic standards and success in achieving Program Learning Outcomes so that our students may thrive once they have moved on to the next stage of their academic careers.
2)	The chemistry faculty will continue to remain current in the latest chemistry literature and useful technology to help convey the material at the appropriate undergraduate level.
3)	The full-time chemistry faculty will work with the Union and current contracts to

he	lp ensure the adju	inct-faculty ar	e properl	y compensated	d for the a	dditional
wo	ork associated with	n inputting Pro	ogram Lea	arning Outcom	e data.	

Theme/Objective/ Strategy Number AHC from Strategic Plan	
1) A.1/A.7/B.7/C.7.	
2) B.7.	
3) B.7.	

TARGET DATE 1) ONGOING

2) ONGOING 3) ONGOING

RECOMMENDATIONS TO ACCOMMODATE CHANGES IN <b>STUDENT</b> CHARACTERISTICS	Theme/Objective/ Strategy Number AHC from Strategic Plan	TARGET DATE
	A.2/B.4/D.5	ONGOING
The chemistry curriculum sections are growing and expanding. Additional lab space is		
being required at both campuses over the next program cycle. LVC has begun looking into		
converting LVC3-109 into a science lab. As they currently only have one functioning		
chemistry lab, this will help free up LVC3-102 for evening course offerings.		
The SM campus will likely look into M-212 since that lab has fume-hoods. Lab benches		
and gas lines will need to be run to make the room fully functional.		
Demographic Changes	A.3/D.5/D.7	ONGOING
The chemistry program will continue to consider accommodations for student whom cannot		
attend day time classes. We need to expand LVC offerings and evening sections.		
Outfitting lecture and lab rooms with Zoom equipment can help during challenging times.		

RECOMMENDATIONS TO IMPROVE THE EDUCATIONAL ENVIRONMENT	Theme/Objective/ Strategy Number AHC from Strategic Plan	TARGET DATE
<b>Curricular Changes</b> The chemistry faculty are updating curriculum to better serve the needs of the students. Our CHEM140 course will be mapped to the C-ID CHM102 to help ease any transfer issues. Sections of this course may be expanded to the sister campus, LVC, as need grows.	B.8	ONGOING
<b>Co-Curricular Changes</b> A math review CANVAS course may need to be created to help our incoming students meet the Basic Math Skills they require to succeed.	B.8	ONGOING

Neighboring College and University Plans The chemistry faculty will continue to work with neighboring colleges and universities to ensure that courses articulate and topics are aligned.	C.3/C.8/D.6/E.3	ONGOING
<b>Related Community Plans</b> The chemistry faculty will continue to volunteer when asked as we have for science fairs, Friday Night Science, tours of our department, brining hand-on chemistry to other schools, and presenting professional development activities.	A.1/A.5/A.6/E.7/E. 8	ONGOING

RECOMMENDATIONS THAT REQUIRE RESOURCES	Theme/Objective/ Strategy Number AHC from Strategic Plan	TARGET DATE	
Facilities		A.1/A.4/B.1/B.2/	1)ONGOING
<ol> <li>Service the fume-hoods annually as Cal C</li> <li>LVC3-102, 3-114, 2-212, and 2-102 need</li> <li>Need new whiteboards for M205/M213</li> </ol>	smart podium upgrades.	B.3/E.1/E.2	2)FALL2023
<ul> <li>4) Need new projector screens for M205/M-</li> <li>5) LVC 3-102 requires new ballasts for light</li> <li>6) Expansion into M212? (\$235,000)</li> </ul>			3)FALL2025
			4)SPRING2026
			5) SPRING2023
			6) SPRING2027
Equipment	LVC Classes (\$10,000)	A.1/A.4/B.1/B.2/	1) ONGOING
<ol> <li>SM Gloves (\$10,000)</li> <li>SM Equipment under \$500 (\$4,000) + inflation (\$1200)</li> </ol>	LVC Gloves (\$10,000) LVC Equipment under \$500 (\$3,000) + inflation (\$1200)	B.3/D.6/D.7/E.1/ E.2	2) ONGOING
<ul> <li>3) SM Goggles (\$12,000)</li> <li>4) SM Analytical Balance (\$3,500)</li> <li>5) SM 16 Centrifuges (\$38,400)</li> </ul>	LVC Goggles (\$12,000) LVC 7 Analytical Balances (\$42,000)		3) FALL2023
<ul> <li>6) ChemDraw Software (\$4,250)</li> <li>7) SM Repairs (\$500)</li> </ul>	LVC Repairs (\$500)		4) FALL2023
<ul> <li>8) LVC Water Bath (\$900)</li> </ul>			5) FALL2023-
<ul><li>9) LVC Fume-hoods and gas lines for new c</li><li>10) SM gas lines and lab benches/stools for new</li></ul>	ew chemistry lab (M-212)		SPRING2024
<ul> <li>11) LVC MelTemp equipment for CHEM140</li> <li>12) Student laptops at SM and LVC campuses</li> <li>13) LVC 16 Stirrers (\$5,400)</li> </ul>			6) FALL2025
15) LVC 10 Suiters (\$5,400)			7) ONGOING
			8) FALL2022
			9)FALL2024
			10)FALL2026
			11)FALL2027
			12) ONGOING
			13) FALL2023

	fing	A.1/A.4/B.1/B.2/	1) FALL2023 –
1)	Need full-time chemists to help fill the demand and need of the current sections offered and expanding.	B.3/E.1/E.2	FALL2025
2) 3)	SM Chemistry and Biology Lab Associate Position (+\$55,000) LVC Chemistry and Biology Lab Associate Position (+\$55,000)		2)SPRING2023
5)	Eve chemistry and Biology Lab Associate Position (+\$55,000)		2)37 KING2023
			3)FALL2026

# <u>EXHIBITS</u> Student Data Summary Student Data Statistics Articulation Status of Courses Course Review Verification Sheets

#### STUDENT DATA SUMMARY

Data analysis is a critical component of program review. The three categories below should be used as guidelines in developing a summary of the student data.

State at least three positive factors about the discipline/program identified by students. Include the number (or percentage) of students responding and any implications for planning.

The vast majority of students surveyed were satisfied (most of those "highly satisfied") with the following:

- 77% somewhat or highly satisfied with the "quality of instruction within the program" (Q2.1)
- 72% somewhat or highly satisfied with the "contribution towards [their] intellectual goal" (Q2.5)
- 71% somewhat or highly satisfied with the "physical facilities and space (e.g., classrooms, labs)" (Q2.11)
- 74% somewhat or highly satisfied with the "instructional equipment (e.g., computers, lab equipment)" (Q2.12)

In addition, of the students surveyed, 59% agree that "[they] would recommend taking courses in Chemistry" (Q6.1)

State at least three negative factors about the discipline/program identified by students. Include the number (or percentage) of students responding and any implications for planning.

- 1) Survey Question #2.3: 16% of students surveyed are dissatisfied with the "advice about the program from counselors." 40% are neither satisfied nor dissatisfied, with only 44% being satisfied. (Chemistry is a challenging course, and the workload is often underestimated by students and counselors. With several students skipping recommended/required prerequisites via appeals through counseling, students may end up feeling underprepared for the classes in which they enrolled. Better coordination between the program and counseling is needed to properly convey the workload of our courses to students. In addition, specialized counselors for the science majors may help.)
- 2) Survey Question #2.8: only 28% of students surveyed are satisfied with "the availability of courses offered in the Chemistry program." 38% are somewhat satisfied, with 25% being neither satisfied nor dissatisfied, and the remaining 10% being dissatisfied. (The demand from students for our courses has been increasing, while the availability of full-time faculty has not kept up. Future growth hires will likely help to staff more classes. However, the availability of lecture and lab space for additional classes is a more daunting hurdle to overcome. The possibility for remote options for some of our courses can also help to address our long waitlists.)
- 3) Survey Question #2.14: only 31% of students surveyed are satisfied with "course assistance through tutorial services (e.g through the Tutorial Center, Math Lab, Writing Center)." 38% are somewhat satisfied, with 23% being neither satisfied nor dissatisfied, and the remaining 8% being dissatisfied. (The new STEM center is well equipped to help chemistry students, with the

recent purchase of new molecular model kits and textbook hardcopies for students to use/borrow. In addition, embedded tutors have been implemented into some CHEM150 sections, with students giving positive feedback. The larger issue may be that students are generally unaware of the tutorial services made available to them, and more effort should be expended by instructors to make students aware of those resources.)

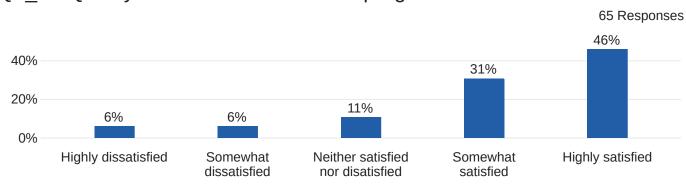
State any other information (use responsive numbers) that you obtained from student data (e.g. focus groups, questionnaires, or SGIDs) that may be of special interest to the self study team. What planning implications will result from this information?

It should be noted that our available student survey data is limited. It would have been extremely helpful to compare data from before, during, and after the COVID lockdown. This would help pinpoint issues with in-person v. online learning of chemistry within our program.

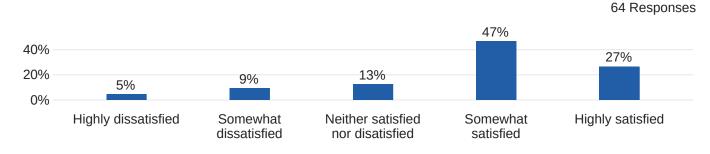
It should also be noted that 50% of the students surveyed were full-time students (12+ units enrolled)(Q11), and 39% of surveyed students plan on taking additional courses in Chemistry (Q6.2), which means that a significant number of the surveyed students may major in chemistry or a related subject. This makes the values gleaned from this survey even more poignant.

## Chemistry Fall 2021 Total Responses: 65

Please answer the following questions as they pertain to your experience in this course and all other courses in the Chemistry program at Allan Hancock College.

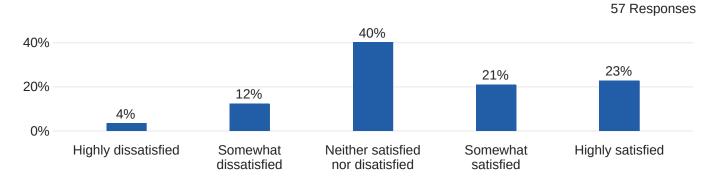


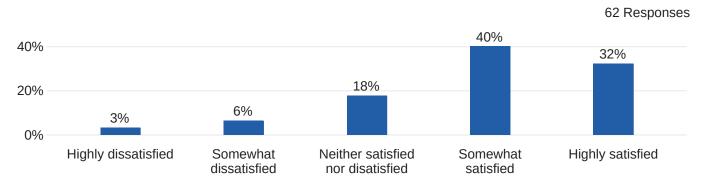
Q2\_2 - The way textbooks and other materials used in courses within the program help me learn



## Q2\_1 - Quality of instruction within the program

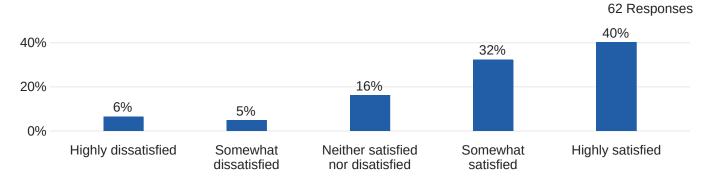
# Q2\_3 - Advice about the program from counselors



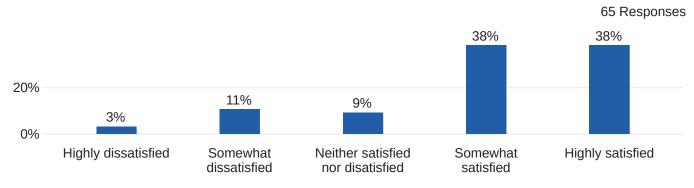


# Q2\_4 - The way this program meets your educational goals

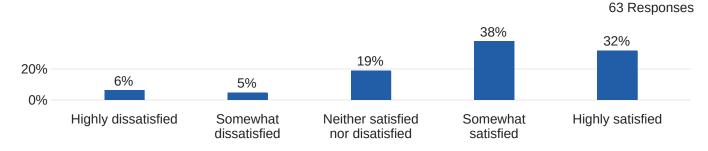
Q2\_5 - Contribution towards your intellectual growth



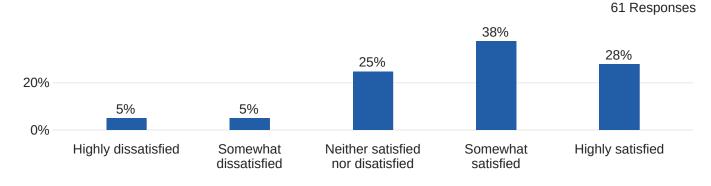
# Q2\_6 - Clarity of course goals and learning objectives



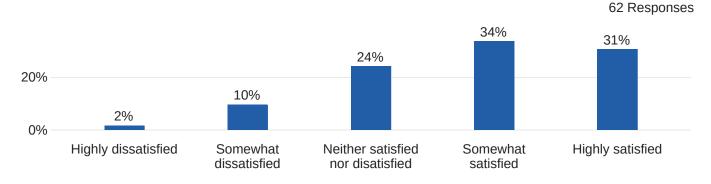
# Q2\_7 - Feedback and assessment of progress towards learning objectives



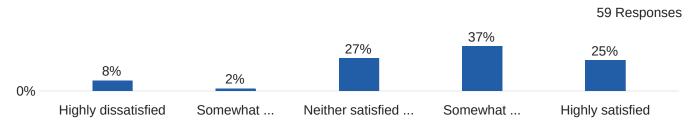
Q2\_8 - The availability of courses offered in the Chemistry program



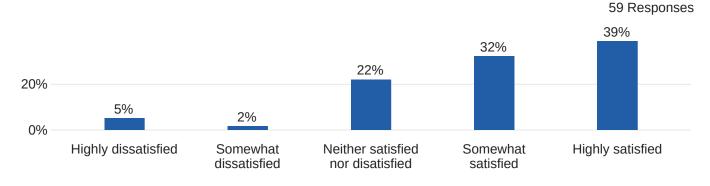
Q2\_9 - The content of courses offered in the Chemistry program



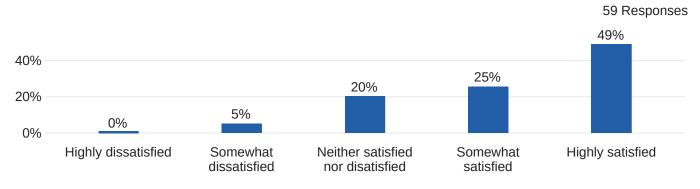
Q2\_10 - The coordination of courses offered in the Chemistry program and courses offered in other departments that may be required for your major



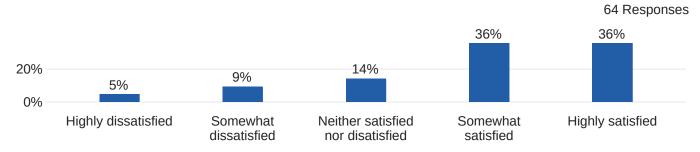
Q2\_11 - The physical facilities and space (e.g., classrooms, labs)



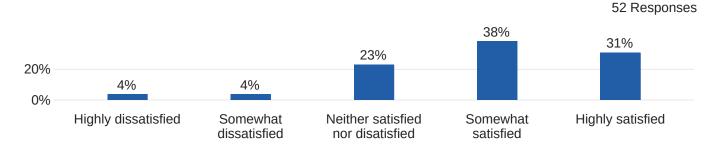
Q2\_12 - Instructional equipment (e.g., computers, lab equipment)



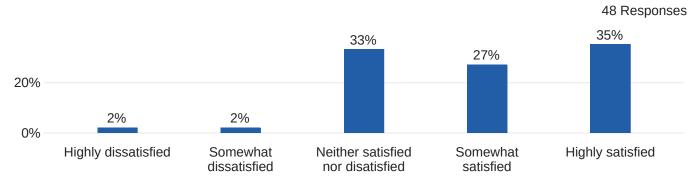
# Q2\_13 - Presentation of classes via the college's Canvas course management system



Q2\_14 - Course assistance through tutorial services (e.g through the Tutorial Center, Math Lab, Writing Center)

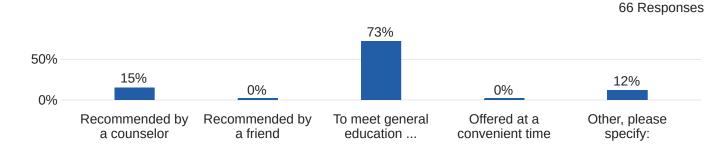


# Q2\_15 - Availability of appropriate resources in the libraries

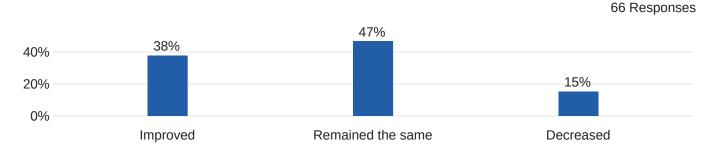


Part II. Please answer the following questions about the Computer Business Information Systems (CBIS) program.

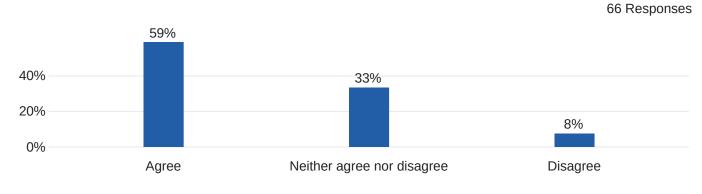
Q4 - Which of the following best describes your reason for taking this and other courses in Chemistry? - Selected Choice

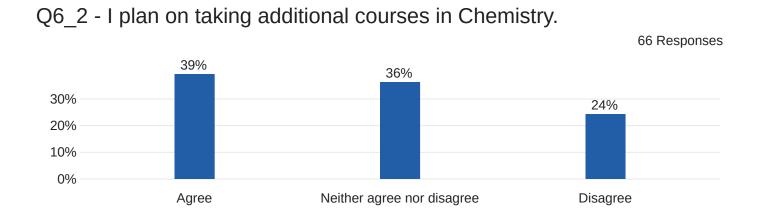


Q5 - Compared to the beginning of the semester, your attitude about Chemistry has...

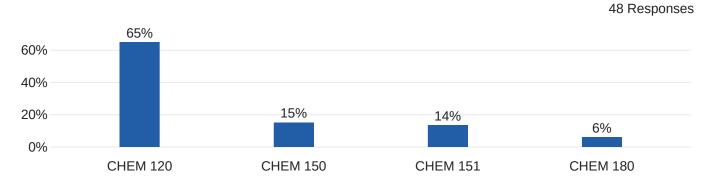


# Q6\_1 - I would recommend taking courses in Chemistry.

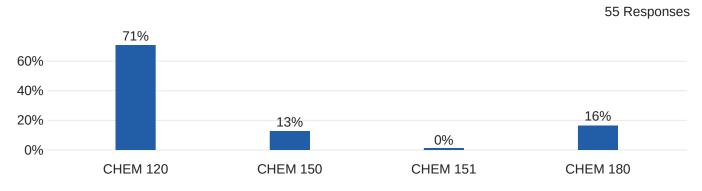




## Q7 - Which of the following courses have you taken in Chemistry?

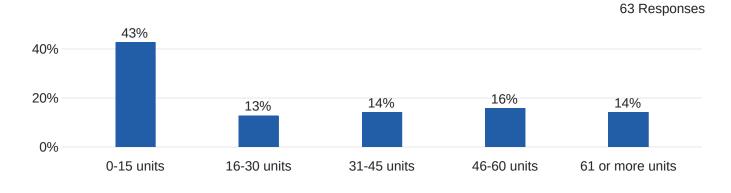


## Q8 - Which courses are you taking this semester in Chemistry?

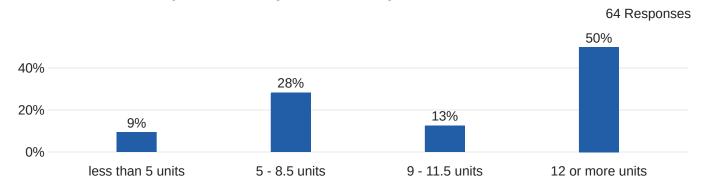


#### Part III. Background questions.

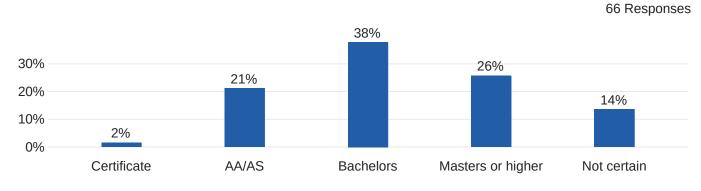
## Q10 - How many units have you completed prior to this semester?



Q11 - In how many units are you currently enrolled?



## Q12 - What is your final academic goal?



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# Program Data

# STEP 1 Choose subjects: CHEM

Subjects: CHEM

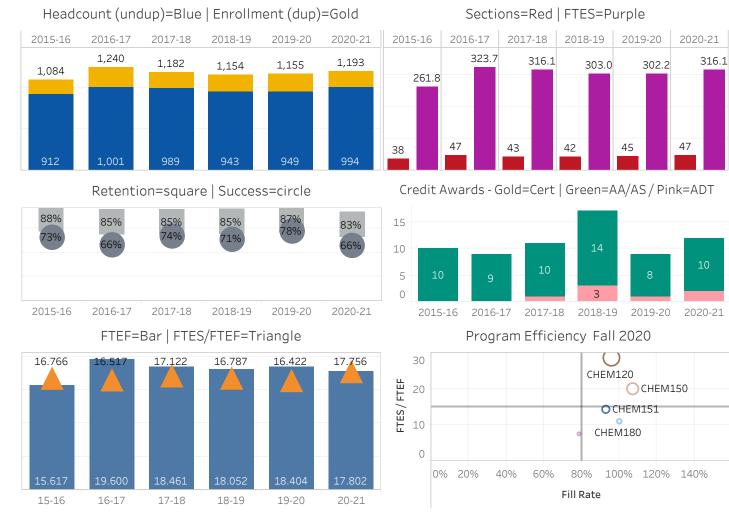
### STEP 2 Choose awards: Chemistry

Awards: Chemistry

## STEP 3 Choose majors: Chemistry



- Contents
- 1 Enrollment, headcount, sections, FTES, retention, success
- 2 Demographics
- **3 -** Equity outcomes
- 4 Online\Face to face comparison
- 5 Efficiency
- 6 Program awards & majors
- 7 Faculty load
- A Course demographic detail
- B Awards by major detail



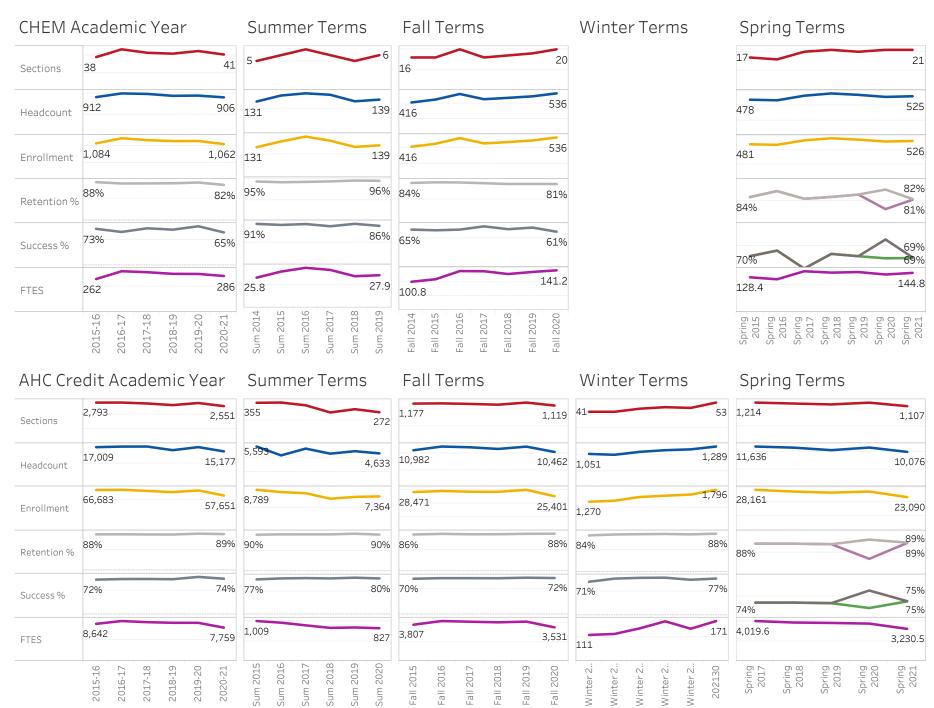
#### Quick Program Facts

Data Source: Student-MIS; Award, Major & Faculty-Banner | Headcount-unduplicated students; Enrollment-duplicated students; Retention-students who receive a grade in the course; Success-students who receive a passing grade in the course; FTES/FTEF target is 15+; Fill Rate target is 80%+

1 Outcon	1 Outcomes CHEM					course_ All							EW Grade EW						
	Sum 2014	Sum 2015	Fall 2015	Spring 2016	Sum 2016	Fall 2016	Spring 2017	Sum 2017	Fall 2017	Spring 2018	Sum 2018	Fall 2018	Spring 2019	Sum 2019	Fall 2019	Spring 2020	Fall 2020	Spring 2021	
Sections	5	6	16	16	7	20	20	6	16	21	5	17	20	6	18	21	20	21	
Headcount	131	156	455	470	165	528	532	159	459	563	132	477	544	139	497	516	536	525	
Enrollment	131	156	455	473	177	528	535	159	459	564	132	477	545	139	497	519	536	526	
retained	124	145	387	421	166	447	439	151	382	472	128	389	468	134	404	377	436	427	
Retention %	95%	93%	85%	89%	94%	85%	82%	95%	83%	84%	97%	82%	86%	96%	82%	90%	81%	82%	
success	119	138	291	357	160	344	318	136	325	409	120	314	384	120	340	356	329	362	
Success %	91%	88%	64%	75%	90%	65%	59%	86%	71%	73%	91%	66%	70%	86%	69%	85%	61%	69%	
FTES	25.8	31.1	109.7	121.1	34.3	138.6	150.8	32.5	138.0	145.7	27.1	128.2	147.8	27.9	135.6	138.7	141.2	144.8	

# Outcomes Allan Hancock College Credit

	Sum 2015	Fall 2015	Winter 2016	Spring 2016	Sum 2016	Fall 2016	Winter 2017	Spring 2017	Sum 2017	Fall 2017	Winter 2018	Spring 2018	Sum 2018	Fall 2018	Winter 2019	Spring 2019	Sum 2019	Fall 2019	Spring 2020	Sum 2020	Fall 2020	Spring 2021
Sections	355	1,177	41	1,220	357	1,184	41	1,214	333	1,168	45	1,186	270	1,145	47	1,159	299	1,208	1,212	272	1,119	1,107
Headco	5,593	10,982	1,051	11,341	4,354	12,111	1,023	11,636	5,306	11,889	1,118	11,320	4,596	11,380	1,171	10,580	4,940	12,091	11,342	4,633	10,462	10,076
Enrollm	8,789	28,471	1,270	28,153	8,305	29,268	1,314	28,161	8,052	28,754	1,480	26,960	6,868	28,650	1,535	26,193	7,252	30,166	26,977	7,364	25,401	23,090
Retentio n %	90%	86%	84%	89%	90%	88%	87%	88%	90%	87%	87%	88%	90%	87%	88%	88%	92%	88%	92%	90%	88%	89%
Success %	77%	70%	71%	73%	80%	71%	77%	74%	80%	71%	79%	74%	80%	71%	79%	74%	81%	72%	85%	80%	72%	75%
FTES	1,009	3,807	111	3,715	967	4,197	115	4,020	900	4,126	139	3,869	835	4,061	169	3,827	846	4,136	3,763	827	3,531	3,231



#### 1 Retention & Success by academic year by course CHEM

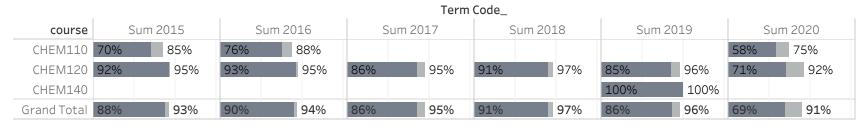
course_	2015-16	2016-17	2017-18	2018-19	2019-20	2020-21
CHEM110	71% 88%	63% 84%	48% 81%		45% 71%	57% 74%
CHEM120	70% 88%	64% 85%	73% 86%	68% 86%	78% 88%	68% 85%
CHEM140	77% 80%	83% 90%	85% 89%	72% 72%	100% 100%	78% 86%
CHEM150	69% 85%	60% 79%	71% 81%	75% 87%	74% 83%	65% 80%
CHEM151	91% 95%	88% 93%	86% 90%	77% 85%	87% 92%	53% 74%
CHEM180		65% 78%	53% 68%	47% 58%	62% 67%	37% 58%
CHEM181		81% 94%	100% 100%	86% 86%	100% 100%	100% 100%
Grand Total	73% 88%	66% 85%	74% 85%	71% 85%	78% 87%	65% 82%

Retention % and Success % for each course\_broken down by Academic Year. Color shows details about Retention % and Success %. The data is filtered on TERM\_CODE, CB04, subject and course. The TERM\_CODE filter excludes 201510, 201520, 201540, 202110 and 202130. The CB04 filter keeps C, D and N. The subject filter keeps CHEM. The course filter has multiple members selected.

#### Measure Names

Retention %

## 1 Retention & Success by summer term by course CHEM



#### Measure Names

Retention %

# 1 Retention & Success by fall term by course CHEM

course_	Fa	II 2015	Fa	all 2016		Fall 2017		Fall 2018		Fall 2019	Fall 2020		
CHEM110	70%	93%	58%	79%					35%	60%	41%	68%	
CHEM120	51%	81%	60%	85%	66%	81%	62%	82%	70%	85%	66%	86%	
CHEM150	90%	93%	74%	84%	79%	88%	77%	88%	71%	81%	63%	79%	
CHEM151	69%	83%	78%	96%	76%	83%	40%	52%	71%	77%	48%	75%	
CHEM180			65%	78%	53%	68%	47%	58%	62%	67%	37%	58%	
Grand Total	64%	85%	65%	85%	71%	83%	66%	82%	69%	82%	61%	81%	

#### Measure Names

Retention %

# 1 Retention & Success by spring term by course CHEM

course_	Spring 20	16	Spring 2	017	Spring	2018	Spring	2019	Spring 20	)20	Spring	2021
CHEM110	73% 8	35%	54%	83%	48%	81%			64%	91%	72%	80%
CHEM120	82%	94%	51%	31%	72%	86%	62%	84%	85%	88%	71%	84%
CHEM140	77% 80	C%	83%	90%	85%	89%	72% 7	2%	100%	100%	78%	86%
CHEM150	48% 77	7% ·	42% 73	3%	60% 7	1%	72%	87%	78%	36%	67%	80%
CHEM151	98%	99%	91%	92%	89%	92%	86%	93%	93%	97%	57%	74%
CHEM181			81%	94%	100%	100%	86%	86%	100%	100%	100%	100%
Grand Total	75%	89%	59%	82%	73%	84%	70%	86%	85%	90%	69%	82%

#### Measure Names

Retention %

#### 2 Program Demographics CHEM

Choose individual course via filter or see Appendix A for full demographic course details

						Academ	iic Year				
	2015-16	5	2016-17	,	2017-1	.8	2018-19	Э	2019-20	0	2020-21
Age Category	Headcount	FTES	Headcount	FTES	Headcount	FTES	Headcount	FTES	Headcount	FTES	Headcount
Under 20	267	72.6	341	108.1	393	114.9	410	126.1	403	132.7	387
20-24	457	136.8	471	157.8	418	144.3	385	126.1	377	114.5	358
25-29	115	32.4	118	36.5	109	34.3	108	33.4	97	27.7	83
30-34	35	7.3	35	9.8	36	11.0	34	11.4	47	14.8	55
35-39	22	5.2	25	6.1	17	4.5	13	3.5	19	6.0	24
40-49	22	4.8	16	3.8	20	5.1	8	1.9	19	5.0	13
50+	11	2.7	5	1.7	8	2.0	3	0.6	7	1.6	3
	2015-16	6	2016-17	,	2017-1	.8	2018-19	Ð	2019-20	0	2020-21
ETHNICITY	Headcount	FTES	Headcount	FTES	Headcount	FTES	Headcount	FTES	Headcount	FTES	Headcount
Asian	33	12.1	38	12.1	26	9.4	25	8.3	30	11.6	17
Black	16	3.8	18	5.6	19	6.4	16	5.3	16	5.0	21

Black	16	3.8	18	5.6	19	6.4	16	5.3	16	5.0	21	6.6
Filipino	32	10.3	42	13.2	43	14.3	46	16.4	33	11.7	35	12.8
Hispanic	527	151.8	584	191.2	567	182.4	489	153.2	456	142.4	422	132.5
NativeAm	14	4.4	19	6.0	13	4.0	20	5.9	19	6.6	15	4.7
Other					1	0.2	1	0.2				
PacIsl	4	1.4	6	2.0	9	2.6	4	1.6	4	1.3	6	2.4
White	286	78.1	293	91.1	312	96.9	340	111.6	384	121.4	382	119.8

	2015-1	6	2016-1	7	2017-1	18	2018-1	.9	2019-2	20	2020-2	21
	Headcount	FTES										
Female	493	131.7	533	162.9	570	172.8	551	169.6	581	176.4	543	170.2
Male	419	130.1	464	157.2	417	142.8	387	132.3	358	122.8	342	108.7
Unknown			3	1.1	2	0.5	3	0.6	3	0.7	13	4.3

	2015-1	6	2016-1	7	2017-1	8	2018-1	.9	2019-2	0	2020-2	1
	Headcount	FTES										
First Time	74	16.3	95	23.0	81	19.1	114	28.3	129	32.9	128	30.6
First Time Transfer	36	6.8	42	9.5	40	9.7	37	9.4	23	5.7	36	9.6
Continuing	773	227.4	856	277.6	841	274.5	809	256.1	810	250.1	756	234.7
Returning	47	10.0	43	9.5	30	7.8	23	5.6	36	8.8	20	5.6
Special Admit	5	1.1	8	1.7	22	4.9	14	3.1	11	2.3	10	2.8
Unknown	1	0.2										
Grand Total	912	261.8	1,000	321.3	989	316.1	941	302.5	942	299.9	898	283.3

FTES

116.8 114.3

25.9

16.2 7.0

4.4

1.3

FTES 4.5

course\_ All

# 2 Demographics Allan Hancock College Credit

	2015-1	.6	2016-1	7	2017-1	.8	2018-1	.9	2019-2	20	2020-2	1
Age Category	Headcount	FTES										
Under 20	4,528	2,759	5,805	3,105	6,308	3,155	6,018	3,326	7,482	3,583	6,828	3,029
20-24	6,054	3,341	5,700	3,398	5,460	3,190	5,057	3,070	4,867	2,853	4,251	2,441
25-29	2,555	1,118	2,440	1,255	2,395	1,212	2,071	1,101	2,060	1,089	1,831	986
30-34	1,533	528	1,379	578	1,327	556	1,173	560	1,130	507	1,109	550
35-39	969	292	924	357	891	328	758	319	844	342	706	296
40-49	1,262	356	1,042	379	1,040	384	801	328	874	324	732	306
50+	966	248	789	227	676	210	608	189	583	185	447	151
	2015-1	.6	2016-1	7	2017-1	.8	2018-1	.9	2019-2	20	2020-2	1
ETHNICITY	Headcount	FTES										
Asian	582	275	512	264	469	214	386	186	378	187	280	140
Black	673	359	583	326	555	278	459	259	491	278	437	232
Filipino	473	292	483	309	462	269	450	305	488	259	405	234
Hispanic	8,196	4,670	8,206	4,873	7,475	4,482	6,604	4,071	7,536	4,047	6,704	3,456
NativeAm	263	133	307	144	348	167	358	198	360	190	325	164
Other	2	0	4	1	5	2	2	1	2	1	2	1
PacIsl	97	50	119	62	141	62	131	74	167	81	128	62
White	6,728	2,862	7,016	3,146	7,819	3,541	7,236	3,751	7,129	3,648	6,533	3,319
	2015-1	.6	2016-1	7	2017-1	.8	2018-1	.9	2019-2	20	2020-2	1
	Headcount	FTES										
Female	8,360	4,479	8,768	4,922	8,937	4,913	8,454	4,877	8,777	4,837	8,274	4,467
Male	8,643	4,159	8,340	4,181	8,126	4,049	7,027	3,916	7,521	3,767	6,316	3,053
Unknown	3	2	109	23	181	51	121	52	228	88	209	88
	2015-1	.6	2016-1	7	2017-1	.8	2018-1	.9	2019-2	20	2020-2	1
	Headcount	FTES										
First Time	2,920	1,185	2,777	1,194	2,562	1,089	2,666	1,240	2,620	1,189	2,263	995
First Time Transfer	2,634	616	2,111	541	2,352	656	1,766	564	1,540	447	1,312	380
Continuing	10,178	5,991	10,502	6,487	9,986	6,305	9,576	6,120	9,325	5,977	8,237	5,234
Returning	3,196	675	2,277	551	2,382	539	1,964	496	2,231	504	1,926	495
Special Admit	935	173	2,260	353	2,578	424	2,281	425	3,521	574	3,288	505
Unknown	6	2	4	0	1	0	1	0	2	0		
Grand Total	17,004	8,641	17,217	9,126	17,235	9,014	15,597	8,845	16,523	8,691	14,794	7,608
	17,004	0,041	1,611	5,120	1,200	5,014	10,007	0,043	10,525	0,091	17,134	7,000

Percentage Point Gap (PPG)-compare a group outcome to the overall outcome, if group is 3% less or lower than overall then group is disproportionately impacted.

PPG Mod-same as PPG except overall outcome is modified to NOT include group outcome.

PPG Impact-amount of students needed to have a positive outcome in order to have the group reach equity.

					Academ 2020					
	Headcount	Enrollment	EW count	FTES	Retention %	PPG Retention Mod	PPG Retention Impact	Success %	PPG Success Mod	PPG Success Impact
Under 20	387	452	0	117	81.2%	-0.6%	3	64.2%	-4.5%	21
20-24	358	412	1	114	82.7%	1.3%		64.5%	0.3%	
25-29	83	95	1	26	75.5%	-4.1%	4	64.9%	3.3%	
30-34	55	58	1	16	78.9%	-2.2%	2	71.9%	6.2%	
35-39	24	25	1	7	91.7%	5.5%		83.3%	14.5%	
40-49	13	16	1	4	93.3%	6.9%		73.3%	2.6%	
50+	3	4	0	1	100.0%			75.0%		
Grand Total	906	1,062	5	286	81.6%			65.4%		

Percentage Point Gap (PPG)-compare a group outcome to the overall outcome, if group is 3% less or lower than overall then group is disproportionately impacted.

PPG Mod-same as PPG except overall outcome is modified to NOT include group outcome.

PPG Impact-amount of students needed to have a positive outcome in order to have the group reach equity.

					Academ 2020					
	Headcount	Enrollment	EW count	FTES	Retention %	PPG Retention Mod	PPG Retention Impact	Success %	PPG Success Mod	PPG Success Impact
Asian	17	18	0	5	77.8%	-2.7%	1	72.2%	9.3%	
Black	21	26	0	7	84.6%	2.6%		65.4%	0.9%	
Filipino	35	45	0	13	84.4%	0.7%		66.7%	0.9%	
Hispanic	422	491	1	132	79.8%	-3.6%	18	61.8%	-6.2%	31
Native Am	15	17	0	5	82.4%	0.7%		64.7%	-4.8%	1
Pac Isl	6	8	0	2	62.5%			62.5%		
White	382	446	4	120	83.9%	4.2%		69.5%	6.8%	
Unknown	8	11	0	3	72.7%			45.5%		
Grand Total	906	1,062	5	286	81.6%			65.4%		

Percentage Point Gap (PPG)-compare a group outcome to the overall outcome, if group is 3% less or lower than overall then group is disproportionately impacted.

PPG Mod-same as PPG except overall outcome is modified to NOT include group outcome.

PPG Impact-amount of students needed to have a positive outcome in order to have the group reach equity.

					Academ	ic Year				
					2020	)-21				
	Headcount	Enrollment	EW count	FTES	Retention %	PPG Retention Mod	PPG Retention Impact	Success %	PPG Success Mod	PPG Success Impact
Female	545	641	3	171	82.1%	1.5%		64.4%	-2.6%	17
Male	345	402	2	110	80.8%	-1.7%	7	66.8%	2.5%	
Unknown	16	19	0	5	84.2%	3.1%		68.4%	0.9%	
Grand Total	906	1,062	5	286	81.6%			65.4%		

Percentage Point Gap (PPG)-compare a group outcome to the overall outcome, if group is 3% less or lower than overall then group is disproportionately impacted.

PPG Mod-same as PPG except overall outcome is modified to NOT include group outcome.

PPG Impact-amount of students needed to have a positive outcome in order to have the group reach equity.

					Academ	ic Year				
					2020	)-21				
	Headcount	Enrollment	EW count	FTES	Retention %	PPG Retention Mod	PPG Retention Impact	Success %	PPG Success Mod	PPG Success Impact
First Time	130	130	0	31	81.5%	0.5%		58.5%	-10.0%	13
First Time Tran	37	37	0	10	83.8%	1.4%		67.6%	4.4%	
Continuing	762	865	5	237	81.3%	-2.7%	24	65.8%	3.7%	
Returning	20	20	0	6	85.0%	2.0%		70.0%	3.5%	
Special Admit	10	10	0	3	100.0%	17.6%		100.0%	13.3%	
Grand Total	906	1,062	5	286	81.6%			65.4%		

#### Equity:

Percentage Point Gap (PPG)-compare a group outcome to the overall outcome, if group is 3% less or lower than overall then group is disproportionately impacted.

PPG Mod-same as PPG except overall outcome is modified to NOT include group outcome.

PPG Impact-amount of students needed to have a positive outcome in order to have the group reach equity

			A	cademic Year			
				2020-21			
	Headcount	Enrollment	EW count	FTES	Retention %	PPG AHC Retention Mod	PPG AHC Retention Impact
Under 20	6,318	21,130	40	2,684	89.4%	1.3%	
20-24	3,806	14,590	42	2,106	87.5%	-1.7%	242
25-29	1,639	5,059	13	842	87.6%	-0.8%	41
30-34	1,009	3,021	15	472	88.6%	0.3%	
35-39	642	1,812	15	258	90.5%	1.4%	
40-49	660	1,821	7	266	89.0%	0.2%	
50+	407	1,058	13	134	91.0%	1.9%	
Grand Total	13,986	48,491	145	6,762	88.6%		

			A	cademic Yeai 2020-21	•		
	Headcount	Enrollment	EW count	FTES	Success %	PPG AHC Success Mod	PPG AHC Success Impact
Under 20	6,318	21,130	40	2,684	71.0%	-3.9%	831
20-24	3,806	14,590	42	2,106	73.1%	-0.6%	89
25-29	1,639	5,059	13	842	75.4%	2.2%	
30-34	1,009	3,021	15	472	77.6%	4.6%	
35-39	642	1,812	15	258	80.9%	6.7%	
40-49	660	1,821	7	266	79.9%	5.7%	
50+	407	1,058	13	134	81.1%	6.5%	
Grand Total	13,986	48,491	145	6,762	73.4%		





#### Equity:

Percentage Point Gap (PPG)-compare a group outcome to the overall outcome, if group is 3% less or lower than overall then group is disproportionately impacted.

PPG Mod-same as PPG except overall outcome is modified to NOT include group outcome.

PPG Impact-amount of students needed to have a positive outcome in order to have the group reach equity

			A	cademic Year			
				2020-21			
	Headcount	Enrollment	EW count	FTES	Retention %	PPG AHC Retention Mod	PPG AHC Retention Impact
Asian	242	833	0	120	90.3%	1.7%	
Black	398	1,352	7	202	88.1%	-1.1%	15
Filipino	378	1,445	2	207	90.6%	1.8%	
Hispanic	6,317	21,790	55	3,035	87.9%	-1.6%	349
Native Am	308	1,013	8	145	85.9%	-2.7%	28
Other	2	6	0	1	83.3%		
Pac Isl	125	418	0	56	87.1%	-1.6%	7
White	5,871	20,576	72	2,859	89.4%	1.7%	
Unknown	363	1,058	1	137	88.7%	0.0%	
Grand Total	13,986	48,491	145	6,762	88.6%		

	Academic Year 2020-21												
	Headcount	Enrollment	EW count	FTES	Success %	PPG AHC Success Mod	PPG AHC Success Impact						
Asian	242	833	0	120	77.4%	4.8%							
Black	398	1,352	7	202	69.3%	-5.8%	78						
Filipino	378	1,445	2	207	78.9%	5.3%							
Hispanic	6,317	21,790	55	3,035	70.9%	-4.6%	1,010						
Native Am	308	1,013	8	145	67.7%	-5.9%	60						
Other	2	6	0	1	83.3%								
Pac Isl	125	418	0	56	68.9%	-3.7%	15						
White	5,871	20,576	72	2,859	76.2%	5.0%							
Unknown	363	1,058	1	137	73.1%	-0.6%	6						
Grand Total	13,986	48,491	145	6,762	73.4%								

Academic Year 2015-16 2016-17 2017-18 2018-19 2019-20 2020-21



#### Equity:

Percentage Point Gap (PPG)-compare a group outcome to the overall outcome, if group is 3% less or lower than overall then group is disproportionately impacted.

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PPG Impact-amount of students needed to have a positive outcome in order to have the group reach equity

			A	cademic Year									
		2020-21											
	Headcount	Enrollment	EW count	FTES	Retention %	PPG AHC Retention Mod	PPG AHC Retention Impact						
Female	7,724	27,366	75	3,881	88.9%	0.6%							
Male	5,985	20,284	69	2,770	88.3%	-0.5%	99						
Unknown	283	841	1	111	87.0%	-1.8%	15						
Grand Total	13,986	48,491	145	6,762	88.6%								





	Academic Year 2020-21												
	Headcount	Enrollment	EW count	FTES	Success %	PPG AHC Success Mod	PPG AHC Success Impact						
Female	7,724	27,366	75	3,881	74.4%	2.3%							
Male	5,985	20,284	69	2,770	72.4%	-1.8%	361						
Unknown	283	841	1	111	66.1%	-8.0%	68						
Grand Total	13,986	48,491	145	6,762	73.4%								

#### Equity:

Percentage Point Gap (PPG)-compare a group outcome to the overall outcome, if group is 3% less or lower than overall then group is disproportionately impacted.

PPG Mod-same as PPG except overall outcome is modified to NOT include group outcome.

PPG Impact-amount of students needed to have a positive outcome in order to have the group reach equity

			A	cademic Year	r						
	2020-21										
	Headcount	Enrollment	EW count	FTES	Retention %	PPG AHC Retention Mod	PPG AHC Retention Impact				
First Time	2,157	7,000	27	898	86.4%	-2.2%	157				
First Time Tran	978	2,011	4	302	89.6%	1.8%					
Continuing	8,004	31,977	98	4,704	87.9%	-2.5%	798				
Returning	1,544	2,932	11	385	87.8%	-0.3%	10				
Special Admit	3,298	4,570	5	472	97.7%	9.7%					
Unknown	1	1	0	0	100.0%						
Grand Total	13,986	48,491	145	6,762	88.6%						

			A	cademic Year			
				2020-21			
	Headcount	Enrollment	EW count	FTES	Success %	PPG AHC Success Mod	PPG AHC Success Impact
First Time	2,157	7,000	27	898	60.4%	-13.5%	945
First Time Tran	978	2,011	4	302	78.4%	5.8%	
Continuing	8,004	31,977	98	4,704	74.0%	0.6%	
Returning	1,544	2,932	11	385	73.1%	0.8%	
Special Admit	3,298	4,570	5	472	87.3%	14.6%	
Unknown	1	1	0	0	100.0%		
Grand Total	13,986	48,491	145	6,762	73.4%		



**DemoChoice** Student Type

# 4 Online / Onsite course comparison CHEM \*All online courses and matching onsite courses\*

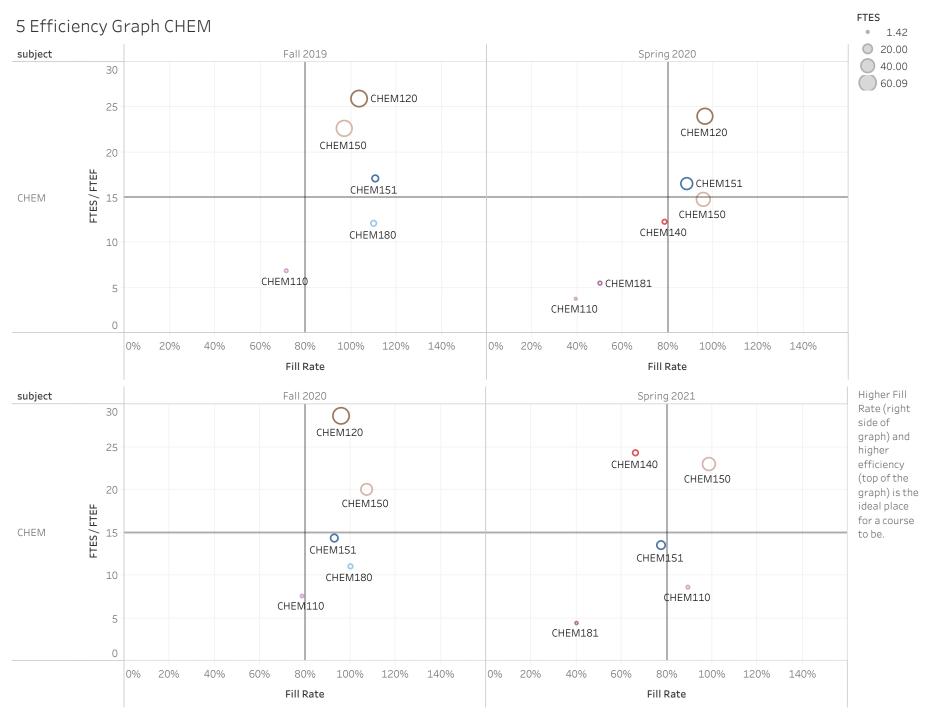
				Academic Year																		
				201	5-16		2016-17		2017-18		2019-20		2020-21									
subject	course	Course Type	Hea	Enr	Sect	FTES	Hea	Enr	Sect	FTES	Hea	Enr	Sect	FTES	Hea	Enr	Sect	FTES	Hea	Enr	Sect	FTES
CHEM	CHEM110	Online	79	83	3	10.75	65	68	3	8.81	27	27	1	3.50	30	31	2	4.02	58	59	3	7.64

# 4 Online / Onsite Retention & Success course comparison CHEM \*All online courses and matching onsite courses\*

	Academic Year										
subject_	course	Course	2015-16		2016-17		2017-18		2019-20		2020-21
CHEM	CHEM1	Online 7	71% 88%	63%	84%	48%	81%	45%	71%	58%	75%
Measure	Vames										
Reten	tion %										

# 4 Online / Onsite credit course comparison Allan Hancock College

				Academic	Year		
Course Type		2015-16	2016-17	2017-18	2018-19	2019-20	2020-21
Online	Headcount	7,580	7,006	7,152	6,744	7,040	7,440
	Enrollment	15,710	15,695	15,548	15,081	15,957	18,025
	Sections	509	517	501	457	487	586
	Retention %	83%	83%	84%	85%	87%	87%
	Success %	64%	66%	67%	68%	73%	71%
	FTES	1,496	1,524	1,523	1,490	1,569	1,790
Onsite	Headcount	13,623	14,458	14,466	13,515	14,715	13,013
	Enrollment	50,973	51,353	49,698	48,165	50,024	39,626
	Sections	2,284	2,279	2,231	2,164	2,278	1,965
	Retention %	90%	90%	89%	89%	91%	90%
	Success %	75%	76%	76%	75%	80%	76%
	FTES	7,145	7,775	7,511	7,403	7,313	5,969
Grand Total	Headcount	17,009	17,251	17,276	15,700	17,034	15,177
	Enrollment	66,683	67,048	65,246	63,246	65,981	57,651
	Sections	2,793	2,796	2,732	2,621	2,765	2,551
	Retention %	88%	88%	88%	88%	90%	89%
	Success %	72%	74%	74%	73%	78%	74%
	FTES	8,642	9,298	9,034	8,893	8,881	7,759



## 5 Efficiency Table CHEM

Academic Year	Term Code_	course_	FTES	FTEF+	FTES / FTEF	Enrollment	Maximum Enrollment	MaxEnroll	Fill Rate
2019-20	Sum 2019	CHEM120	25	2.156	11.8	126	140	28	90%
		CHEM140	3	0.386	6.6	13	28	28	46%
		Total	28	2.542	11.0	139	168	28	83%
	Fall 2019	CHEM110	3	0.376	6.9	20	28	28	71%
		CHEM120	60	2.316	25.9	261	252	28	104%
		CHEM150	56	2.475	22.7	163	168	28	97%
		CHEM151	10	0.575	17.1	31	28	28	111%
		CHEM180	7	0.575	12.1	22	20	20	110%
		Total	136	6.317	21.5	497	496	28	100%
	Spring 2020	CHEM110	1	0.376	3.8	11	28	28	39%
		CHEM120	56	2.316	24.0	243	252	28	96%
		CHEM140	5	0.376	12.3	22	28	28	79%
		CHEM150	43	2.875	14.8	134	140	28	96%
		CHEM151	31	1.900	16.5	99	112	28	88%
		CHEM181	3	0.575	5.5	10	20	20	50%
		Total	139	8.418	16.5	519	580	28	89%
	Total		302	17.277	17.5	1,155	1,244	28	93%
2020-21	Sum 2020	CHEM110	2	0.782	2.0	12	28	28	43%
		CHEM120	29	1.176	24.3	119	140	28	85%
		Total	30	1.958	15.4	131	168	28	78%
	Fall 2020	CHEM110	3	0.376	7.6	22	28	28	79%
		CHEM120	77	2.704	28.6	322	336	28	96%
		CHEM150	38	1.900	20.0	120	112	28	107%
		CHEM151	17	1.150	14.3	52	56	28	93%
		CHEM180	6	0.575	11.0	20	20	20	100%
		Total	141	6.705	21.1	536	552	28	97%
	Spring 2021	CHEM110	3	0.376	8.6	25	28	28	89%
		CHEM120	61	1.928	31.6	253	252	28	100%
		CHEM140	9	0.376	24.3	37	56	28	66%
		CHEM150	48	2.100	23.0	138	140	28	99%
		CHEM151	21	1.525	13.5	65	84	28	77%

## 5 Efficiency Table CHEM

Academic Year	Term Code_	course_	FTES	FTEF+	FTES / FTEF	Enrollment	Maximum Enrollment	MaxEnroll	Fill Rate
2020-21	Spring 2021	CHEM181	3	0.575	4.4	8	20	20	40%
		Total	145	6.880	21.0	526	580	28	91%
	Total		316	15.543	20.3	1,193	1,300	28	92%
Grand Total			618	32.820	18.8	2,348	2,544	28	92%

## 6 Degree/Certificate Chemistry

	Program Desc	Degree	Degree Major	Degree Desc (group)	2015-2016	2016-2017	2017-2018	2018-2019	2019-2020	2020-2021
Unduplicated	Chemistry	AA	Chemistry	Associate in Arts	10	9	10	14	8	10
		AS-T	Chemistry for Transf	Associate in Science-Transfer			1	3	1	
			Chemistry for Transf	Associate in Science-Transfer						2
Duplicated	Chemistry	AA	Chemistry	Associate in Arts	10	9	10	14	8	10
		AS-T	Chemistry for Transf	Associate in Science-Transfer			1	3	1	
			Chemistry for Transf	Associate in Science-Transfer						2
Unduplicated	Total				10	9	10	16	8	12
Duplicated	Total				10	9	11	17	9	12

## 6 Majors Chemistry - Headcount

	2015-16	2016-17	2017-18	2018-19	2019-20	2020-21
Chemistry	120	151	140	125	113	79
Chemistry for Transfer CSU			15	35	47	35
Chemistry for Transfer UC					11	23
Grand Total	120	151	154	158	163	134

Academic Year Graduation Desc

## 6 Chemistry Award | Major Match

--If a student has the same program of study and major as the award earned they will be a 'Major Match'. If not they will be a 'Major Split'.

--Headcount & Percentages are the students who are a major match/split for a specific award.

--Data is sorted by program/major of the earned award.

								A	cademi	crear	Graduat	Ion De	SC			
					2015-	2016	2016-	2017	2017-	2018	2018-	2019	2019	2020	2020-	2021
Program	Degree	Degree Major	Degree Desc (group)	Major	HC	%	НC	%	НC	%	HC	%	НС	%	НC	%
Chemistry	AA	Chemistry	Associate in Arts	Match	1	10%	2	22%	1	10%	1	7%			1	10%
				Split	9	90%	7	78%	9	90%	13	93%	8	100%	9	90%
	AS-T	Chemistry for Transfer	Associate in Science-Transfer	Match							2	67%	1	100%		
		CSU		Split					1	100%	1	33%				
		Chemistry for Transfe	Associate in Science-Transfer	Split											2	100%
	Total				10	100%	9	100%	10		16		8		12	

#### Academic Year Graduation Desc

## 6 Degree/Certificate Allan Hancock College

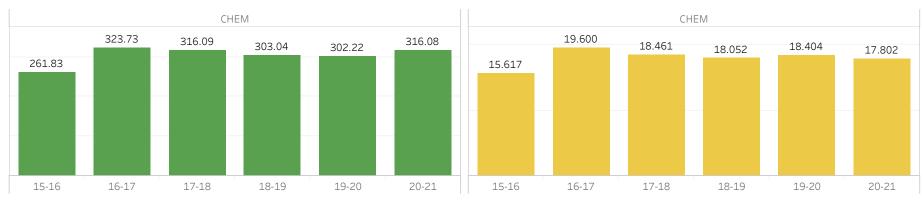
			Ac	ademic Year G	raduation Des	c	
	Degree Desc (group)	2015-2016	2016-2017	2017-2018	2018-2019	2019-2020	2020-2021
Unduplicated	Associate in Arts	494	523	493	589	882	885
	Associate in Arts - Transfer	92	126	159	164	218	262
	Associate in Science	277	319	313	321	304	310
	Associate in Science-Transfer	95	128	126	191	228	249
	Certificate of Accomplishment	381	419	416	372	423	328
	Certificate of Achievement	681	795	791	876	810	1,156
	NC Cert 48 to <96 hrs	3	10	22	21	22	5
	NC Cert 144 to <192 hrs						6
	NC Cert 192 to <288 hrs	7	5	1	6	13	
	NC Cert 288 to <480 hrs	2	27	46	38	32	3
	NC Cert 480 to <960 hrs			2	9	32	
Duplicated	Associate in Arts	709	726	737	814	1,437	1,616
	Associate in Arts - Transfer	95	130	163	165	229	341
	Associate in Science	307	347	345	350	335	332
	Associate in Science-Transfer	99	133	138	207	237	323
	Certificate of Accomplishment	404	501	491	417	478	373
	Certificate of Achievement	722	846	870	958	865	1,636
	NC Cert 48 to <96 hrs	3	10	23	21	22	5
	NC Cert 144 to <192 hrs						6
	NC Cert 192 to <288 hrs	7	5	1	6	13	
	NC Cert 288 to <480 hrs	2	34	46	39	33	3
	NC Cert 480 to <960 hrs			2	9	32	
Unduplicated	Total	1,491	1,703	1,673	1,804	1,972	1,983
Duplicated	Total	2,348	2,732	2,816	2,986	3,681	4,635

### 7 FTEF+Overload, FTES & Efficiency - CHEM

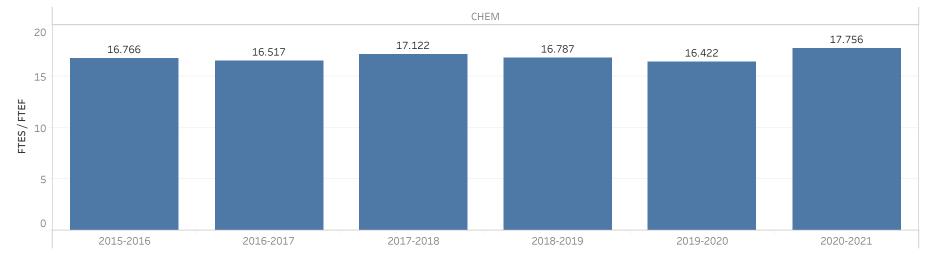
								Academ	ic Year								
2	015-2016		2	016-2017		2	017-2018		2	018-2019		2	019-2020		2	020-2021	
		FTES/			FTES/			FTES/			FTES/			FTES/			FTES/
FTEF+	FTES	FTEF	FTEF+	FTES	FTEF	FTEF+	FTES	FTEF	FTEF+	FTES	FTEF	FTEF+	FTES	FTEF	FTEF+	FTES	FTEF
15.617	261.83	16.8	19.600	323.73	16.5	18.461	316.09	17.1	18.052	303.04	16.8	18.404	302.22	16.4	17.802	316.08	17.8

FTEF





FTEF/ FTES

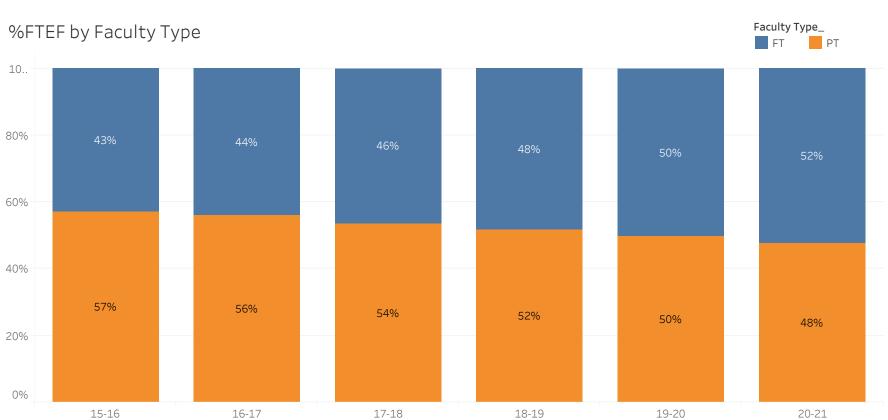


## Faculty Type

	ections FTEF Overload_ Faculty Section
CHEM         Instructional - FT         5.004         2.530         2         17         7.955         2.795         4	26 8.771 2.757 4 2.
Instructional - PT 8.083 0.000 6 22 8.850 0.000 6	25 6.933 0.000 8 20
Grand Total 13.087 2.530 8 38 16.805 2.795 10	48 15.704 2.757 12 4
2018-2019 2019-2020	2020-2021
	ections FTEF Overload_ Faculty Section
CHEM         Instructional - FT         9.98         3.06         5         28         8.39         2.50         4	25 6.94 2.80 4 2.
Instructional - PT 5.02 0.00 5 14 7.52 0.00 7	21 8.07 0.00 6 2
Grand Total 14.99 3.06 9 42 15.91 2.50 11	46 15.01 2.80 10 4
%FTEF by Faculty Type     Instructional - FT     Faculty Type       Instructional - PT     Instructional - PT	aculty count by type
10 28%	4 4 5
30%     38%     47%     56%     53%     46%     6     6       50%     67%     6     6     6	8 5 7
15-16 16-17	17-18 18-19 19-20
40% 62%	Overload
53% 47% 54%	CHEM
20%	2.7570 3.0580 2.4970
0%	

## 7 FTEF+Overload by Faculty Type Allan Hancock College

			Academie	Year		
Faculty Type	2015-2016	2016-2017	2017-2018	2018-2019	2019-2020	2020-2021
Instructional - FT	310.594	331.703	344.107	343.923	340.591	328.688
Instructional - PT	359.820	355.797	331.111	315.432	300.351	263.265
Total	670.414	687.500	675.218	659.355	640.942	591.953
	670.414	687.500	675.218	659.355	640.942	591.953
		Instructional - FT310.594Instructional - PT359.820Total670.414	Instructional - FT         310.594         331.703           Instructional - PT         359.820         355.797           Total         670.414         687.500	Faculty Type2015-20162016-20172017-2018Instructional - FT310.594331.703344.107Instructional - PT359.820355.797331.111Total670.414687.500675.218	Instructional - FT         310.594         331.703         344.107         343.923           Instructional - PT         359.820         355.797         331.111         315.432           Total         670.414         687.500         675.218         659.355	Faculty Type2015-20162016-20172017-20182018-20192019-2020Instructional - FT310.594331.703344.107343.923340.591Instructional - PT359.820355.797331.111315.432300.351Total670.414687.500675.218659.355640.942



## Appendix A: Program/Course Demographics by Outcome CHEM

							Academ	nic Year					
			201	3-19			2019	<del>9</del> -20			2020	0-21	
		Headcou	FTES	Retention %	Success %	Headcou	FTES	Retention %		Headcou	FTES	Retention %	
CHEM110	Under 20					7	0.9	71%	57%	17	2.2	76%	53%
	20-24					8	1.0	50%	38%	18	2.3	72%	56%
	25-29					4	0.6	100%	20%	6	0.8	67%	67%
	30-34					6	0.8	100%	83%	2	0.3	100%	50%
	35-39					2	0.3	0%	0%	2	0.3	100%	100%
	40-49					2	0.3	50%	50%	1	0.3	50%	50%
	50+					1	0.1	100%	0%				
CHEM120	Under 20	320	73.0	88%	69%	297	68.5	90%	76%	284	68.4	83%	65%
	20-24	209	48.0	84%	65%	205	47.4	86%	77%	171	42.8	86%	68%
	25-29	68	15.4	83%	71%	59	13.2	82%	75%	52	13.6	81%	72%
	30-34	18	4.5	79%	63%	28	6.6	96%	93%	29	7.1	89%	86%
	35-39	11	2.6	91%	82%	9	1.9	100%	100%	15	4.0	93%	80%
	40-49	7	1.6	100%	100%	12	2.6	92%	92%	8	1.9	100%	100%
	50+	3	0.6	67%	33%	4	0.9	100%	100%	2	0.5	100%	50%
CHEM140	Under 20	2	0.4	50%	50%	11	2.2	100%	100%	5	1.2	100%	100%
	20-24	18	3.8	78%	78%	12	2.4	100%	100%	21	5.1	95%	86%
	25-29	4	0.9	75%	75%	4	0.8	100%	100%	3	0.7	33%	33%
	30-34	1	0.2	0%	0%	4	0.8	100%	100%	6	1.6	83%	67%
	35-39					2	0.4	100%	100%	2	0.5	50%	50%
	40-49					1	0.2	100%	100%				
	50+					1	0.2	100%	100%				
CHEM150	Under 20	130	41.3	92%	78%	134	46.8	84%	71%	94	33.2	76%	62%
	20-24	128	42.2	85%	71%	110	36.6	84%	77%	104	35.8	85%	69%
	25-29	35	12.1	79%	74%	22	7.3	75%	70%	20	7.3	71%	52%
	30-34	14	4.4	86%	79%	12	4.0	91%	82%	13	5.0	57%	57%
	35-39	2	0.6	100%	100%	8	2.5	88%	88%	6	2.3	100%	100%
	40-49					4	1.3	33%	0%	5	1.9	100%	50%
	50+									2	0.8	100%	100%
CHEM151	Under 20	32	10.2	91%	84%	35	11.7	91%	88%	36	11.4	78%	61%
	20-24	75	25.7	86%	75%	63	20.6	91%	88%	58	20.9	73%	47%

## Appendix A: Program/Course Demographics by Outcome CHEM

							Academ	nic Year					
			2018-	19			2019	9-20			2020	)-21	
		Headcou	FTES	Retention %	Success %	Headcou	FTES	Retention %	Success %	Headcou	FTES	Retention %	Success %
CHEM151	25-29	13	4.4	64%	64%	16	5.1	94%	88%	7	2.5	75%	63%
	30-34	7	2.2	86%	86%	7	2.5	88%	75%	6	1.9	67%	67%
	35-39	1	0.3	100%	100%	3	1.0	100%	100%				
	40-49	1	0.3	100%	100%	1	0.3	100%	100%	1	0.3	100%	0%
CHEM180	Under 20	3	1.0	67%	33%	6	1.9	100%	100%	1	0.3	100%	100%
	20-24	14	4.4	64%	57%	13	4.1	62%	54%	16	5.1	53%	33%
	25-29	2	0.6	0%	0%	1	0.3	0%	0%	2	0.6	50%	50%
	30-34									1	0.3	100%	0%
	40-49					1	0.3	100%	100%				
	50+					1	0.3	0%	0%				
CHEM181	Under 20	1	0.3	100%	100%	2	0.6	100%	100%				
	20-24	6	1.9	83%	83%	7	2.2	100%	100%	7	2.2	100%	100%
	25-29					1	0.3	100%	100%	1	0.3	100%	100%

#### Appendix B: Major match detail

--If a student has the same program of study and major as the award earned they will be a 'Major Match'. If not they will be a 'Major Split'.

--Headcount & Percentages are the students who are a major match/split for a specific award.

--Data is sorted by program/major of the earned award.

						I	Aca	ademic Year (	Graduation D	esc	
Major Match	Program Desc	Degree	Degree Major	Student Major	Degree Desc (group)	2015-2016	2016-2017	2017-2018	2018-2019	2019-2020	2020-2021
Match	Chemist	AA	Chemistry	Chemistry	Associate in Arts	1	2	1	1		1
		AS-T	Chemistry for Transfer CSU	Chemistry for Transfer CSU	Associate in Science-Tra				2	1	
		Total				1	2	1	3	1	1
	Total					1	2	1	3	1	1
Split	Chemist	AA	Chemistry	Biology	Associate in Arts			1			
				Chemistry for Transfer CSU	Associate in Arts					1	
				Civil Engineering	Associate in Arts	1	1	2			
				Computer Science	Associate in Arts				2		
				Electronic Engineering Tech	Associate in Arts				2		
				Electronic Technology	Associate in Arts					1	
				Engineering	Associate in Arts	6	4	4	9	4	8
				Engineering Technology	Associate in Arts		1				
				Engr Tech: Mechatronics	Associate in Arts			1			
				Math: Physics Emphasis	Associate in Arts						1
				Mathematics and Science CSU	Associate in Arts	1	1				
				Mathematics: Comp Sci Emph	Associate in Arts					1	
				Spanish for Transfer CSU	Associate in Arts					1	
				Undeclared	Associate in Arts	1					
				Viticulture	Associate in Arts			1			
		AS-T	Chemistry for Transfer CSU	Chemistry	Associate in Science-Tra				1		
				Engineering	Associate in Science-Tra			1			
			Chemistry for Transfer UC	Biology for Transfer UC	Associate in Science-Tra						1
				Chemistry for Transfer CSU	Associate in Science-Tra						1
		Total				9	7	9	14	8	11
	Total					9	7	9	14	8	11
Grand 1	Total					10	9	10	16	8	12

#### **COURSE REVIEW VERIFICATION**

Discipline: Chemistry Year: 2021-2022

As part of the program evaluation process, the self-study team has reviewed the course outlines supporting the

discipline/program curriculum. The review process has resulted in the following recommendations:

1. The following course outlines are satisfactory as written and do not require modification (list all such courses):

#### CHEM 120, 140, 150, 151, 180, and 181.

- 2. The following courses require minor modification to ensure currency. The self study team anticipates submitting such modifications to the AP&P, FALL 20\_\_\_\_\_ SPRING 20\_\_\_\_:
- 3. The following courses require major modification. The self study team anticipates submitting such modifications to the AP&P committee, FALL 20\_\_\_\_\_ SPRING 20\_\_\_\_:

# GRADUATION REQUIREMENTS: General Education (GE), Multicultural/Gender Studies (MCGS) and Health & Safety (H&W) Courses.

The following courses were reviewed as meeting an **AHC GE** requirement. The AP&P GE Criteria and Category Definitions (GE Learning Outcomes) forms were submitted to the AP&P for review on:

The following courses were reviewed as meeting the **MCGS** requirement. The AP&P MCGS Criteria and Category Definitions (MCGS Learning Outcomes – To Be Developed) forms were submitted to the AP&P for review on:

The following courses were reviewed as meeting the **H&W** requirement. The AP&P H&W Studies Criteria (To Be Developed) and Category Definitions (H&W Learning Outcomes – To Be Developed) forms were submitted to the AP&P chair for review on: \_\_\_\_\_\_

Course Review Team Members:

Name	Signature	Date	
Name	Signature	Date	
Name	Signature	Date	
Name	Signature	Date	
Name	Signature AP&P Chair	Date	
Name	Signature Academic Dean	Date	

# <u>APPENDICES</u> Approved Course Outlines Advisory Committee

#### CONTENT REVIEW WITHIN THE SAME DISCIPLINE OR ACROSS DISCIPLINES

Course Prefix and Number <u>CHEM 120</u>	
Department Life and Physical Sciences Responsible Instructor Dustin Nouri	
PCA being reviewed: <u>UC/CSU Comparison Sheets</u> Type: <u>Prerequisite</u> Use one form for each prerequisite/corequisite/advisory if the course has more than one Review Team (Recommended: four instructors; preferably two of whom teach the course being rev one who teaches the preceding course, and one who teaches the subsequent course, as appropriate	∕iewed; ite)
Sean Gottlieb     Danae Madrid	,
Dustin Nouri	

**Recommended Materials:** 

- 1. Course outline for course being reviewed preceding course and subsequent course
- 2. For each course, current text, typical tests, sample projects, quizzes, and any other relevant evaluation tools as used within the courses and evidenced by the course outline, written grading standards (possibly from syllabus).
- 3. EVA report from Computer Services reflecting student success based on completion and noncompletion of prerequisite course.

#### Process:

- 1. <u>Examine objectives of course being reviewed</u>. Are objectives current? <u>YES</u> Will student successfully completing this course have a reasonable chance of success in subsequent course? <u>YES</u>
- 2. <u>Examine objectives of preceding course</u>. Are the objectives equivalent of the entrance skills necessary to succeed in the course being reviewed? <u>YES</u>
- 3. Examine the evaluation tools used within the course.

Do the tests, quizzes, projects, assignments reflect skills which the student would have acquired in the preceding course? <u>YES</u>

4. Examine the text used for the course Does the textbook require a base of knowledge the student would have obtained in the preceding course? <u>YES</u>

<u>X</u> The prerequisite/corequisite/advisory is appropriate.	
The prerequisite/corequisite/advisory should be deleted.	
The prerequisite/corequisite/advisory should be modified.	
The course outline should be modified to reflect outcomes of content review, and to include entrance/exit skills.  Dustin Nouri  2021  9  2  Ashley Wise	all 2021 9/21/2022
Initiator Date Department Chair	Date
Approved:,Academic Dean	Date

CONTENT REVIEW
WITHIN THE SAME DISCIPLINE OR ACROSS DISCIPLINES

Course Prefix and Number <u>CHEM 14</u>	10	
Department Life and Physical Sciences Res	ponsible Instruct <u>or_Dustin Nc</u>	uri
PCA being reviewed: <u>UC/CSU Comparison</u> Use one form for each prerequisite/co Review Team (Recommended: four instructor	orequisite/advisory if the course has i	more than one
one who teaches the preceding course, and o	ne who teaches the subseque	ent course, as appropriate)
Sean Gottlieb	Danae Madrid	
Dustin Nouri		

**Recommended Materials:** 

- 1. Course outline for course being reviewed preceding course and subsequent course
- 2. For each course, current text, typical tests, sample projects, quizzes, and any other relevant evaluation tools as used within the courses and evidenced by the course outline, written grading standards (possibly from syllabus).
- 3. EVA report from Computer Services reflecting student success based on completion and noncompletion of prerequisite course.

#### Process:

- 1. <u>Examine objectives of course being reviewed</u>. Are objectives current? <u>YES</u> Will student successfully completing this course have a reasonable chance of success in subsequent course? <u>YES</u>
- <u>course.</u> Do the tests, quizzes, projects, assignments reflect skills which the student would have acquired in the preceding course? <u>YES</u>

3. Examine the evaluation tools used within the

- 2. <u>Examine objectives of preceding course</u>. Are the objectives equivalent of the entrance skills necessary to succeed in the course being reviewed? <u>YES</u>
- 4. <u>Examine the text used for the course</u> Does the textbook require a base of knowledge the student would have obtained in the preceding course? <u>YES</u>

<u>X</u> The prerequisite/corequisite/advisory is appropriate.	
The prerequisite/corequisite/advisory should be deleted.	
The prerequisite/corequisite/advisory should be modified.	
The course outline should be modified to reflect outcomes of content review, and to include entrance/exit skills. <b>Dustin Nouri</b> 2021 Ashley Wise 2021 9/2/20	22
Initiator Date Department Chair Date	
Approved:,Academic Dean Date	

CONTENT REVIEW
WITHIN THE SAME DISCIPLINE OR ACROSS DISCIPLINES

Course Prefix and Nu	mber <u>CHEM 150</u>				
Department <u>Life and</u>	Physical Sciences Respons	sible Instruct <u>or Dus</u>	stin Nou	ri	
	UC/CSU Comparison She				
Review Team (Recom one who teaches the	nmended: four instructors; pr preceding course, and one w	eferably two of who ho teaches the sub	om teac sequen	h the course being r t course, as approp	eviewed; riate)
Sean Got	tlieb	Danae Mad	rid		

Dustin Nouri

**Recommended Materials:** 

- 1. Course outline for course being reviewed preceding course and subsequent course
- 2. For each course, current text, typical tests, sample projects, quizzes, and any other relevant evaluation tools as used within the courses and evidenced by the course outline, written grading standards (possibly from syllabus).
- 3. EVA report from Computer Services reflecting student success based on completion and noncompletion of prerequisite course.

Process:

- 1. <u>Examine objectives of course being reviewed</u>. Are objectives current? <u>YES</u> Will student successfully completing this course have a reasonable chance of success in subsequent course? <u>YES</u>
- 2. <u>Examine objectives of preceding course</u>. Are the objectives equivalent of the entrance skills necessary to succeed in the course being reviewed? <u>YES</u>
- 3. Examine the evaluation tools used within the course.

Do the tests, quizzes, projects, assignments reflect skills which the student would have acquired in the preceding course? <u>YES</u>

4. <u>Examine the text used for the course</u> Does the textbook require a base of knowledge the student would have obtained in the preceding course? <u>YES</u>

<u>X</u> The prerequisite/corequisite/advisory is appropriate.	
The prerequisite/corequisite/advisory should be deleted.	
The prerequisite/corequisite/advisory should be modified.	
The course outline should be modified to reflect outcomes of content review, and to include entranse exit skills. <b>Dustin Nouri</b> 2021 9/27 <u>Ashley Wise</u> 2021 9/21/2	2022
Initiator Date Department Chair Date	
Approved:Academic Dean Date <u>1/20/202</u>	

CONTENT REVIEW			
WITHIN THE SAME DISCIPLINE OR ACROSS DISCIPLINES			

Course Prefix and Number <u>CHEM 151</u>	
Department Life and Physical Sciences Responsible Instructor Dustin Nouri	
PCA being reviewed: <u>UC/CSU Comparison Sheets</u> Type: <u>Prerequisite</u> Use one form for each prerequisite/corequisite/advisory if the course has more than one	
Review Team (Recommended: four instructors; preferably two of whom teach the course being review one who teaches the preceding course, and one who teaches the subsequent course, as appropriate)	/ed;

Sean Gottlieb	Danae Madrid	
Dustin Nouri		

**Recommended Materials:** 

- 1. Course outline for course being reviewed preceding course and subsequent course
- 2. For each course, current text, typical tests, sample projects, quizzes, and any other relevant evaluation tools as used within the courses and evidenced by the course outline, written grading standards (possibly from syllabus).
- 3. EVA report from Computer Services reflecting student success based on completion and noncompletion of prerequisite course.

#### Process:

- 1. <u>Examine objectives of course being reviewed</u>. Are objectives current? <u>YES</u> Will student successfully completing this course have a reasonable chance of success in subsequent course? <u>YES</u>
- 2. Examine objectives of preceding course. Are the objectives equivalent of the entrance skills necessary to succeed in the course being reviewed? <u>YES</u>
- 3. Examine the evaluation tools used within the course.

Do the tests, quizzes, projects, assignments reflect skills which the student would have acquired in the preceding course? <u>YES</u>

4. <u>Examine the text used for the course</u> Does the textbook require a base of knowledge the student would have obtained in the preceding course? <u>YES</u>

$\underline{\mathbf{X}}$ The prerequisite/corequisite/advisory	is appropriate.	
The prerequisite/corequisite/advisory	should be deleted.	
The prerequisite/corequisite/advisory	should be modified.	
The course outline should be modified review, and to include entrance/exit s Dustin Nouri		ally Win - 2021 9/21/2022
Initiator Date	Department Chair	Date
Approved:	Academic Dean	Date

#### CONTENT REVIEW WITHIN THE SAME DISCIPLINE OR ACROSS DISCIPLINES

Course Prefix and Number <u>CHEM</u>	[ 180			
Department Life and Physical Sciences	Responsible Instructor Dr	ustin Not	uri	
PCA being reviewed: <u>UC/CSU Compa</u> Use one form for each prerequ	rison Sheets uisite/corequisite/advisory if the co	Type: urse has m		
Review Team (Recommended: four instr one who teaches the preceding course, a				əd;
Sean Gottlieb	Danae Ma	drid		

Recommended Materials:

Dustin Nouri

- 1. Course outline for course being reviewed preceding course and subsequent course
- 2. For each course, current text, typical tests, sample projects, quizzes, and any other relevant evaluation tools as used within the courses and evidenced by the course outline, written grading standards (possibly from syllabus).
- 3. EVA report from Computer Services reflecting student success based on completion and noncompletion of prerequisite course.

Process:

- 1. <u>Examine objectives of course being reviewed</u>. Are objectives current? <u>YES</u> Will student successfully completing this course have a reasonable chance of success in subsequent course? <u>YES</u>
- 2. Examine objectives of preceding course. Are the objectives equivalent of the entrance skills necessary to succeed in the course being reviewed? <u>YES</u>
- 3. Examine the evaluation tools used within the course.

Do the tests, quizzes, projects, assignments reflect skills which the student would have acquired in the preceding course? <u>YES</u>

4. <u>Examine the text used for the course</u> Does the textbook require a base of knowledge the student would have obtained in the preceding course? <u>YES</u>

$\underline{\mathbf{X}}$ The prerequisite/corequisite/advisory is appropriate.	
The prerequisite/corequisite/advisory should be deleted.	
The prerequisite/corequisite/advisory should be modified. (circle one)	
The course outline should be modified to reflect outcomes of content review, and to include entrance/exit skills.  Dustin Nouri 2021 Ashley Wise	Only Win + 1/2022
Initiator Department Chair	Date
Approved:,Academic Dean	Date <u>a popor</u>

CONTENT REVIEW WITHIN THE SAME DISCIPLINE OR ACROSS DISCIPLINES	
Course Prefix and Number <u>CHEM 181</u>	
Department Life and Physical Sciences Responsible Instructor Dustin Nouri	
PCA being reviewed: <u>UC/CSU Comparison Sheets</u> Type: <u>Prerequisite</u> Use one form for each prerequisite/corequisite/advisory if the course has more than one	
Review Team (Recommended: four instructors; preferably two of whom teach the course being roome who teaches the preceding course, and one who teaches the subsequent course, as appropriate the subsequent course.	eviewed; iate)

Sean Gottlieb	Danae Madrid
Dustin Nouri	

**Recommended Materials:** 

- 1. Course outline for course being reviewed preceding course and subsequent course
- 2. For each course, current text, typical tests, sample projects, quizzes, and any other relevant evaluation tools as used within the courses and evidenced by the course outline, written grading standards (possibly from syllabus).
- 3. EVA report from Computer Services reflecting student success based on completion and noncompletion of prerequisite course.

#### Process:

- 1. <u>Examine objectives of course being reviewed</u>. Are objectives current? <u>YES</u> Will student successfully completing this course have a reasonable chance of success in subsequent course? <u>YES</u>
- 2. Examine objectives of preceding course. Are the objectives equivalent of the entrance skills necessary to succeed in the course being reviewed? <u>YES</u>
- 3. Examine the evaluation tools used within the course.

Do the tests, quizzes, projects, assignments reflect skills which the student would have acquired in the preceding course? <u>YES</u>

4. <u>Examine the text used for the course</u> Does the textbook require a base of knowledge the student would have obtained in the preceding course? <u>YES</u>

$\underline{X}_{(circle one)} The prerequisite/corequisite/advisory$	is appropriate.
The prerequisite/corequisite/advisory	should be deleted.
The prerequisite/corequisite/advisory	should be modified.
The course outline should be modified review, and to include entrance/exit s Dustin Nouri	
Initiator Date	Department Chair Date
Approved:	Academic Dean Date70/2022

#### **PROGRAM REVIEW -- VALIDATION TEAM MEMBERS**

TO: Academic Dean

Date: 1/26/2022

From: Dustin Nouri

We recommend the following persons for consideration for the validation team:

DEPARTMENT

Board Policy <u>requires</u> that the validation team be comprised of the dean of the area, one faculty member from a related discipline/program, and two faculty members from unrelated disciplines.

\_\_\_\_\_CHEMISTRY\_\_

Sean Gottlieb and Danae Madrid	Chemistry	
(Name)	(Related Discipline/Program)	
Patrick McGuire	Automotive	
(Name)	(Unrelated Discipline/Program)	
Michael Wagner	Computer Science/Mathematics	
(Name)	(Unrelated Discipline/Program)	

At the option of the self-study team, the validation team IIIIIy also include one or more of the following: a someone from a four-year institution in the same discipline; someone from another community college in the same discipline; a high school instructor in the same discipline; a member of an adviso committee for the ro am. Please com lete the followin as relevant to our ro am review.

(Name)		(Title)
Affiliation:	Telephone	Contact Number:
Address		
<u>M</u> ailin	City/State/Zip	email address
(Name)		(Title)
Affiliation:	Telephone	Contact Number:
Address(Mailing)	City/State/Zip	email address
(Name)		(Title)
Affiliation:	Telephone	Contact Number:
Address(Mailing)	City/State/Zip	email address
APPROVED:	enne Dezn	
, Acag	chile Deall	Date

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# **VALIDATION** Executive Summary Plan of Action - Post Validation

#### EXECUTIVE SUMMARY (Validation Team Report)

The Validation Team for the 2022 Chemistry six-year program review—consisting of Computer Science faculty member Michael Wagner, Automotive faculty member Patrick McGuire, authors of the Program Review and Chemistry faculty members Dustin Nouri and Sean Gottlieb, and Dean Sean J. Abel—met to review and discuss the comprehensive program review for approximately 75 minutes on Friday September 2, 2022. It was clear that each member of the team had reviewed the document with care and came prepared to provide feedback and suggestions to the document's authors.

#### 1. MAJOR FINDINGS

#### Strengths of the program/discipline:

The team members noted the attention to detail, consideration, and thought that was evident throughout the document. The team was able to discuss the content of the document and work with the author to clarify the impact of the document to the program.

As the team reviewed and reflected upon the document together, they commented on the positive way that the program meets the needs of all students through courses focused on transfer and meeting the C-ID and Associate of Science in Chemistry for Transfer, general education, and prerequisites for degrees and certificates in the various Allied Health programs. Student survey data reflected satisfaction with the program and how it meets their needs and educational goals. This has resulted in program courses being well-enrolled to the point where there are consistently waitlists on a significant majority of the course sections including steady growth at the Lompoc Valley Center campus. The chemistry faculty have further helped student success and completion by ensuring thorough guided pathways mapping and implementation of that mapping by scheduling to follow those maps.

This program growth resulted in the recent addition of new full-time faculty members (one replacement and one growth). The team discussed this addition to the program's high quality, caring faculty as well as the college's support of a mid-year faculty search to replace a recent, unexpected full-time faculty member's resignation. Such broad college support indicates the institutional importance of chemistry as a discipline and the impact it has on nearly every student attending the college. By keeping maps up-to-date, chemistry participates in excellent inter-disciplinary cooperation and coordination for the benefit of our students.

Other areas of strength include chemistry facilities, particularly the laboratory classroom spaces. These spaces well support the course and program curriculum and are designed with safety in mind. The chemistry program is also typically well-funded for needed equipment and supplies, although there are always needs to be met as processes and technologies change. Our lab specialists (one in Santa Maria, one in Lompoc) keep equipment up-to-date as needed. An example of this would be the recent replacement of analytical balances, similar to other

equipment at the end of its useful life.

#### Concerns regarding the program/discipline:

As the team discussed the document with the authors, challenges for the program were remarked upon by all. Three highly impactful main themes were discussed—student preparation for chemistry courses, space/facilities resources, and anticipated staffing needs.

The authors pointed out a concern regarding student preparation for chemistry courses, particularly after several semesters of pandemic-related online and/or hybrid classes. This has been particularly prominent in mathematics preparation. This concern is compounded by the change in the enforcement of AB705 rules which have eliminated developmental mathematics courses. Students are frequently having to catch up in their mathematics skills while struggling with entirely different mathematics concepts in chemistry, and presumably, the other sciences.

The Chemistry (and most other science) programs are struggling with space to accommodate enrollment, resulting in frequent large waitlist which impacts the students' abilities to complete their degree programs in a timely manner. Although the spaces we have are excellent, the college lacks sufficient classrooms which can accommodate double lab lectures of nearly 60 students. As the chemistry program continues to grow, there is a need for an additional chemistry lab on the Santa Maria campus and additional shared lab space on the Lompoc Valley Center campus.

The final impactful main theme focused on staffing. Given the college's location, it can be very difficult to find highly qualified part time faculty members who are able teach at a broad spectrum of times. The expansion of full-time faculty by one or more members may help ameliorate staffing issues during the day, but that remains to be seen over time. As the program grows, another staffing concern is lab specialists/associates/assistants on both campuses. Current incumbents are managing with the existing schedule but additions may strain their capacities. Additionally, student survey reflected a need for chemistry related tutorial services. The underlying reason for this was unclear. The team discussed strategies for improving the connection between chemistry students and the STEM center for tutoring.

Lastly, the team discussed two other specific challenges. The first of these focused on the costs for science classes in general. Particularly for chemistry, some published materials are very expensive for students and this may be a barrier to completion. The second was a concern regarding disconnect between the desires of the Chancellor's office/legislature related to ADT units and the reality of course unit requirements in the C-ID descriptors. The legislation requires that Associate Degrees for Transfer be only 60 units; however, transfer institutions require specific courses in the C-ID descriptors to have more units than are noted in the ADT templates. This is a challenge that is likely beyond the scope of the program to correct, but needed to be noted.

#### 2. RECOMMENDATIONS

Based on the discussion and program challenges, the team and authors proposed recommendations. The subject of the first recommendation was how to bridge this program review process to the process starting in the 2022-2023 academic year. One team member

indicated that they would assist the authors in formatting the action plan to appropriately match the new program review process. As we discussed the above listed concerns for the program, the team outlined various recommendations. It was recommended that faculty members in the chemistry program continue to communicate with other disciplines, especially mathematics, to benefit student success. Of particular note was the recommendation to investigate how mathematics and chemistry can collaborate to improve students' chemistry-related mathematics skills. Similar to other life and physical science program reviews, the team recommended that the program continue to advocate for additional lecture and lab space through the master planning process as well as investigate expanding online chemistry general education offerings in order to help ameliorate the impact of waitlists in the program. Because the chemistry program is continuing on a growth trajectory, the team recommends that the program continues to advocate for additional full-time faculty members and lab specialists/associates/assistants to meet the needs of increasing numbers of students. As part of equipment procurement, the team recommends that a detailed repair and replacement schedule be developed for high use items. This will assist in budgeting processes and avoid frequent and irregular large funding requests. To continue to improve student success rates, members of the team suggested that the program investigate ways to increase student connections with the STEM center and that faculty expand the use of locally created, campus published course materials to reduce cost barriers.

Summary prepared by Sean J. Abel Dean, Academic Affairs

#### PLAN OF ACTION – POST-VALIDATION (Sixth-Year Evaluation)

#### DEPARTMENT: Life and Physical Sciences PROGRAM: CHEMISTRY

In preparing this document, refer to the Plan of Action developed by the discipline/program during the self-study, and the recommendations of the Validation Team. Note that while the team should strongly consider the recommendations of the validation team, these are recommendations only. However, the team should provide a rationale when choosing to disregard or modify a validation team recommendation.

Identify the actions the discipline/program plans to take during the next six years. Be as specific as possible and indicate target dates. Additionally, indicate by the number each institutional goal and objective which is addressed by each action plan. (See Institutional Goals and Objectives) The completed final plan should be reviewed by the department as a whole.

-	COMMENDATIONS TO IMPROVE <b>STUDENT LEARNING OUTCOMES</b> AND IIEVMENT	Theme/Objective/ Strategy Number AHC from Strategic Plan	TARGET DATE
1)	The chemistry faculty will continue to support the MESA and STEM programs as well as the Learning Resource Center at the Santa Maria and Lompoc Valley Center campuses to help inform students of their support services. The faculty will continue to promote high academic standards and success in achieving Program Learning Outcomes so that our students may thrive once they have moved on to the next stage of their academic careers.	1) A.1/A.7/B.7/C.7. 2) B.7. 3) B.7.	1) ONGOING 2) ONGOING 3) ONGOING
2)	The chemistry faculty will continue to remain current in the latest chemistry literature and useful technology to help convey the material at the appropriate undergraduate level.		
3)	The full-time chemistry faculty will work with the Union and current contracts to help ensure the adjunct-faculty are properly compensated for the additional work associated with inputting Program Learning Outcome data.		

RECOMMENDATIONS TO ACCOMMODATE CHANGES IN <b>STUDENT</b> <b>CHARACTERISTICS</b>	Theme/Objective/ Strategy Number AHC from Strategic Plan	TARGET DATE
Enrollment Changes	A.2/B.4/D.5	ONGOING
The chemistry curriculum sections are growing and expanding. Additional lab space is		
being required at both campuses over the next program cycle. LVC has begun looking into		
converting LVC3-109 into a science lab. As they currently only have one functioning		
chemistry lab, this will help free up LVC3-102 for evening course offerings.		
The SM campus will likely look into M-212 since that lab has fume-hoods. Lab benches		
and gas lines will need to be run to make the room fully functional.		
Demographic Changes	A.3/D.5/D.7	ONGOING
The chemistry program will continue to consider accommodations for student whom cannot		
attend day time classes. We need to expand LVC offerings and evening sections.		
Outfitting lecture and lab rooms with Zoom equipment can help during challenging times.		

RECOMMENDATIONS TO IMPROVE THE EDUCATIONAL ENVIRONMENT	Theme/Objective/ Strategy Number AHC from Strategic Plan	TARGET DATE
<b>Curricular Changes</b> The chemistry faculty are updating curriculum to better serve the needs of the students. Our CHEM140 course will be mapped to the C-ID CHM102 to help ease any transfer issues. Sections of this course may be expanded to the sister campus, LVC, as need grows.	B.8	ONGOING
<b>Co-Curricular Changes</b> A math review CANVAS course may need to be created to help our incoming students meet the Basic Math Skills they require to succeed.	B.8	ONGOING

Neighboring College and University Plans The chemistry faculty will continue to work with neighboring colleges and universities to ensure that courses articulate and topics are aligned.	C.3/C.8/D.6/E.3	ONGOING
<b>Related Community Plans</b> The chemistry faculty will continue to volunteer when asked as we have for science fairs, Friday Night Science, tours of our department, brining hand-on chemistry to other schools, and presenting professional development activities.	A.1/A.5/A.6/E.7/E. 8	ONGOING

RECOMMENDATIONS THAT REQUIRE RESOURCES	ADDITIONAL	Theme/Objective/ Strategy Number AHC from Strategic Plan	TARGET DATE
<ul> <li>Facilities</li> <li>Service the fume-hoods annually as Cal C</li> <li>LVC3-102, 3-114, 2-212, and 2-102 need</li> <li>Need new whiteboards for M205/M213</li> <li>Need new projector screens for M205/M-5)</li> <li>LVC 3-102 requires new ballasts for light</li> <li>Expansion into M212? (\$235,000)</li> </ul>	smart podium upgrades. 213	A.1/A.4/B.1/B.2/ B.3/E.1/E.2	1)ONGOING 2)FALL2023 3)FALL2025 4)SPRING2026 5) SPRING2023
Equipment 1) SM Gloves (\$10,000) 2) SM Equipment under \$500 (\$4,000) + inflation (\$1200) 3) SM Goggles (\$12,000) 4) SM Analytical Balance (\$3,500) 5) SM 16 Centrifuges (\$38,400) 6) ChemDraw Software (\$10,625) 7) SM Repairs (\$500) 8) LVC Water Bath (\$900) 9) LVC Fume-hoods and gas lines for new c 10) SM gas lines and lab benches/stools for n 11) LVC MelTemp equipment for CHEM140 12) Student laptops at SM and LVC campuses 13) SM Large HotPlate (\$1200)	ew chemistry lab (M-212) (\$7200)	A.1/A.4/B.1/B.2/ B.3/D.6/D.7/E.1/ E.2	<ul> <li>6) SPRING2027</li> <li>1) ONGOING</li> <li>2) ONGOING</li> <li>3) FALL2023</li> <li>4) FALL2023</li> <li>5) FALL2023-</li> <li>SPRING2024</li> <li>6) FALL2025</li> <li>7) ONGOING</li> <li>8) FALL2022</li> <li>9)FALL2024</li> <li>10)FALL2024</li> <li>10)FALL2026</li> <li>11)FALL2027</li> <li>12) ONGOING</li> <li>13) FALL2023</li> </ul>

Staffing		A.1/A.4/B.1/B.2	2/ 1) FALL2023 –
and expanding.	b help fill the demand and need of the current sections offe	B.3/E.1/E.2	FALL2025
	y Lab Associate Position (+\$55,000) by Lab Associate Position (+\$55,000)		2)SPRING2023
5) EVE Chemistry and Broke	55 240 1 1500 and 1 051101 (* \$22,000)		2,5111102025
			3)FALL2026

#### VALIDATION TEAM SIGNATURE PAGE

Source Allito

Patrick McGuire Patrick McGuire (Sep 14, 2022 20:20 PDT)

Michael Wagner Michael Wagner (Sep 15, 2022 13:50 PDT)

Sean Abel (Sep 15, 2022 13:55 PDT)

19/2022 9/

# PR Chem Validation Team signature page

#### Final Audit Report

2022-09-15

Created:	2022-09-06
By:	Florentina Perea (fperea@hancockcollege.edu)
Status:	Signed
Transaction ID:	CBJCHBCAABAALXHhxe2en1CEfklxKS0O7MIA8Jfsi-H-

# "PR Chem Validation Team signature page" History

- Document created by Florentina Perea (fperea@hancockcollege.edu) 2022-09-06 - 8:49:54 PM GMT- IP address: 209.129.94.61
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- Signer sean.gottlieb@hancockcollege.edu entered name at signing as Sean Gottlieb 2022-09-09 3:36:44 PM GMT- IP address: 134.16.64.40
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- Signer pmcguire@hancockcollege.edu entered name at signing as Patrick McGuire 2022-09-15 - 3:19:59 AM GMT- IP address: 98.97.60.24
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- Signer mwagner@hancockcollege.edu entered name at signing as Michael Wagner 2022-09-15 8:49:59 PM GMT- IP address: 174.194.199.46
- Document e-signed by Michael Wagner (mwagner@hancockcollege.edu)
   Signature Date: 2022-09-15 8:50:00 PM GMT Time Source: server- IP address: 174.194.199.46
- Document emailed to Sean Abel (sean.abel@hancockcollege.edu) for signature 2022-09-15 8:50:02 PM GMT
- Email viewed by Sean Abel (sean.abel@hancockcollege.edu) 2022-09-15 - 8:55:00 PM GMT- IP address: 209.129.94.61
- Document e-signed by Sean Abel (sean.abel@hancockcollege.edu)
   Signature Date: 2022-09-15 8:55:05 PM GMT Time Source: server- IP address: 209.129.94.61
- Agreement completed. 2022-09-15 - 8:55:05 PM GMT



#### PLAN OF ACTION - Post-Validation

Review and Approval

Plan Prepared By

DUSAN NOURI	
	Date:
	Date:
	Date:
	Date:

Reviewed:

Department Chair\* \_Date: 9/21/2022 lm me

\*Signature of Department Chair indicates approval by department of Plan of Action.

Reviewed:

Dean of Academic Affairs

z Date:\_

Vice President, Academic Affairs

\_\_\_\_\_ Date: 7/20/23